

**Structure, Electronic Properties, Morphology Evolution, and Photocatalytic Activity in PbMoO<sub>4</sub> and Pb<sub>1-2x</sub>Ca<sub>x</sub>Sr<sub>x</sub>MoO<sub>4</sub> (x= 0.1, 0.2, 0.3, 0.4 and 0.5) Solid Solutions**

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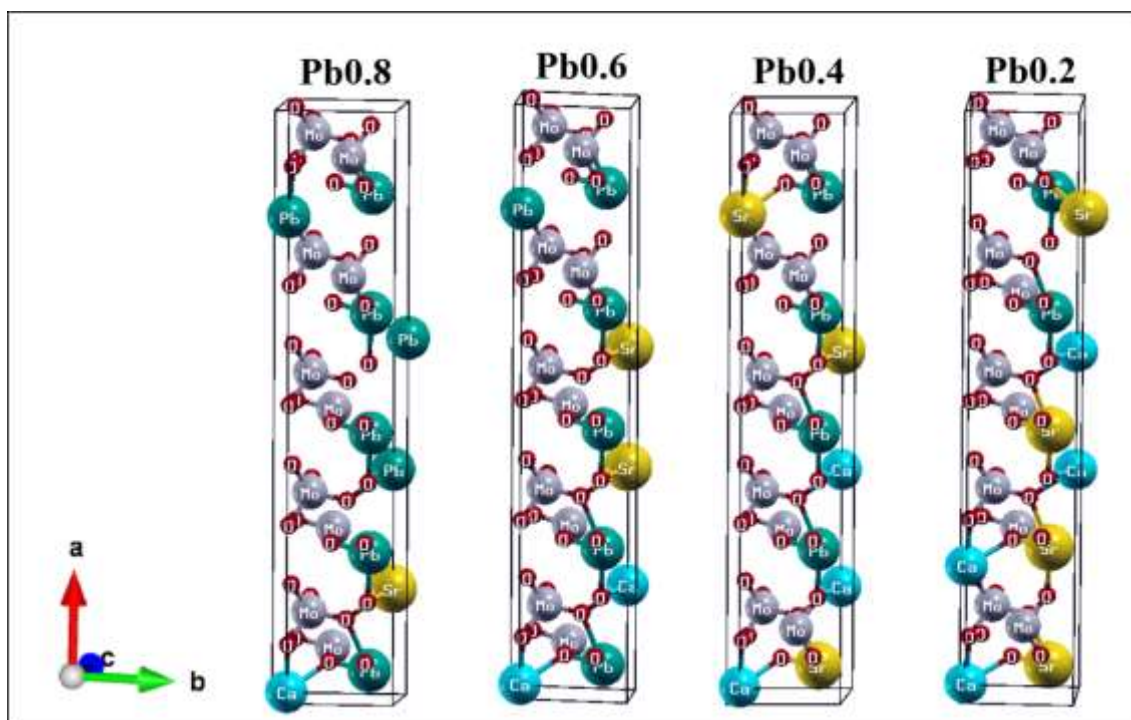
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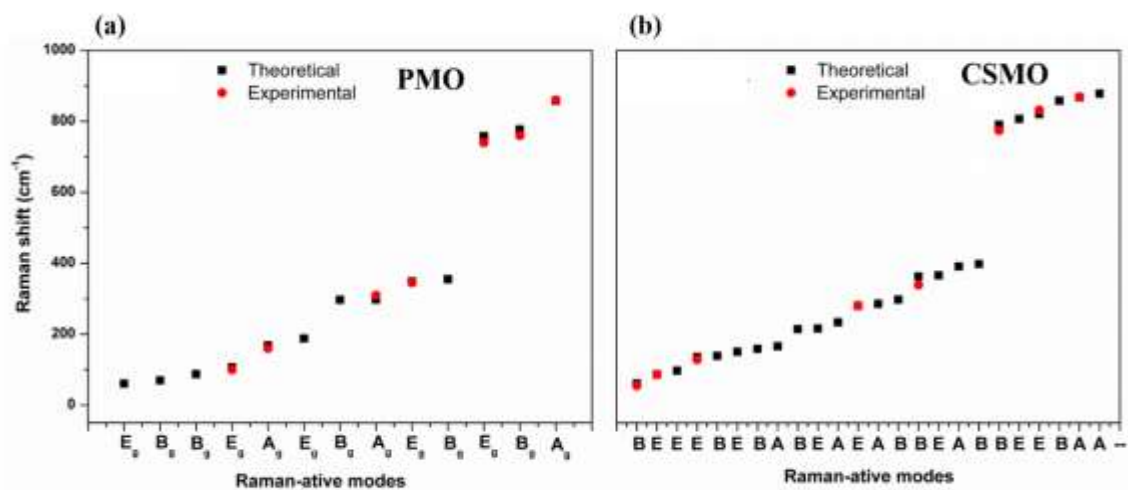
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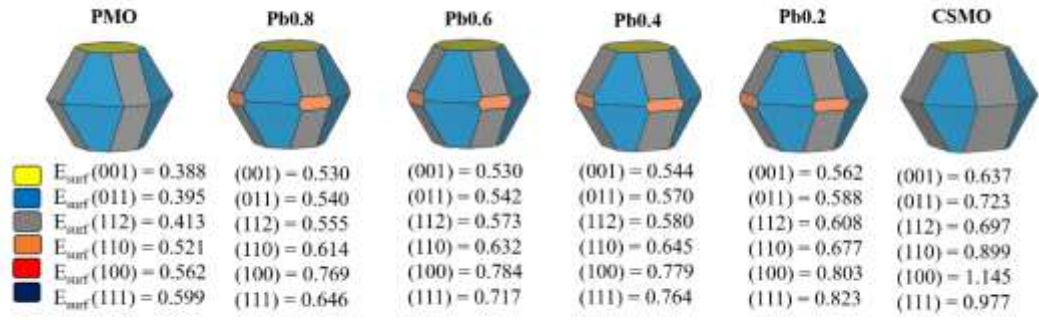
Supporting Information



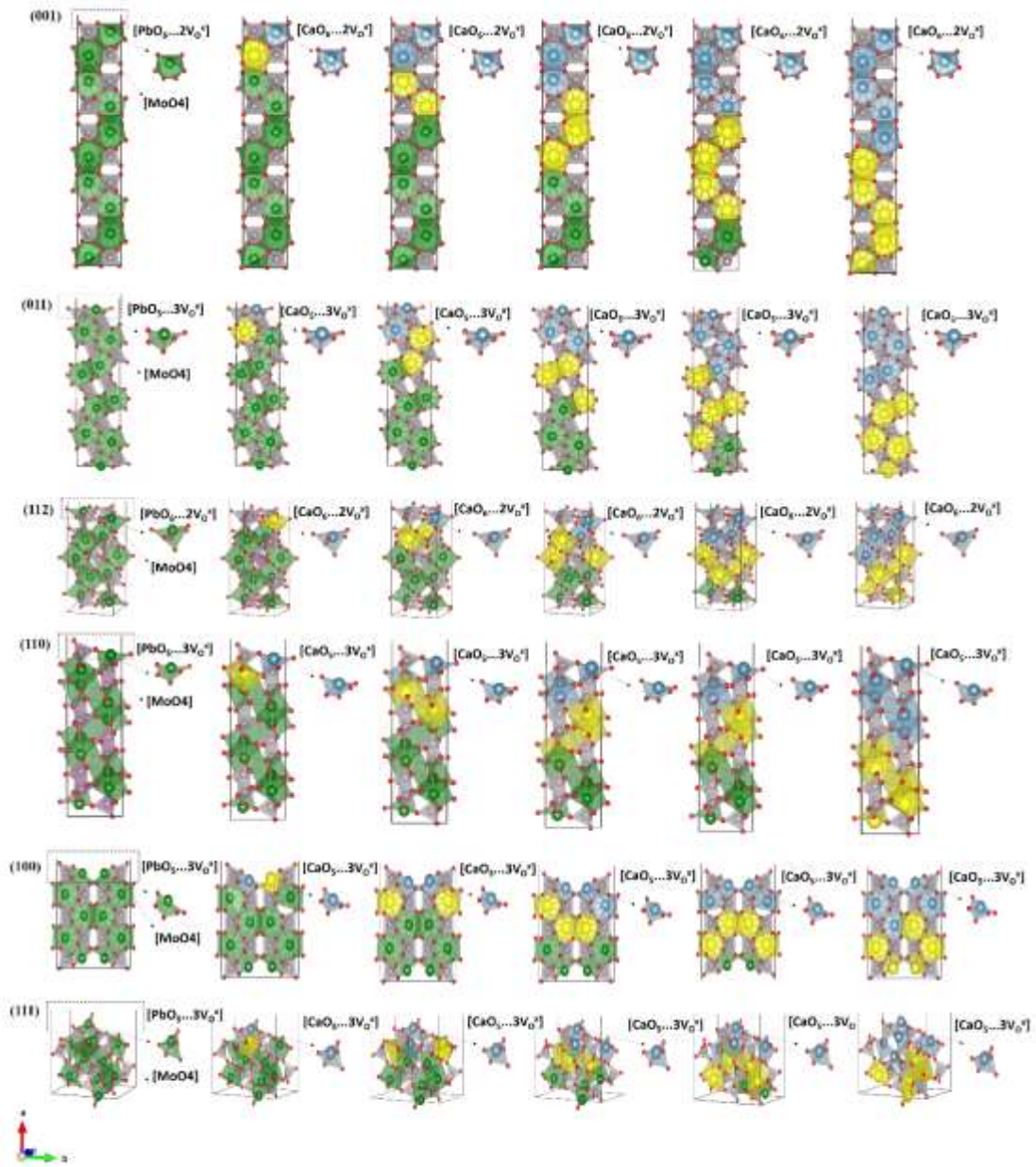
**Figure S1.** 5x1x1 supercell for  $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$  (Pb=0.8, Pb0.6, Pb0.4, Pb0.2) solid solutions.



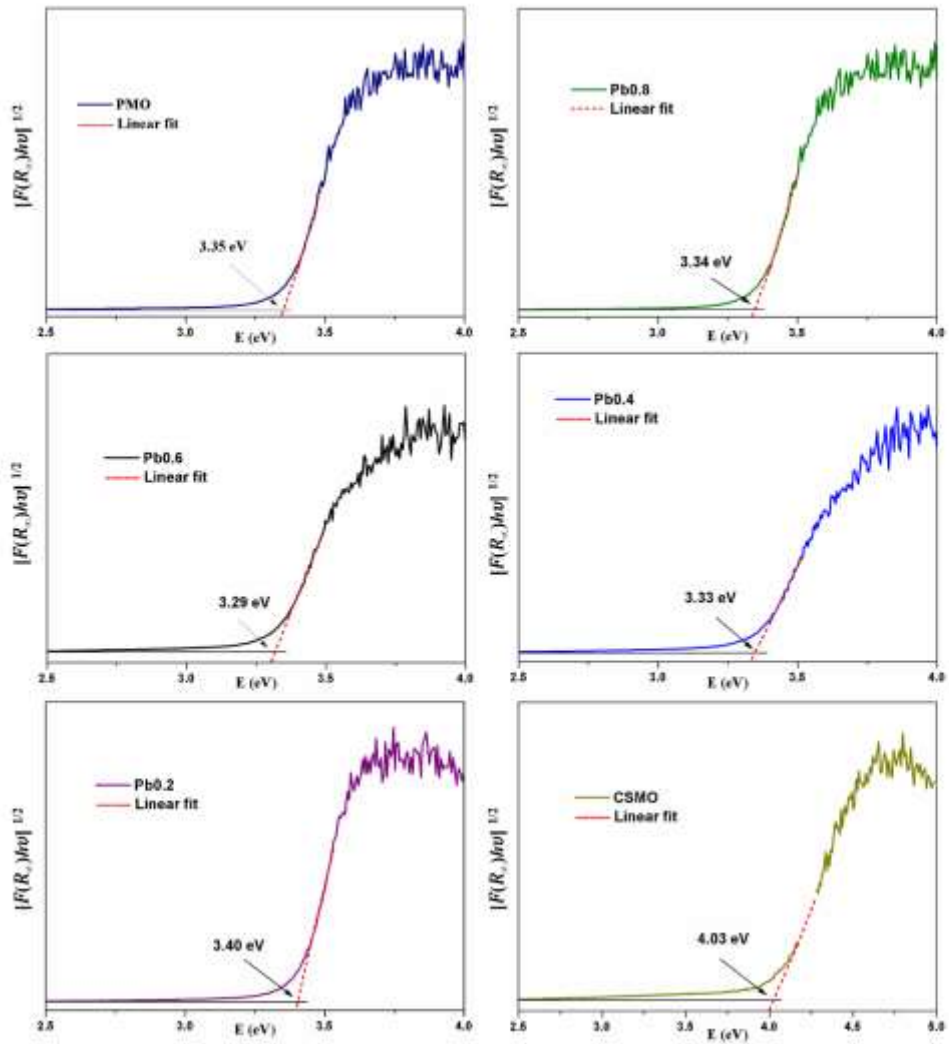
**Figure S2.** Theoretical and experimental values of the Raman modes for: (a) PMO, and (b) CSMO.



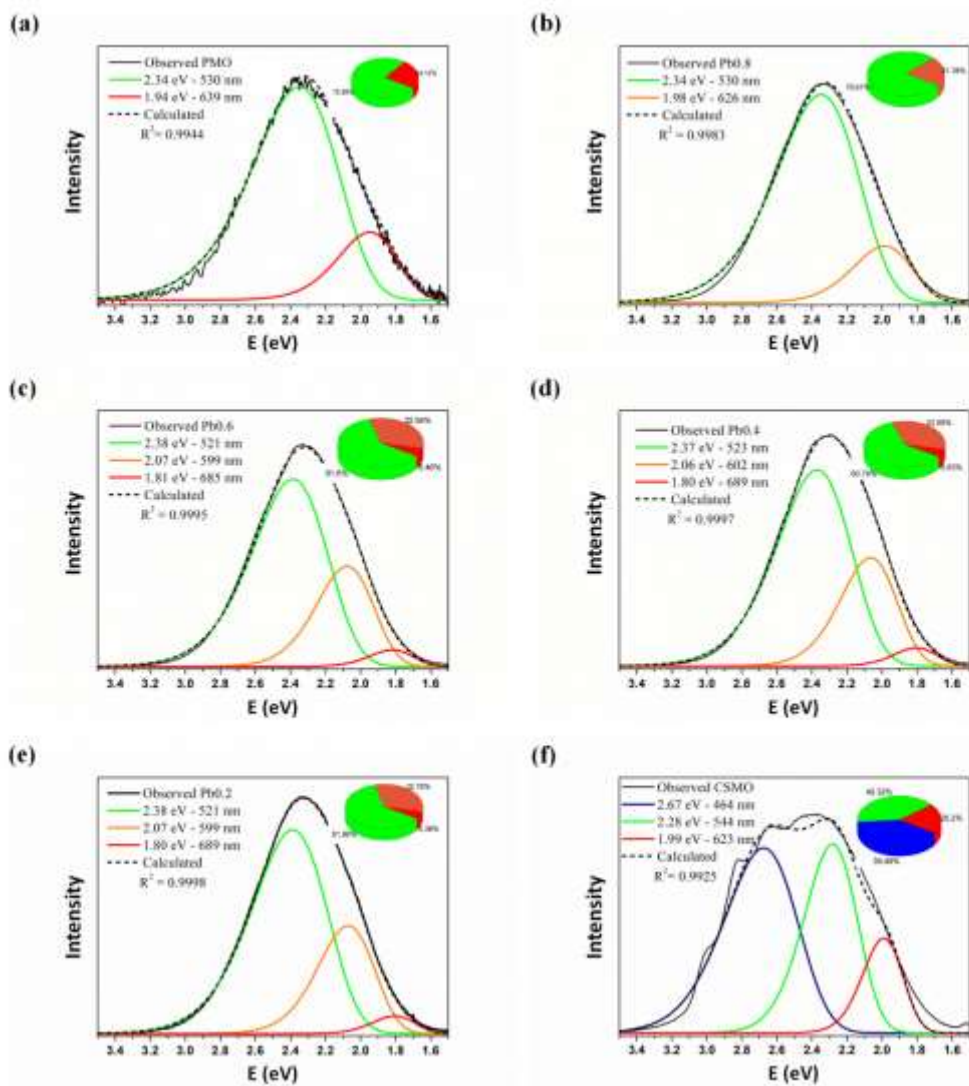
**Figure S3.** Values of  $E_{\text{surf}}$  in ( $\text{J}/\text{m}^2$ ) and possible morphologies of PMO and  $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$  solid solutions. PMO ( $x=0$ ), Pb0.8 ( $x=0.1$ ), Pb0.6 ( $x=0.2$ ), Pb0.4 ( $x=0.3$ ), Pb0.2 ( $x=0.4$ ), CSMO ( $x=0.5$ ).



**Figure S4.** Schematic representations of surfaces: a) (001), b) (112), c) (100), d) (110), e) (111), and f) (011) for PMO and  $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$  ( $\text{Pb}=0.8, \text{Pb}0.6, \text{Pb}0.4, \text{Pb}0.2$  and CSMO) solid solutions.

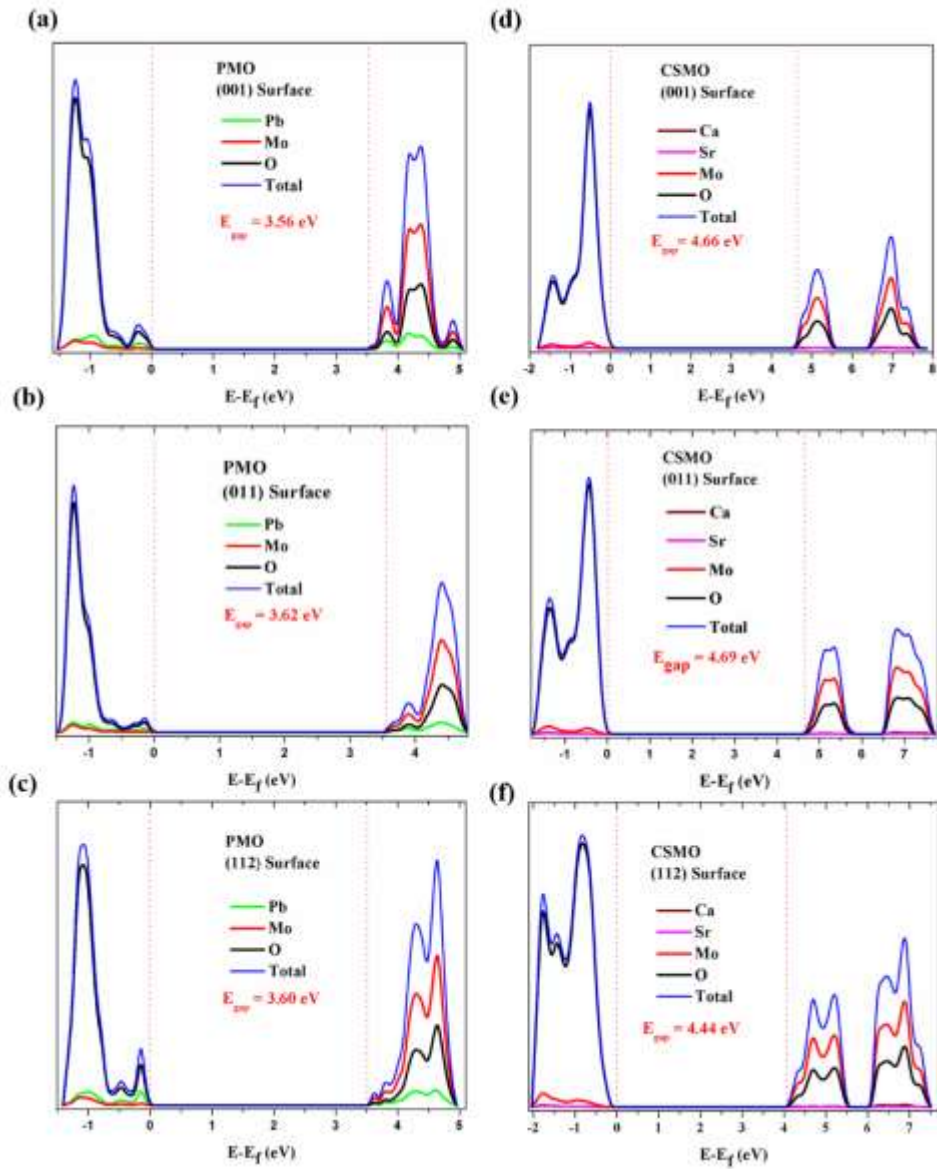


**Figure S5.** Calculated  $E_{\text{gap}}$  values by using the Wood and Tauc for PMO and  $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$  (Pb=0.2, Pb0.4, Pb0.6, Pb0.8 and CSMO) solid solutions.

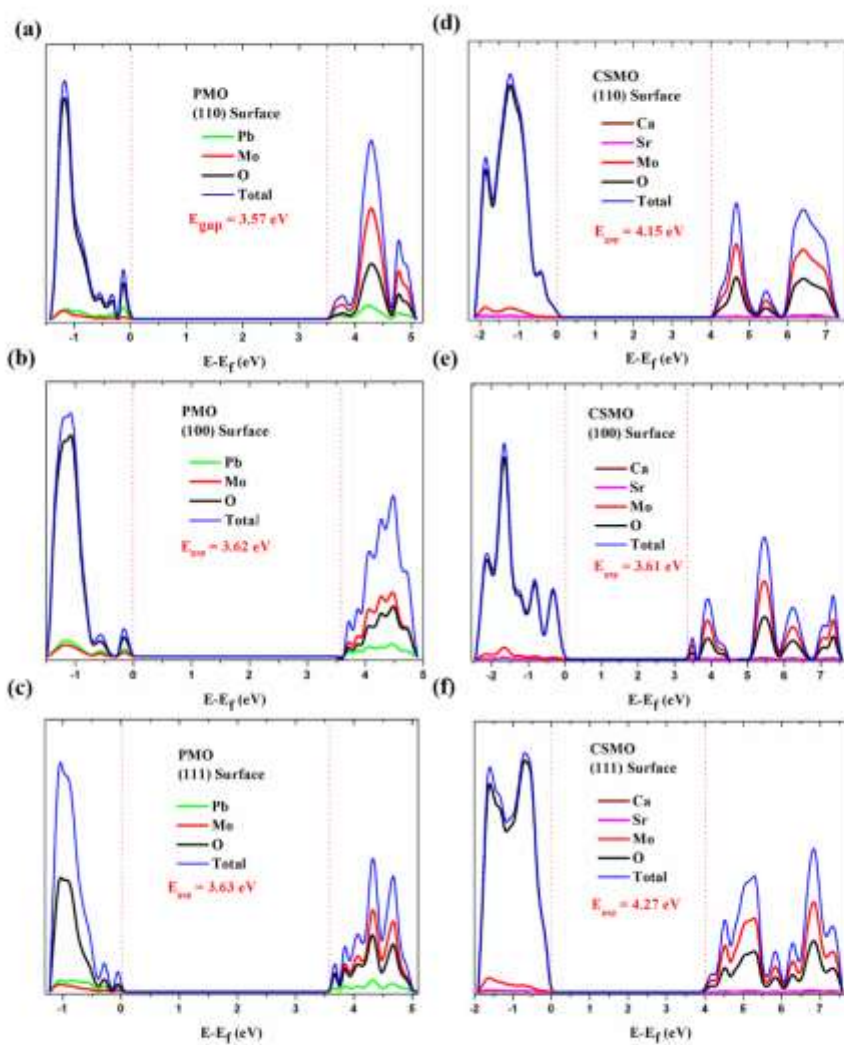


**Figure S6.** PL deconvolution curves: (a) PMO, (b) Pb0.8, (c) Pb0.6, (d) Pb0.4, (e) Pb0.2 and (f) CSMO.





**Figure S7.** DOS of PMO surfaces (a) (001), (b) (011), (c) (112), and DOS of CSMO surfaces (d) (001), (e) (011), (f) (112).



**Figure S8.** DOS of PMO surfaces (a) (110), (b) (100), (c) (111) and DOS of CSMO surfaces (d) (110), (e) (100), (f) (111).



**Table S1.** Theoretical values of the Pb-O, Ca-O, and Mo-O distances at the [PbO<sub>6</sub>], [PbO<sub>5</sub>], [CaO<sub>6</sub>], [CaO<sub>5</sub>] and [MoO<sub>4</sub>] clusters at the exposed surfaces of the PMO and Pb<sub>1-2x</sub>Ca<sub>x</sub>Sr<sub>x</sub>MoO<sub>4</sub> (Pb= 0.2, Pb0.4, Pb0.6, Pb0.8 and CSMO) solid solutions in (Å) and the broken bond density (D<sub>b</sub>) in (nm<sup>-2</sup>).

Surface Bonds (Å)												
(001) Surface												
Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]
2	2.494	2.383	2.376	2.399	2.372	2.398	1.777	1.786	1.782	1.787	1.784	1.789
2	2.533	2.449	2.445	2.454	2.446	2.413	1.860	1.850	1.861	1.848	1.861	1.836
2	2.692	2.652	2.445	2.654	2.671	2.560						
Surface area (nm <sup>2</sup> )	0.305	0.305	0.305	0.305	0.305	0.290						
Broken bond number (N <sub>b</sub> )	2	2	2	2	2	2						
D <sub>b</sub> (nm <sup>-2</sup> )	6.557	6.557	6.557	6.557	6.557	6.896						
E <sub>relax</sub>	0.134	0.046	0.103	0.193	0.271	0.095						
(011) Surface												
Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]
1	2.338	2.277	2.268	2.267	2.269	2.302	1.779	1.798	1.844	1.795	1.791	1.795
2	2.403	2.347	2.345	2.348	2.348	2.320	1.789	1.803	1.822	1.800	1.798	1.796
3	2.498	2.389	2.382	2.396	2.379	2.336	1.823	1.813	1.798	1.816	1.821	1.821
4	2.568	2.419	2.432	2.419	2.428	2.400	1.853	1.836	1.791	1.841	1.845	1.831
5	2.588	2.471	2.529	2.492	2.527	2.433						
Surface area (nm <sup>2</sup> )	0.371	0.371	0.371	0.371	0.371	0.380						
Broken bond number (N <sub>b</sub> )	3	3	3	3	3	3						
D <sub>b</sub> (nm <sup>-2</sup> )	8.086	8.086	8.086	8.086	8.086	7.894						
E <sub>relax</sub>	0.294	0.052	0.558	0.600	0.661	0.397						
(112) Surface												
Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[CaO <sub>6</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]
1	2.450	2.344	2.342	2.343	2.327	2.329	1.775	1.780	1.794	1.793	1.794	1.779
2	2.456	2.380	2.422	2.420	2.444	2.358	1.791	1.791	1.795	1.795	1.795	1.791
3	2.519	2.431	2.483	2.460	2.472	2.411	1.837	1.833	1.830	1.827	1.836	1.830
4	2.552	2.502	2.513	2.533	2.481	2.486	1.842	1.850	1.847	1.849	1.845	1.838
5	2.703	2.525	2.547	2.590	2.499	2.500						
6	2.736	2.711	2.691	2.670	2.766	2.530						
Surface area (nm <sup>2</sup> )	0.567	0.567	0.567	0.567	0.567	0.538						
Broken bond number (N <sub>b</sub> )	4	4	4	4	4	4						
D <sub>b</sub> (nm <sup>-2</sup> )	7.054	7.054	7.054	7.054	7.054	7.434						
E <sub>relax</sub>	1.592	0.084	0.084	0.122	0.158	1.509						
(110) Surface												
Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO

Bond (Å)	[PbO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]
1	2.343	2.309	2.323	2.316	2.320	2.470	1.735	1.739	1.817	1.739	1.738	1.747
2	2.389	2.404	2.428	2.476	2.412	2.517	1.801	1.813	1.888	1.815	1.816	1.805
3	2.455	2.409	2.436	2.828	2.436	2.544	1.842	1.857	1.738	1.845	1.847	1.847
4	2.554	2.443	2.433	2.829	2.436	2.554	1.889	1.872	1.845	1.890	1.889	1.872
5	2.856	2.451	2.484	2.841	2.480	2.608						
<b>Surface area (nm<sup>2</sup>)</b>	0.478	0.478	0.478	0.478	0.478	0.498						
<b>Broken bond number (N<sub>b</sub>)</b>	4	4	4	4	4	4						
<b>D<sub>b</sub> (nm<sup>-2</sup>)</b>	8.368	8.368	8.368	8.368	8.368	8.032						
<b>E<sub>relax</sub></b>	1.861	0.103	0.121	0.157	0.199	0.330						

**(100) Surface**

Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]
1	2.303	2.258	2.263	2.255	2.282	2.213	1.730	1.731	1.731	1.737	1.730	1.739
2	2.329	2.285	2.290	2.287	2.308	2.268	1.806	1.796	1.812	1.807	1.809	1.812
3	2.256	2.309	2.320	2.300	2.308	2.298	1.846	1.850	1.837	1.842	1.847	1.835
4	2.800	2.396	2.471	2.434	2.471	2.395	1.878	1.877	1.875	1.858	1.875	1.848
5	2.839	2.423	2.475	2.458	2.508	2.453						
<b>Surface area (nm<sup>2</sup>)</b>	0.676	0.676	0.676	0.676	0.676	0.640						
<b>Broken bond number (N<sub>b</sub>)</b>	6	6	6	6	6	6						
<b>D<sub>b</sub> (nm<sup>-2</sup>)</b>	8.875	8.875	8.875	8.875	8.875	9.375						
<b>E<sub>relax</sub></b>	0.573	0.028	0.166	0.183	0.246	0.746						

**(111) Surface**

Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[CaO <sub>5</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]	[MoO <sub>4</sub> ]
1	2.351	2.283	2.365	2.283	2.322	2.492	1.763	1.790	1.732	1.799	1.795	1.785
2	2.364	2.292	2.368	2.307	2.377	2.496	1.808	1.806	1.856	1.801	1.806	1.793
3	2.422	2.445	2.435	2.431	2.457	2.521	1.820	1.816	1.861	1.808	1.815	1.817
4	2.528	2.457	2.450	2.446	2.465	2.538	1.836	1.822	1.831	1.832	1.817	1.825
5	2.854	2.497	2.481	2.471	2.531	2.667						
<b>Surface area (nm<sup>2</sup>)</b>	1.004	1.004	1.004	1.004	1.004	0.994						
<b>Broken bond number (N<sub>b</sub>)</b>	7	7	7	7	7	7						
<b>D<sub>b</sub> (nm<sup>-2</sup>)</b>	6.972	6.972	6.972	6.972	6.972	7.042						
<b>E<sub>relax</sub></b>	1.220	0.073	0.115	0.166	0.220	0.301						

**Table S2.** Theoretical and experimental values of the  $E_{\text{gap}}$  for PMO and  $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$  (Pb=0.8, Pb0.6, Pb0.4, Pb0.2 and CSMO) solid solutions.

Samples	$E_{\text{gap}}$ of the Bulk (eV)					
	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Experimental	3.35	3.34	3.29	3.33	3.40	4.03
Theoretical	3.61	3.65	3.75	3.89	3.92	4.74

**Table S3.** Calculated values of the  $E_{\text{gap}}$  for PMO and  $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$  surfaces.

Surfaces	$E_{\text{gap}}$ of the surfaces (eV)					
	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
(001)	3.56	3.56	3.57	3.56	3.64	4.66
(100)	3.62	3.59	3.61	3.81	3.99	3.61
(011)	3.62	3.65	3.7	3.8	4.06	4.69
(110)	3.57	3.61	3.67	3.66	4.0	4.15
(111)	3.63	3.69	3.68	3.79	3.96	4.27
(112)	3.60	3.64	3.69	3.77	3.97	4.44

**Table S4.** Calculated values (at atoms and clusters) of atomic Mulliken charge for PMO and CSMO bulk, and surfaces.

Bulk				Surfaces			
PMO		CSMO		PMO (001)		CSMO (001)	
Species	Population	Species	Population	Species	Population	Species	Population
Pb	1.26	Ca	1.62	[PbO <sub>6</sub> ]	-0.41	[CaO <sub>6</sub> ]	-0.29
Mo	1.67	Sr	1.85	[MoO <sub>4</sub> ]	0.42	[MoO <sub>4</sub> ]	0.28
O5	-0.24	1Mo	1.73	(112)		(112)	
O6	-0.24	2Mo	1.71	[PbO <sub>6</sub> ]	-0.36	[CaO <sub>6</sub> ]	-0.20
O7	-0.24	O5	-0.29	[MoO <sub>4</sub> ]	0.35	[MoO <sub>4</sub> ]	0.21
O8	-0.24	O6	-0.29	(011)		(011)	
O9	-0.24	O7	-0.29	[PbO <sub>5</sub> ]	-0.44	[CaO <sub>5</sub> ]	-0.29
O10	-0.24	O8	-0.29	[MoO <sub>4</sub> ]	0.35	[MoO <sub>4</sub> ]	0.17
O11	-0.24	O9	-0.29	(110)		(110)	
O12	-0.24	O10	-0.29	[PbO <sub>5</sub> ]	-0.18	[CaO <sub>5</sub> ]	-0.41
[PbO <sub>8</sub> ]	-0.69	O11	-0.29	[MoO <sub>4</sub> ]	0.16	[MoO <sub>4</sub> ]	0.12
[MoO <sub>4</sub> ]	0.69	O12	-0.29	(100)		(100)	
		[CaO <sub>8</sub> ]	-0.68	[PbO <sub>5</sub> ]	-0.20	[CaO <sub>5</sub> ]	-0.16
		[SrO <sub>8</sub> ]	-0.45	[MoO <sub>4</sub> ]	0.29	[MoO <sub>4</sub> ]	0.09
		1[MoO <sub>4</sub> ]	0.58	(111)		(111)	
		2[MoO <sub>4</sub> ]	0.56	[PbO <sub>5</sub> ]	-0.35	[CaO <sub>5</sub> ]	-0.49
				[MoO <sub>4</sub> ]	0.18	[MoO <sub>4</sub> ]	0.20

**Table S5.** The comparison of photocatalytic activity of as prepared nanoparticles (Pb0.6, Pb0.8 and PMO) with TiO<sub>2</sub> (Degussa P-25).

<b>Structure</b>	<b>Dye concentration</b>	<b>Time (min)</b>	<b>kinetic constan K min<sup>-1</sup></b>	<b>Degraded dye %</b>	<b>Light used</b>	<b>References</b>
Nanoparticles	10mg/L	60	-	78	800W Xenon lamp	[81]
Nanoparticles	2x10 <sup>-6</sup> mol/L	90	0.014	38	25 W, 555 nm, Philips	[82]
Nanoparticles	10 mg/L	150	-	92	150 W Xenon lamp	[83]
Nanoparticles	1x10 <sup>-5</sup> mol/L	60	0.020	88	15 W TUV Philip	This work
Nanoparticles	1x10 <sup>-5</sup> mol/L	60	0.017	70	15 W TUV Philip	This work
Nanoparticles	1x10 <sup>-5</sup> mol/L	60	0.016	87	15 W TUV Philip	This work