1 Supporting Information:

## Boosted photocatalytic activities of Ag<sub>2</sub>CrO<sub>4</sub> through Eu<sup>3+</sup>-doping process

- 6 Josiane C. Souza<sup>1,2\*</sup>, Samantha C. S. Lemos<sup>2</sup>, Marcelo Assis<sup>2</sup>, Carlos H. M. Fernandes<sup>1</sup>,
- 7 Lara K. Ribeiro<sup>1,2</sup>, Yeison Núñez-de la Rosa<sup>3,4</sup>, Márcio D. Teodoro<sup>5</sup>, Lourdes Gracia<sup>2,6</sup>,
- 8 Juan Andrés<sup>2\*</sup>, Lucia H. Mascaro<sup>1</sup>, Elson Longo<sup>1\*</sup>
- 9 <sup>1</sup>CDMF, Federal University of São Carlos (UFSCar), São Carlos, 13565-905, Brazil.
- 10 <sup>2</sup>Department of Physical and Analytical Chemistry, University Jaume I (UJI), Castelló 12071, Spain.
- <sup>3</sup>Department of Chemistry, Federal University of São Carlos (UFSCar), São Carlos, 13565-905, Brazil.
- 12 <sup>4</sup>Faculty of Engineering and Basic Sciences, Fundación Universitaria Los Libertadores, Bogotá, 111221,
- 13 Colombia.
- <sup>5</sup>Department of Physics, Federal University of São Carlos (UFSCar), São Carlos, 13565-905, Brazil.
- <sup>6</sup>Department of Physical Chemistry, University of Valencia, Valencia, 46010, Spain.
- 16 \*Corresponding author: josi3souza@gmail.com; elson.liec@gmail.com; andres@qfa.uji.es

**Table S1.** Cell parameters and quality indicatives obtained from the Rietveld method of

the as-synthesized samples.

Sample	Lattice	rs (Å)	V7 (Å3)	2	D (0/)	D (0/)	
	а	b	с	<b>V</b> (A <sup>3</sup> )	χ-	<b>K</b> <sub>p</sub> (70)	т <sub>wp</sub> (70)
AC	10.0640	7.0220	5.5382	391.38	1.60	11.43	17.55
ACE25	10.0678	7.0280	5.5401	392.00	1.35	10.20	14.43
ACE50	9.9989	6.9830	5.5026	384.21	1.95	14.0	19.0
ACE100	10.0567	7.0268	5.5322	390.94	1.45	10.8	15.5

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Figure S1. Rietveld refinement plot of the AC (A), ACE25 (B), ACE50 (C), ACE100
(D) samples.

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Number ICSD	Lattic	e Parameter	•s (Å)	V (Å3)	Structure	Spacial	
Number ICSD	а	b	c	<b>V</b> (A <sup>3</sup> )	Structure	group	
<b>252779</b> <sup>1</sup>	10.0667	7.0252	5.5405	391.83	orthorhombic	Pnma	
<b>29580</b> <sup>2</sup>	10.0660	7.0200	5.5360	391.00	orthorhombic	Pnma	

7.0290

5.5400

391.86

orthorhombic

Pnma

Table S2. Reported values of the lattice parameters of Ag<sub>2</sub>CrO<sub>4</sub> (Crystal Structure
Database (ICSD).

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16298<sup>3</sup>

10.0630

We have calculated the Raman spectra of Ag<sub>2</sub>CrO<sub>4</sub> in a single, in which the Eu<sup>3+</sup> 46 47 doping percentage is 12.5 %. The calculated Raman spectra of Ag<sub>2</sub>CrO<sub>4</sub> unit cell are displayed in Figure S2. Two intense Ag modes, associated to the symmetrical stretching 48 of Cr–O bonds of in [CrO<sub>4</sub>] tetrahedra, at 834 and 858 cm<sup>-1</sup> are sensed. The low-intensity 49 bands in the region between 350 and 410 cm<sup>-1</sup> are assigned to symmetrical and 50 asymmetrical bending modes (Ag, B1g, B2g, B3g) of O-Cr-O and Ag-O-Cr bond angles, 51 and the weak B<sub>1g</sub> mode observed at 223 cm<sup>-1</sup> correspond to the stretching of Ag–O bonds 52 in  $[AgO_4]$  tetrahedra. The analysis of Figure S2 reveals that the Eu<sup>3+</sup> doping in the 53 Ag<sub>2</sub>CrO<sub>4</sub> unit cell alters the Raman spectra features. The specific impact on the spectrum 54 profile depends on the site of Eu<sup>3+</sup> occupation. As result of the Eu<sup>3+</sup> substitution, the most 55 intense band (858 cm<sup>-1</sup>) shifts to lower energy. In the spectra of the 4a site Eu-doped 56 Ag<sub>2</sub>CrO<sub>4</sub> two weak Ag modes are observed at 222 and 234 cm<sup>-1</sup> corresponding to the 57 stretching of Ag–O and Eu-O bonds (B<sub>1g</sub> mode in pure Ag<sub>2</sub>CrO<sub>4</sub>). Additional contribution 58 is observed regarding to Eu-O-Cr stretching and Cr-O-Ag bending at 263 cm<sup>-1</sup>. Low-59 intensity bands, assigned to symmetrical and asymmetrical bending modes, Cr-O-Eu 60 bending, in the region between 352 and 439 cm<sup>-1</sup> are found. Finally, important 61 62 contributions in the range of 845 to 895 cm<sup>-1</sup> are observed attributed to the bending and stretching of Ag–O–Cr and of Cr–O bonds, being the more intense at 845, 861, 885, 903 63 and 995 cm<sup>-1</sup> (all with A<sub>g</sub> symmetry). For the Eu-doped Ag<sub>2</sub>CrO<sub>4</sub> in the 4c site, several 64 weak Ag modes from 202 to 281 cm<sup>-1</sup> are observed corresponding to the stretching of Ag-65 O and Eu-O bonds. Additional contribution is observed regarding to Eu-O-Cr stretching 66 and Cr-O-Ag bending at 328 cm<sup>-1</sup> and low-intensity bands, associated to symmetrical 67 68 and asymmetrical bending modes of Cr-O-Eu, are found in the region between 353 and

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456 cm<sup>-1</sup>. Finally, intense bands attributed to Cr–O stretching bond are obtained at 829,
911 and 982 cm<sup>-1</sup>. At this point, a note of caution is mandatory, the theoretical Raman
spectra has calculated for a 12.5% of Eu<sup>3+</sup> doping, which is much higher than the
experimental value of 1%.

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Figure S2. Theoretical Raman spectra of  $Ag_2CrO_4$  and  $Eu^{3+}$ -doped  $Ag_2CrO_4$  (resulting in a  $Eu^{3+}$  doping percentage of 12.5%).

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**Table S3**. FWHM values obtained from the Raman band located at 807 cm<sup>-1</sup> for the as-

78 synthesized samples.

Samples	FWHM
AC	8.7
ACE25	18.5
ACE50	16.8
<b>ACE100</b>	16.0

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Figure S3. (A) UV-vis DRS of the samples, (B-E) Tauc plots  $[F(R\infty) \cdot hv]^{1/2}$  vs hv (proton energy. The purple dashed lines correspond to linear extrapolations to determine indirect Egap values. The thin dark-gray lines correspond to sloped baselines extrapolated from the hv < 1.7 eV range.

In the present work, the prepared  $Ag_2CrO_4$  samples show contributions to its baseline from the apparent absorption tail (see **Figures S3B-S3E**). To determine the band gap, a tangent (thin dark-gray) was first drawn to the baseline at low energies, hv < 1.7eV (and a second line tangent (purple dashed) to the slope in the linear region of the absorption onset was drawn. The intersection of the two lines corresponds to the energy of the band gap. (see references: Chen Z., et al <sup>4</sup> and Makuta P., et al. <sup>5</sup>).

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**Figure S4.** PL emission spectra of the AC, ACE25, ACE50, ACE100 samples.



Figure S5. XRD patterns before and after fifth recycling run. (A) AC sample and (B)
ACE25 sample (in the left). Zoom at XRD patterns within the 2θ range of 37.5°-39° for
AC (A) and ACE25 (B) (in the right).



Figure S6. FE-SEM images after fifth recycling run. (A) AC sample and (B) ACE25sample.

**Table S4**. Relative values,  $\Delta E$ , for 3.12% and 6.25% Eu<sup>3+</sup>-doped Ag<sub>2</sub>CrO<sub>4</sub>, considering as reference the model (1) as the most stable. Lattice parameters and volume of the cell is included for the three models.

			Lattice parameters							
Th	eoretical systems	ΔE (eV)	a (Å)	b (Å)	c (Å)	V (Å <sup>3</sup> )				
	$Eu^{3+}-Ag_2CrO_4(1)$	0.00	10.2028	7.1400	5.0985	371.4154				
3.12%	$Eu^{3+}-Ag_2CrO_4(2)$	0.84	10.1891	7.1570	5.1491	375.4853				
6.25%	$Eu^{3+}-Ag_2CrO_4(1)$	0.00	10.2047	7.1176	5.1173	371.6847				
	$Eu^{3+}-Ag_2CrO_4(2)$	1.89	10.3673	7.2213	5.1419	384.9503				
	$Eu^{3+}-Ag_2CrO_4(3)$	1.07	10.2971	7.1828	5.1250	379.0553				

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**Table S5.** Wyckoff position of Ag and Eu and Ag-O and Eu-O bond length for Ag<sub>2</sub>CrO<sub>4</sub>

and 6.25% Eu<sup>3+</sup>-doped Ag<sub>2</sub>CrO<sub>4</sub> systems. (1) Two Ag<sup>+</sup> atoms substituted by two Eu<sup>3+</sup>

atoms in 4a positions, (2) in 4c positions and (3) in 4a an 4c positions.

						Eu <sup>3+</sup> doped Ag <sub>2</sub> CrO <sub>4</sub>						
	Ag <sub>2</sub> CrO <sub>4</sub>		(1)		(2)			(3)				
Wyckoff	Ag(4a)	Ag (4c)	Eu (4a)	Ag (4a)	Ag (4c)	Eu (4c)	Ag (4a)	Ag (4c)	Eu (4a)	Eu (4c)	Ag (4a)	Ag (4c)
	2.347	2.326	2.309	2.315	2.335	2.311	2.322	2.334	2.290	2.280	2.338	2.318
	2.347	2.326	2.309	2.333	2.365	2.322	2.349	2.357	2.317	2.302	2.377	2.332
Ag-O/	2.371	2.437	2.347	2.373	2.446	2.347	2.389	2.434	2.390	2.378	2.397	2.419
Eu-O	2.371	2.697	2.347	2.393	2.656	2.475	2.433	2.570	2.397	2.445	2.418	2.569
(Å)	2.551		2.365	2.549		2.486	2.504	2.674	2.399	2.492	2.540	2.694
	2.551		2.365	2.559		2.488	2.629		2.403	2.532	2.634	
						2.582				2.596		

Based on the three models showed in **Figure S7**, the values of  $\Delta E$  were calculated and compared. An analysis of the  $\Delta E$  values render that that the substitution in the octahedral coordination site is energetically favored over the tetrahedral one at 3.12% and at 6.25% (by 0.84 eV and 1.89 eV, respectively), being the stability order (1)>(3)>(2) , as can be seen in **Table S4**.





Figure S7. Representation of the three  $Eu^{3+}$  doping  $Ag_2CrO_4$  systems simulated in this study. The gray polyhedra indicate the [AgO<sub>6</sub>] substitution site, the green ones the [AgO<sub>4</sub>] substitution site and the dotted yellow circles indicated the position of  $V_{Ag}$  formation.

An analysis of the results show that the extent of structural distortion depends on the substitution site. In model system (1), the substitution in 4c sites leads to a decrease of Eu-O bond lengths with respect Ag-O bonds, as a result, to a smaller lattice parameters and unit cell volume. Conversely, in the case of model system (2) where substitution occurs in the 4a site a slight increase of the Eu-O bond distances, and the lattice parameters and volume are verified. In case (3) the Eu-O bond lengths suffer both a small increase or decrease.



**Figure S8**. Illustration of the local coordination of the  $Eu^{3+}$  doped site in the Ag<sub>2</sub>CrO<sub>4</sub>

model system (3):  $[EuO_6]$  formation for substitution in 4a site and the  $[EuO_7]$  and  $[AgO_5]$ for substitution in 4c site.



Figure S9. Band structures of (A) Ag<sub>2</sub>CrO<sub>4</sub>, (B) Eu<sup>3+</sup>-Ag<sub>2</sub>CrO<sub>4</sub> substituted in [AgO<sub>6</sub>], (C)
Eu<sup>3+</sup>-Ag<sub>2</sub>CrO<sub>4</sub> substituted in [AgO<sub>4</sub>] and (D) Eu<sup>3+</sup>-Ag<sub>2</sub>CrO<sub>4</sub> substituted in both [AgO<sub>6</sub>]
and [AgO<sub>4</sub>].

## 142 **REFERENCES**

- (1) Fabbro, M. T.; Gracia, L.; Silva, G. S.; Santos, L. P. S.; Andrés, J.; Cordoncillo,
  E.; Longo, E. Understanding the Formation and Growth of Ag Nanoparticles on
  Silver Chromate Induced by Electron Irradiation in Electron Microscope: A
  Combined Experimental and Theoretical Study. *J. Solid State Chem.* 2016, *239*,
  220–227.
- Pinatti, I. M.; Tello, A. C. M.; Trench, A. B.; de Foggi, C. C.; Pereira, P. F. S.;
  Teixeira, M. M.; Jacomaci, N.; Andrés, J.; Longo, E. Zinc-Substituted Ag<sub>2</sub>CrO<sub>4</sub>:
  A Material with Enhanced Photocatalytic and Biological Activity. *J. Alloys Compd.* 2020, *835*, 155315.
- 152 (3) Hackert, M. L.; Jacobson, R. A. The Crystal Structure of Silver Chromate. *J. Solid*153 *State Chem.* 1971, *3*, 364–368.
- (4) Chen, Z.; Miller, E.; Dinh, H. N. *Photoelectrochemical Water Splitting Standards, Experimental Methods, and Protocols*; Springer: New York, NY, 2013, 57.
- (5) Makuła, P.; Pacia, M.; Macyk, W. How To Correctly Determine the Band Gap
  Energy of Modified Semiconductor Photocatalysts Based on UV-Vis Spectra. J. *Phys. Chem. Lett.* 2018, 9, 6814–6817.