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1. MS spectra of the $[W_3S_4H_3(edpp)_3]^+$ and $[W_3S_4D_2H(edpp)_3]^+$ clusters



Figure S1. MS-Q-TOF experimental (bottom) and calculated (top) spectra of the $[W_3S_4H_3(edpp)_3]^+$ (left) and $[W_3S_4D_2H(edpp)_3]^+$ (right) cataionic clusters in CH₃CN at 20V.

2. ³¹P{¹H} NMR spectra of $[W_3S_4H_3(edpp)_3]^+$ and $[W_3S_4D_2H(edpp)_3]^+$ species



Figure S2. ³¹P{¹H}-NMR spectra of $[W_3S_4H_3(edpp)_3]BPh_4$ (top) and $[W_3S_4D_2H(edpp)_3]BPh_4$ (bottom) in CD₃CN.

3. FT-IR spectra of $[W_3S_4H_3(edpp)_3]^+$ and $[W_3S_4D_2H(edpp)_3]^+$ compounds



Figure S3. FT-IR spectra of $[W_3S_4H_3(edpp)_3]^+$ (red) and $[W_3S_4D_2H(edpp)_3]^+$ (blue).

4. Spatial positions of the hydridic and protonic H atoms in the trimetallic cluster



Figure S4. Front view of the $[W_3S_4H_3(edpp)_3]^+$ cationic cluster. The hydride atoms and NH_2 groups are highlighted; the other H atoms have been omitted for clarity.

5. Calculated UV-vis spectra for further species formed upon addition of HCI/HBr



Figure S5. Spectra calculated for the different intermediates formed in the reaction of $[W_3S_4H_3(edpp)_3]^+$ with HBr in acetonitrile solution at 25.0°C. The spectra were obtained from the fit of the spectral changes.