$WO_3/BiVO_4$: Impact of charge separation at the timescale of water oxidation – Supplementary Information

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Fig. S1 Typical XRD patterns of BiVO₄, WO₃ and WO₃/BiVO₄ shown alongside reference standards (monoclinic scheelite BiVO₄, monoclinic WO₃ and tetragonal cassiterite F: SnO₂).



Fig. S2 SEM images of (a) $BiVO_4$, (b) WO_3 and (c) $WO_3/BiVO_4$. Scale bar represents 200 nm.





Fig. S3 UV-Vis spectra of photoanode films measured using a Shimadzu UV-2700 equipped with an integrated sphere. (a) Absorptance of films taking into account of transmittance, diffuse reflectance and specular reflectance. (b) UV-Vis spectra of bare BiVO4 and WO₃/BiVO₄ with varying thicknesses of BiVO₄. (c) images of the films: (i) BiVO₄, (ii) WO₃ and (iii) WO₃/BiVO₄



Fig. S4 Spectral output of the Xe lamp at 100 mW.cm⁻² with a KG3 filter (simulated 1 sun irradiance).



Fig. S5 JV curves of $WO_3/BiVO_4$ (solid line) with varying thicknesses of $BiVO_4$ (dashed lines) for thicknesses: (a) 75 nm, (b) 125 nm, (c) 175 nm and (d) 350 nm. The measurements were carried out in pH 7, 0.1 M phosphate buffer, under simulated 1 sun illumination (Xe lamp). The photoelectrochemical response of the WO_3 film is included for reference (dotted line).

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IPCE and APCE. Incident photon to current efficiency (IPCE) is determined using the following equation:

$$IPCE = -\frac{J_{ph} \times 1239.8}{P_{\lambda} \times \lambda}$$

(Equation S1)

where J_{ph} (mA cm⁻²) is the photocurrent density, 1239.8 (eV nm) is the Planck's constant multiplied by the speed of light, P_{λ} is the power intensity of the monochromatic light at a given wavelength, λ (nm).

To calculate the absorbed photon to current efficiency (APCE), firstly the photon flux of the simulated 1 sun illumination source (in figure S4) was determined below 500 nm, the absorption onset of the films.

$$Total no. of photons = \int_{350 nm}^{500 nm} Flux(\lambda) d\lambda$$

(Equation S2)

The absorptance of the films were also determined from the UV-vis data in the region: 350-500 nm using the equation:

Absorptance =
$$1 - 10^{-A}$$

(Equation S3)

Where A is the absorbance of the film.

The number of photons absorbed was then given by:

Number of photons absorbed =
$$\int_{350 nm}^{500 nm} (Total no. of photons \times Absorptance) d\lambda$$

(Equation S4)

From the photocurrents obtained from the films at 1.23 V_{RHE} , the APCE was determined using:

 $\frac{J (A cm^{-2})/1.602 \times 10^{-19}C}{Number of photons absorbed (cm^{-2}s^{-1})}$

Where 1.602×10^{-19} C is the elementary charge.



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Fig. S6 IPCE under monochromatic light and APCE under simulated 1 sun, of films at 1.23 V_{RHE} (a) IPCE of WO₃/BIVO₄ heterojunction (red), illuminated from the front, with a BiVO₄ thickness of 350 nm. The IPCE of bare BiVO₄ (dark grey, dashed) and WO₃ (orange, dashed) are included for reference. (b) The APCE of BiVO₄ (dark grey) and WO₃/BiVO₄ (red) films with varying BiVO₄ thickness illuminated from the front (solid circles) and back (open circles).





Fig. S7 Transient absorption spectra of BiVO₄ and WO₃/BiVO₄, front illumination, in 0.1 M phosphate buffer (pH 7); pump fluence: 500μ J/cm² (355 nm) and 400μ J/cm² (450 nm). Pump fluence was adjusted for normalised photon absorption by the sample. (a) BiVO₄ (grey) and WO₃/ BiVO₄ (red) at 5 µs, under no applied bias, excitation wavelength: 355 nm (b) BiVO₄ (grey) and WO₃/ BiVO₄ (red) at 5 µs, under 1.23 V_{RHE} applied bias, excitation wavelength: 355 nm. (c) Comparison of transient absorption spectra at 10 µs for BiVO₄ and WO₃/BiVO₄ when excited with 355 nm and 450 nm (exciting solely BiVO₄).

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Fig. S8 TA kinetics and TPC of BiVO₄ and WO₃/BiVO₄ under front illumination, excited at 355 nm, with varying applied anodic bias in 0.1 M phosphate buffer; pump fluence: 500 μ/cm². TA kinetic trace at 800 nm of (a) WO₃/BiVO₄, probing electrons in WO₃, and (b) BiVO₄ (absorption tail of broad h⁺ absorption). TPC of (c) BiVO₄ and (d) WO₃/BiVO₄. (e) Normalised TPC at 0.01 s comparing electron extraction times of the heterojunction with that of the bare materials. (f) TA amplitude at 800 nm probing electrons in the WO₃/BiVO₄ system, at 50 µs (before the onset of charge extraction) (blue) compared with the total charge extracted from the $WO_3/BiVO_4$ photoanode (red).



Fig. S9 Comparison of TA kinetics of holes accumulated in BiVO4 (solid) against the integrated transient photocurrent, monitoring charge extraction from the WO₃/BiVO₄ heterojunction photoanode (dash), under applied potential of 1 V_{RHE} (red) and 1.6 V_{RHE} (blue) for (a) water oxidation and (b) sulphite oxidation. Water oxidation exhibits greater bias dependence in hole dynamics and greater hole accumulation, due to the slower kinetics of water oxidation, also reflected in the charge extracted from the film. Whereas for sulphite oxidation, the bias dependence is less pronounced, also shown in the charge extraction, due to sulphite oxidation being more facile in comparison to water oxidation. Less holes accumulate at the surface due to a faster reaction of holes to oxidise sulphite. The films were illuminated from the front (BiVO₄ side), excited at 355 nm, pump fluence: 500 μJ/cm², at 500 nm probing holes in BiVO₄, with and without hole scavenger. Hole scavenger used: 0.1 M Na₂SO₃.

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Fig. S10 Impedance analysis of BiVO₄ and WO₃/BiVO₄ photoanodes in 0.1M phosphate buffer. JV curves from cyclic voltammetry (black and red solid lines) and from impedance (blue dots) of (a) BiVO4 and (b) WO3/BiVO4 in the dark. The good correspondence of both cyclic voltammetry and impedance JV curves indicates good stability of the films during the impedance measurements (c) Resistance of the photoanodes in the dark and under illumination (10% sun) showing the dependence of film resistance on applied potential. The voltage-dependent resistance for the WO₃/BiVO₄ sample reflects

the slope of the steady-state j-V curve, since $R = \left(\frac{dj}{dV}\right)^{-1}$

Fitting of transient absorption kinetics as a function of applied potential

The kinetic traces were fitted using a model reported previously.¹ The following equation was used to model the kinetic data: -t

$$\Delta OD = \alpha t^{-b} + A_{TAS2} e^{\overline{\tau(TAS2)}} + c$$

(Equation S5)

The parameters for BiVO₄:

Potential (V _{RHE})	Power law			Single exponential function	
	а	b	С*	A _{TAS2}	$\tau_{(TAS2)}(s)$
0.6	3.47 x 10 ⁻⁶	-0.27	4.60 x 10 ⁻⁶	-	-
1.2	7.57 x 10 ⁻⁷	-0.41	4.60 x 10 ⁻⁶	2.05 x 10 ⁻⁵	0.18
1.4	3.85 x 10 ⁻⁶	-0.24	4.60 x 10 ⁻⁶	2.70 x 10 ⁻⁵	0.15

*the parameter c was used as an additional parameter for zeroing of kinetic traces

The parameters for WO₃/BiVO₄:

Potential (V _{RHE})	Power law			Single exponential function	
	а	b	С	A _{TAS2}	$ au_{(TAS2)}$ (s)
0.6	4.75 x 10 ⁻⁶	-0.22	-	-	-
1.2	3.53 x 10 ⁻⁷	-0.26	-	2.05 x 10 ⁻⁵	0.29
1.4	4.38 x 10 ⁻⁶	-0.25	-	2.70 x 10 ⁻⁵	0.22



Fig. S11 Bias dependent kinetic traces at 500 nm probing hole dynamics and the corresponding fittings, measured at 0.6 V_{RHE} (black), 1.2 V_{RHE} (red) and 1.4 V_{RHE} (blue) applied potential; recorded in potassium phosphate buffer (0.1 M), excited at 355 nm at pump fluence of 500 µJ/cm². The traces were fitted using a combination of a power law and single exponential functions with parameters noted in Table S1. (a) BiVO₄ and (b) WO₃/BiVO₄.

From the fittings, the single exponential time constants represents the dynamics of holes accumulated within the space charge layer of the film. Comparison of these time constants between the BiVO₄ and the WO₃/BiVO₄ films highlights that the time constant for the composite is higher than that of BiVO4. From previous studies, this slow component has been assigned to competing water oxidation and back electron/hole recombination processes and therefore the parameter is found to be applied-bias dependent. Being consistent with previous studies and the TPC data presented in this work, we observe that this back electron/hole recombination is significantly suppressed in the composite compared to bare BiVO₄.¹

References

1 Y. Ma, S. R. Pendlebury, A. Reynal, F. Le Formal and J. R. Durrant, Chem. Sci., 2014, 5, 2964– 2973.

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