

Supplementary Material

Multifunctional approach to improve water oxidation performance with MOF-based photoelectrodes

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Table S1. Results of energy-dispersive X-ray spectroscopy of CoPi/ MNH and CoPi/ UNH.

Table S2. Elemental ratios from XPS.

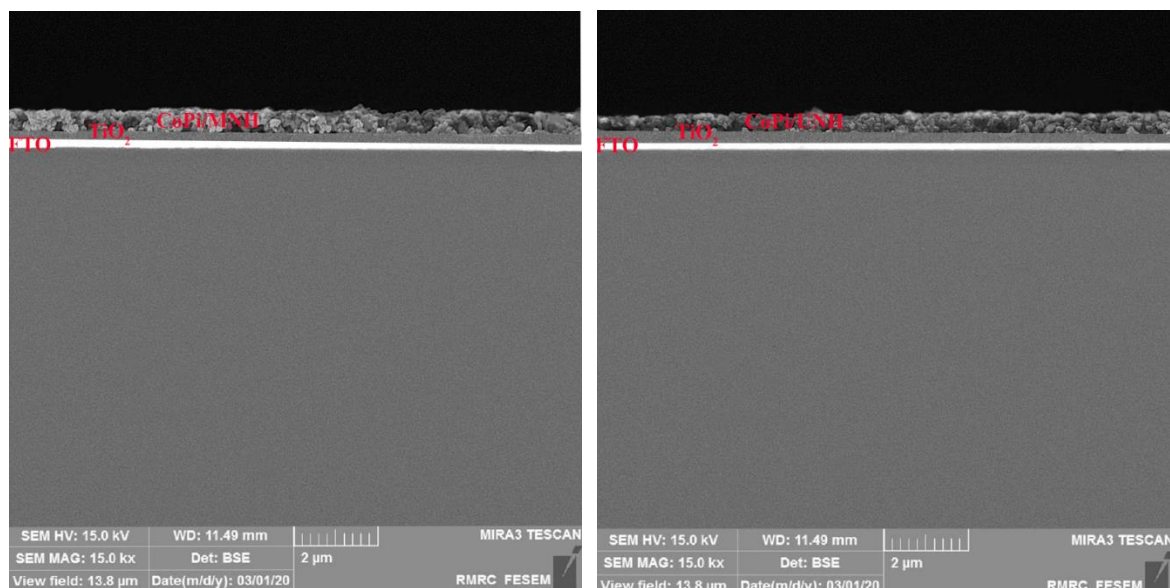


Figure S1. Cross-section view of CoPi/MNH/TiO₂ and CoPi/UNH/TiO₂.

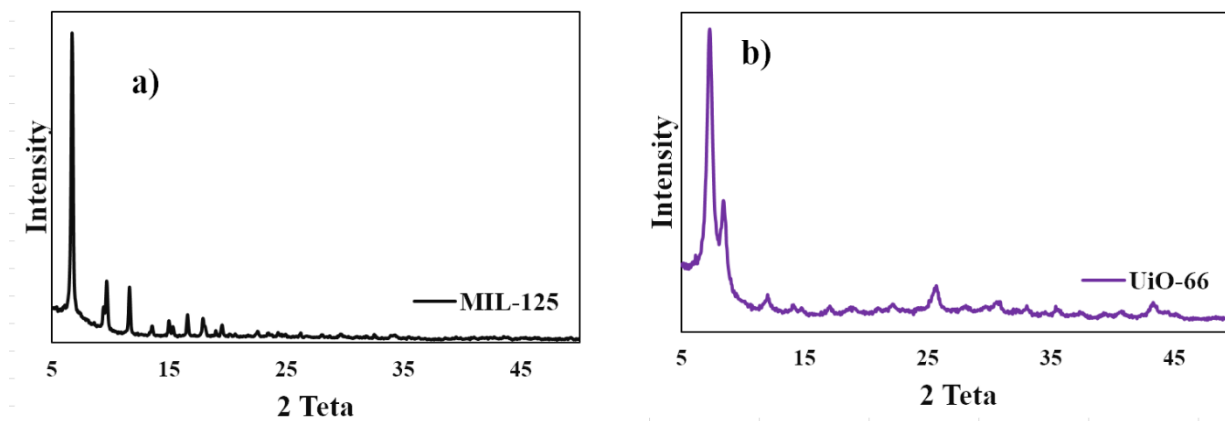


Figure S2. Powder XRD patterns of a) UiO-66 and b) MIL-125.

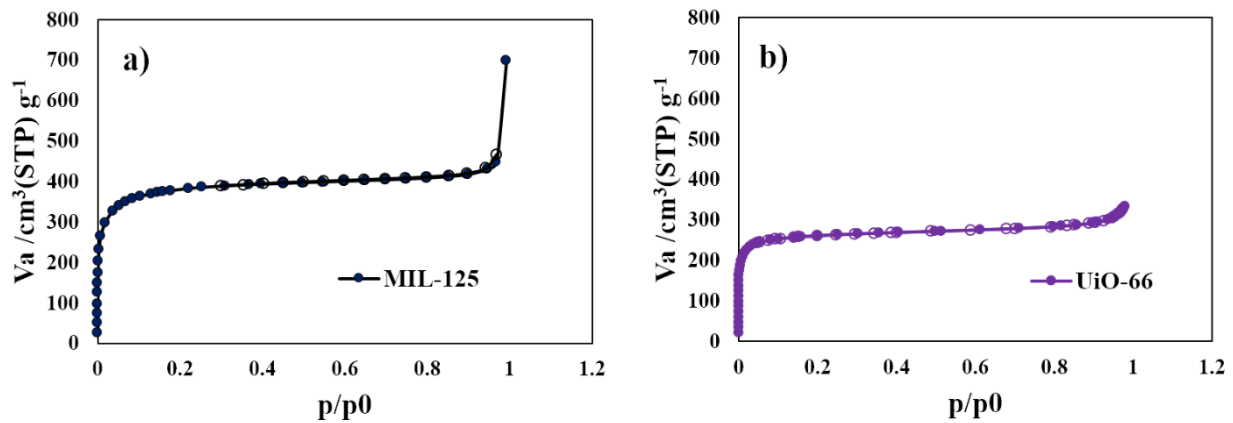


Figure S3. BET surface area analysis: Adsorption-desorption isotherms of the UiO-66 and MIL-125 (a and b).

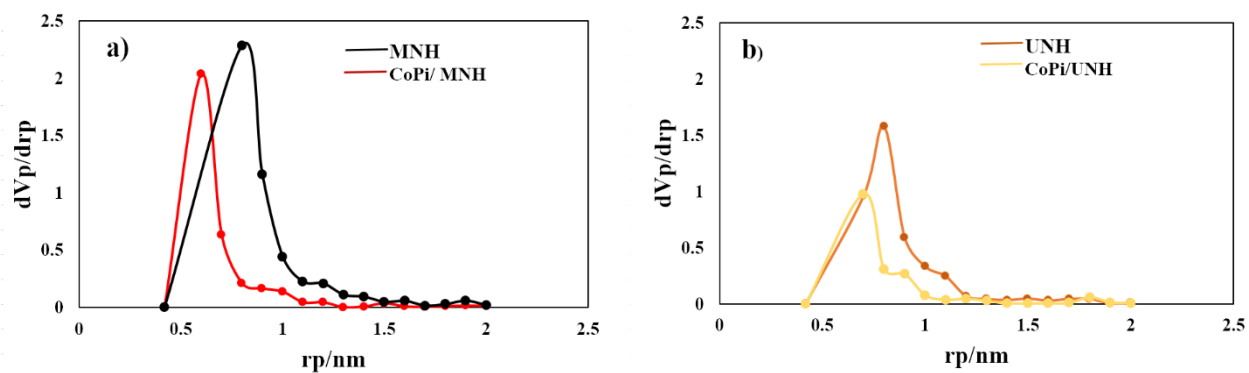


Figure S4. Related pore size distributions for MOF photo catalyst, calculated from MP plots (micropores range).

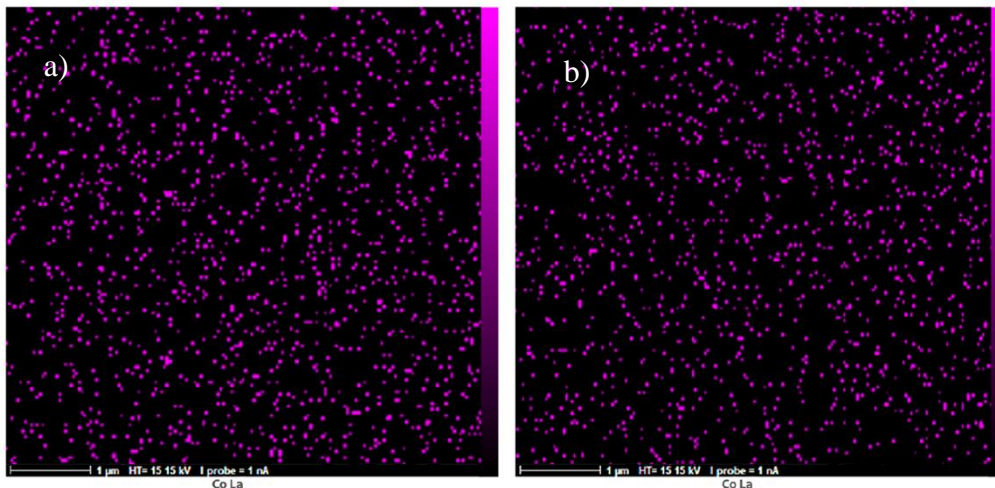


Figure S5. Elemental mapping for CoPi/ MNH (a) and CoPi/ UNH (b).

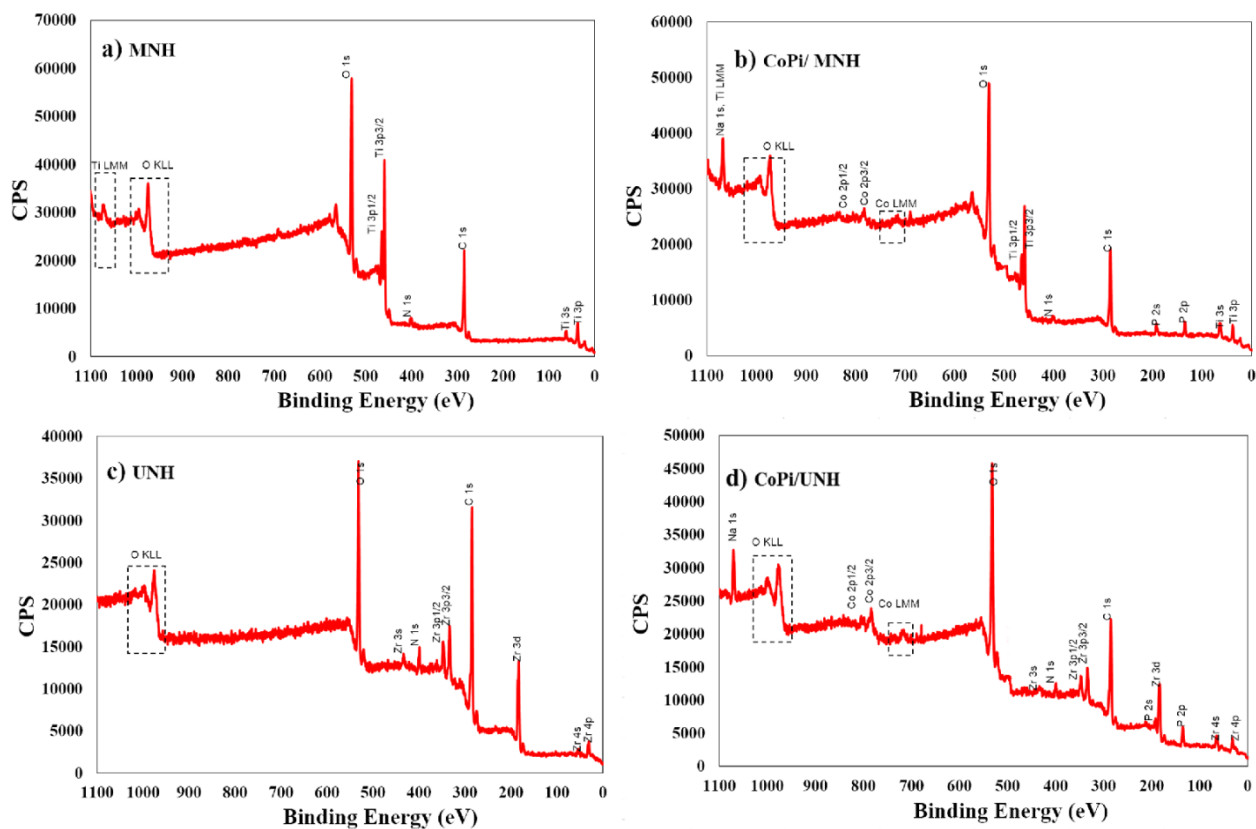


Figure S6. X-ray photoelectron spectroscopy (XPS) analysis, full scan spectrum of (a) CoPi/ MNH (right), and MNH (left), and (b) CoPi/ UNH (right) and UNH(left).

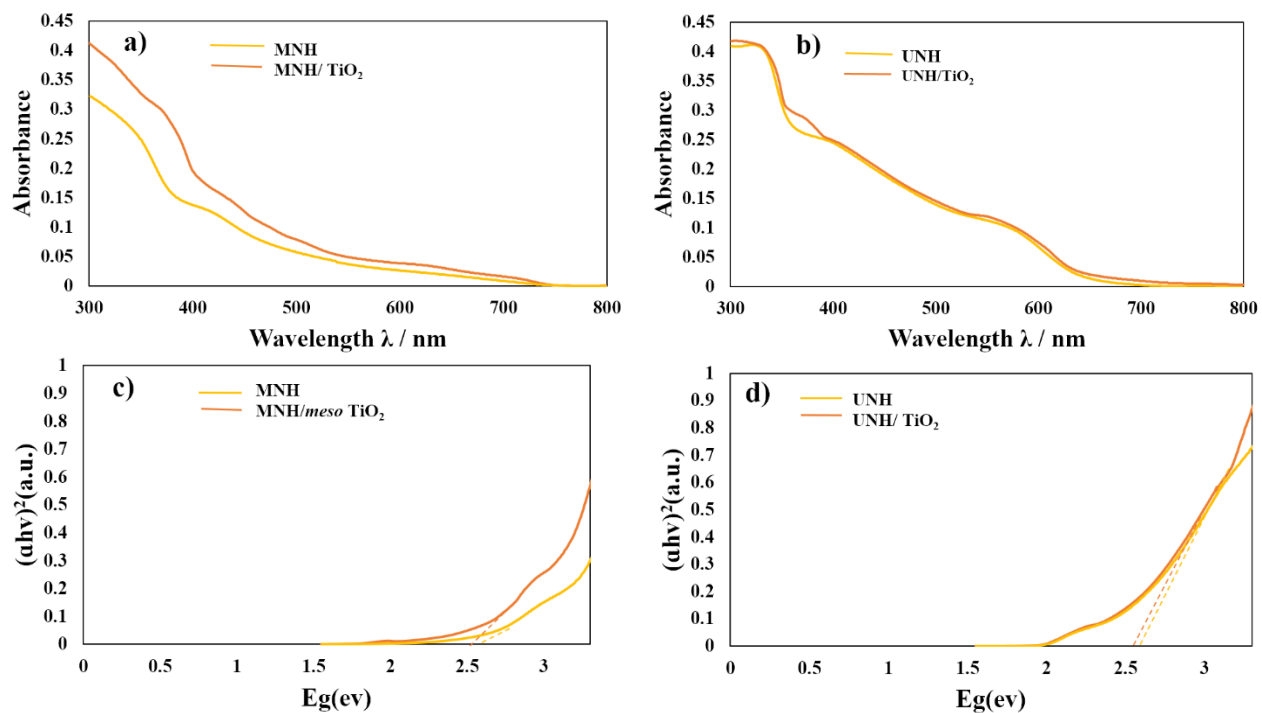


Figure S7. UV-Vis absorption spectra (a and b), and Tauc plots for indirect band gap determination of MOF composites (c and d).

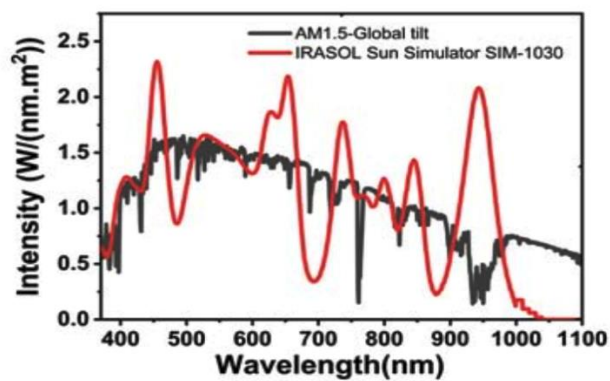


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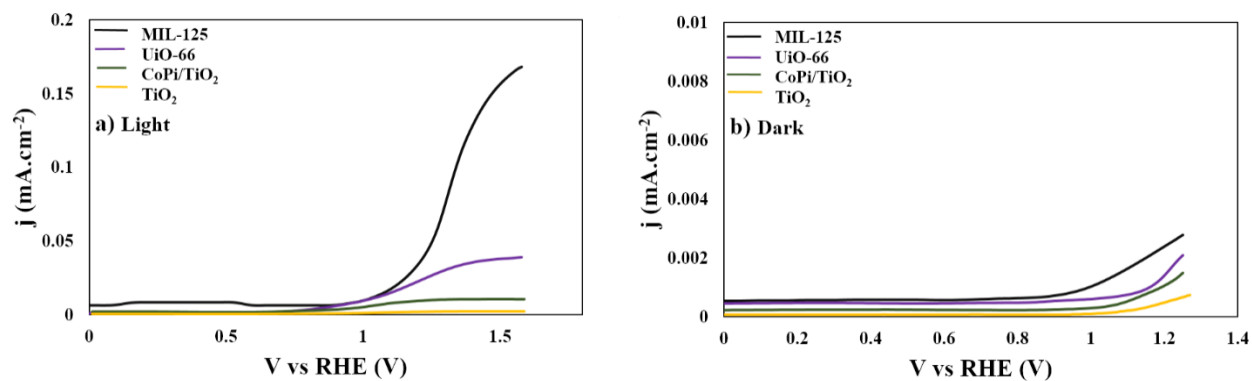


Figure S9. Linear sweep photovoltammograms of MIL-125, UiO-66, CoPi/TiO₂ and TiO₂ in artificial seawater electrolyte a) under illumination at 100 mW·cm⁻² and b) in dark.

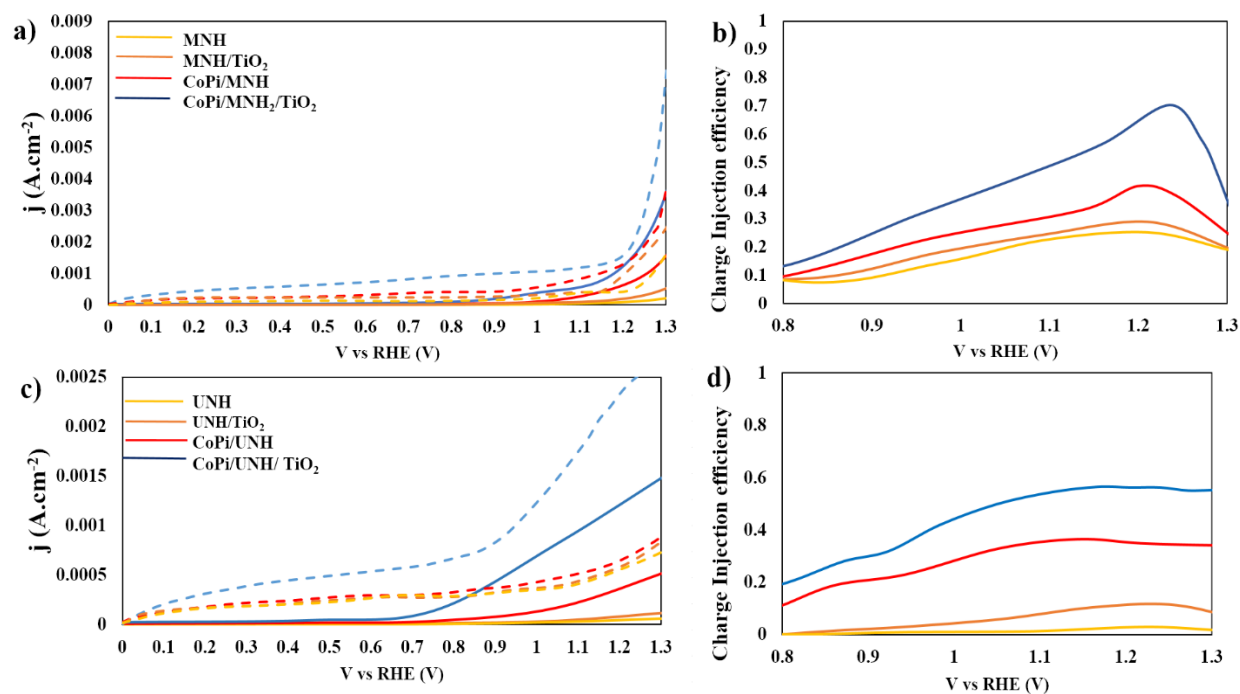


Figure S10: Calculated charge-injection efficiencies of photoelectrodes (j - V curves of the samples in the presence of a hole scavenger (0.5M Na₂SO₃)) without (line) and with (dash line) illumination.

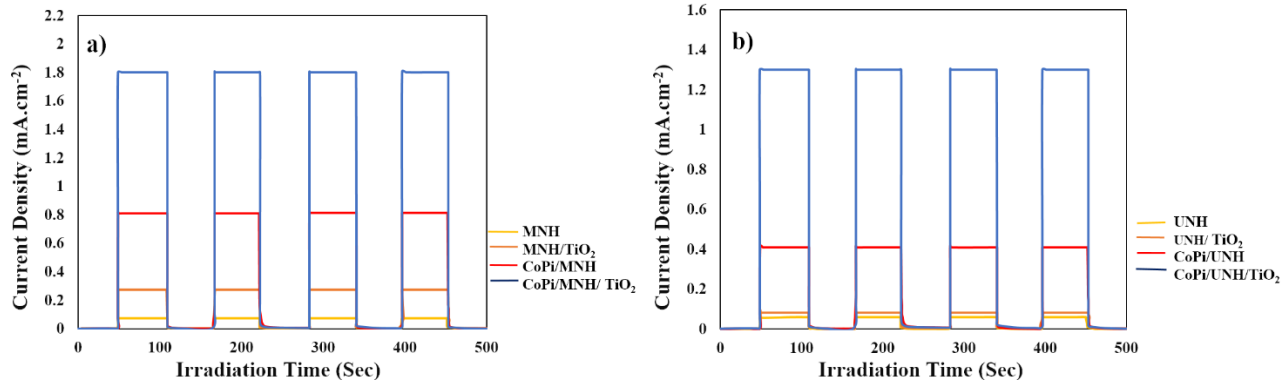


Figure S11. Transient photoelectrochemical measurements of MOFs composite at 1.23V vs RHE.

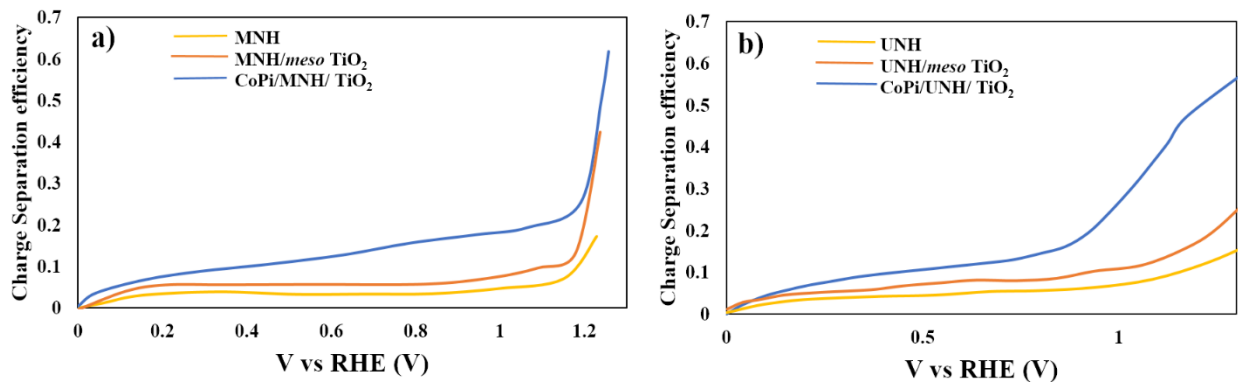


Figure S12. Calculated charge-separation efficiencies of photoelectrodes.

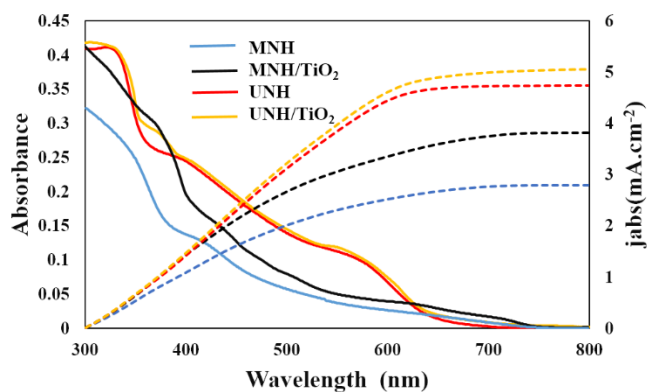


Figure S13. The maximum theoretical photocurrent extracted after integration of the calculated absorptance ($a = 1 - 10^{-A}$).

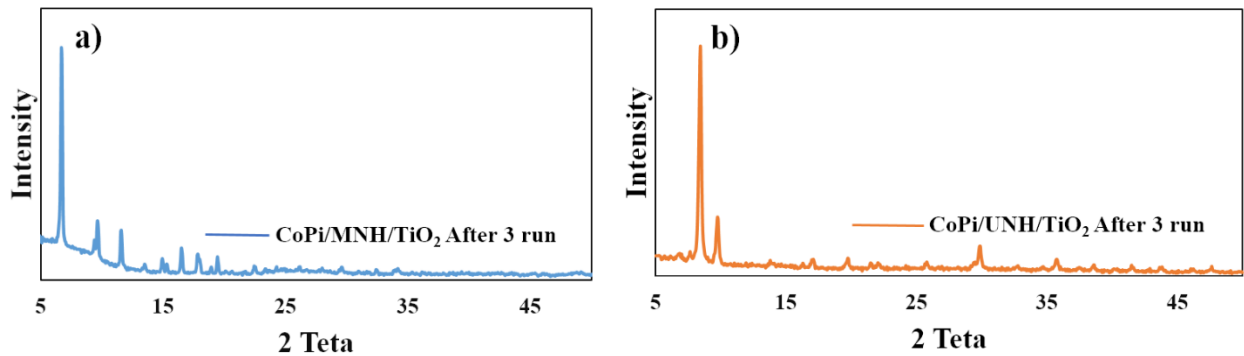


Figure S14. Powder XRD patterns of photoelectrodes after photoelectrochemical testing (3 runs).

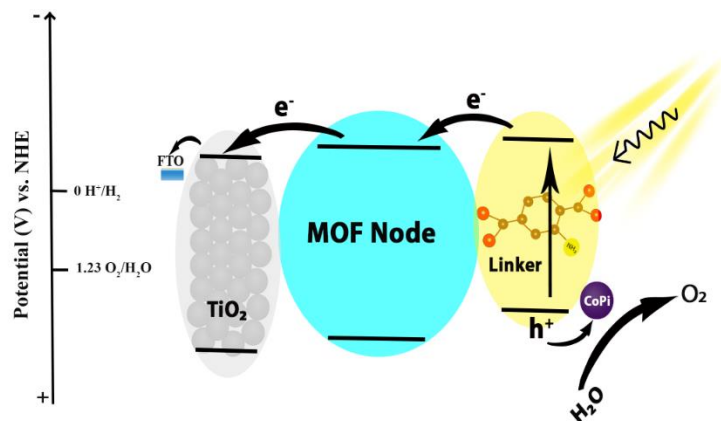


Figure S15. A schematic illustration of photocatalytic water oxidation mechanism over CoPi/MOF/TiO₂ samples.

Table S1. Results of energy-dispersive X-ray spectroscopy of CoPi/ MNH and CoPi/ UNH.

element	C(wt%)	Ti(wt%)	Zr(wt%)	O(wt%)	N(wt%)	Co(wt%)	P(wt%)
CoPi/ MNH	45.5	9.4	-	31.1	2.8	6.1	5.1
CoPi/ UNH	44.8	-	9.1	30.5	2.5	7.1	6

Table S2. Elemental ratios from XPS analysis.

Sample	C	N	O	Zr	Ti	Co	P
MNH	44.428	4.185	38.191	-	13.196	-	-
CoPi/ MNH	34.689	2.059	41.682	-	8.668	1.505	5.126
UNH	64.617	4.56	25.862	4.961	-	-	-
CoPi/ UNH	45.463	2.975	34.846	3.89	-	2.357	5.69