Mass spectrometric identification and structural analysis of the third
generation synthetic cannabinoids on the UK market since the 2013
legislative ban
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21 Abstract

22 To examine the impact of the second legal ban on synthetic cannabinoids (SCs) in the UK in February 2013, we surveyed the UK legal high market just before and after the change in 23 24 legislation, looking for new SCs. The technique gas chromatography – mass spectrometry in electron ionization mode, most widely applied for analysis, was found to be insufficient for the 25 26 identification of several SCs, and therefore liquid chromatography – high resolution-mass spectrometry (LC-HR-MS) was required. LC-HR-MS(/MS) measurements of the protonated 27 molecule and product ions allowed the detection of up to 27 compounds as the third generation 28 29 SCs in the samples analysed as part of this study, including two unknown compounds that were 30 tentatively identified as F2201 and dealkyl-SDB006. Our results showed that banned compounds were removed from the market on the day when the ban was in place, and were 31 replaced by other SCs immediately after the ban. In only one occasion, a banned compound 32 33 (UR-144) was detected after the date when the new legislation came into place. It is also 34 noteworthy that regardless of the change in legislation, new compounds continued to enter the market. Product ion spectral information on the third generation SCs at different collision 35 36 energies given in this paper will be of help for forensic and clinical laboratories and will 37 facilitate the detection and identification of new SCs by laboratories of control. This 38 information is very valuable for law enforcement and policymakers and will be of help in future 39 prevention programs.

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41 Keywords

42 Synthetic cannabinoids (SCs), Legislation, Third generation SCs, LC–QTOF-MS/MS, F2201,
43 Dealkyl-SDB-006

45 **Introduction**

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Synthetic cannabinoids (SCs) have been introduced as drugs of abuse over the past years as a 47 48 legal alternative to cannabis. They are mainly being sold mixed with herbal substances, but can also be bought in resin-like material, as powder, and in liquid e-cigarette refills. The existence 49 50 of synthetic cannabinoid receptor agonists in the abuse market was first reported in 2009 by Japanese and German researchers ^[1-3]. In the UK, the first generation of SCs were banned in 51 52 2009. The rise of new compounds has made it more and more difficult for toxicologists to keep 53 up to date with standard analytical techniques and consequently has put users at risk when 54 abusing these substances. In addition, users often take new substances unknowingly, because branded products change their ingredients over time and, in particular, when new legislation is 55 56 put into place that bans existing SCs.

57 Analysis of street samples containing SCs has been undertaken by mass spectrometry (MS), coupled to either gas chromatography (GC) or liquid chromatography (LC) ^[4-8]. GC-58 MS has the advantage of the use of libraries under electron ionization (EI) conditions, making 59 60 it possible to tentatively identify a substance when no reference standard is available in the 61 laboratory. However, there is little possibility of identifying SCs by match in standardized GC-62 EI-MS libraries when such compound has not been previously reported. In this study, highresolution-mass spectrometry (HR-MS) has resulted in a valuable screening tool because it 63 provides sensitive full spectrum MS data with high mass resolution and mass accuracy ^[9-11]. 64 65 The information provided has made the tentative identification of the compounds detected feasible, with high degree of reliability, even without the use of reference standards. 66

67 New SCs often share a common structure made out of four basic parts: a hydrophobic 68 chain, an aromatic ring structure, a linker and a hydrophobic end-group. This common structure 69 makes it easier to market new compounds, because these parts are interchangeable; the European Monitoring Centre for Drugs and Drug Addiction (EMCDDA) reported 30 new SCs in 2014, making them the second most abundant group among the new psychoactive substances (NPS) reported in Europe ^[12]. In February 2013, a new ban came into place in the UK. It banned the so-called second generation of synthetic cannabinoid receptor agonists ^[13]. This legislation banned five substances and also, contained a generic ban on compounds, being described as "structurally derived from" 14 different compounds.

In this work, the effect of the 2013 ban on the UK market has been assessed. For this 76 purpose, 188 products were acquired in different periods, before and after the ban. The new 77 78 synthetic cannabinoids that emerged have been analysed by both GC-EI-MS and LC-HR-MS 79 with a hybrid quadrupole time-of-flight (QTOF) analyser. In many cases, GC-MS analysis was 80 insufficient to reach the unequivocal identity of the compound, and therefore LC-HR-MS was 81 required for identification. The different compounds identified before and after the ban are discussed, and accurate-mass spectral information of the third generation SCs using different 82 83 collision energies, useful for future analysis by control laboratories, is given.

85 Material and methods

86

87 Reagents and Chemicals

High-performance liquid chromatography (HPLC)-grade water was obtained by purifying demineralised water in a Milli-Q plus system from Millipore (Bedford, MA, USA); HPLCgrade methanol (MeOH), formic acid (HCOOH) and sodium hydroxide (NaOH > 99%) were acquired from Scharlau (Barcelona, Spain); leucine encephalin, methyl-*t*-butyl ether, quinoline and tripelenamine from Sigma-Aldrich (St. Louis, MO, USA); reference standards of SCs from Cayman Chemical (Ann Arbor, MI, USA), which has been dissolved in methanol at a concentration of 1 mg/mL.

95 Samples

96 Three periods in the sampling campaign can be distinguished: 1) December 1^{st} , 2012 – February 26th, 2013, the date when the new ban in the UK came into place, 49 samples were 97 bought just before the new legislation. 2) February 26th, 2013 – June 30th, 2013, 54 samples 98 99 were acquired immediately after the ban was in place. Samples from these first two sampling 100 campaigns were bought from websites and head-shops or acquired from police authorities. All 101 samples were powders or herbal material sold as smoking mixtures. 3) July 1st, 2013 – January 102 31st, 2015, 85 samples were bought from the Internet regardless of the description. Among 103 these samples were powders, herbal mixtures, one resin-like sample, and liquid e-cigarette 104 refills, which we subject to detailed analysis by LC-QTOF-MS(/MS).

105 Sample preparation

Approximately 1 mg of powder was dissolved in 1 mL of methanol in a 1.5 mL polypropylene tube. Solutions were then vortexed for 1 min or subjected to ultrasonic-assisted extraction for 15 min, and afterwards centrifuged at 8,000 rpm (6,030 g) for 5 min. For herbal mixtures, approximately 50 mg was mixed in 1 mL of methanol and vortexed for 30 min and centrifuged at 8,000 rpm (6,030 g) for 1 min. For LC–HR-MS, an aliquot of 100 μ L of the methanol extract was ten-fold diluted with water. For GC–MS analysis, a 10 μ L aliquot of the supernatant was diluted with 1 mL of methyl-*t*-butyl ether, containing 100 μ g/mL quinoline and tripelenamine.

113 Instrumentation

LC-QTOF-MS(/MS) analyses were performed using an Acquity Ultra-Performance Liquid 114 Chromatography UPLC system (Waters, Milford, MA, USA), which was interfaced to a hybrid 115 quadrupole-orthogonal acceleration-TOF mass spectrometer (QTOF XEVO G2, Waters 116 117 Micromass, Manchester, UK), using an orthogonal Z-spray-ESI interface operating in positive 118 ion mode. The chromatographic separation was performed using an Acquity UPLC BEH C18 119 analytical column (100×2.1 mm with 1.7 µm particle size; Waters). The column temperature 120 was set to 40 °C. The mobile phases used were $A = H_2O$ and B = MeOH, both with 0.01% 121 HCOOH, at a flow rate of 300 µL/min [more details in supplementary material (SM) and ^[7]].

GC-MS analyses were done using an Agilent 7890A GC with 5975C VL MSD (Agilent, Santa Clara, CA, USA) equipped with a split-splitless injector and an HP5-MS column (30 m length, 0.25 mm internal diameter, 0.25 μ m film thickness) and running on Agilent ChemStation. 1 μ L was injected using 5:1 split ratio. The column was held at 80°C for 4 min and then ramped up at 40°C/min to 290°C and held to a total run time of 40 min. A mass range of m/z 40 to 400 was scanned with scan-time 0.25 sec.

128 **Results and discussion**

129 LC-QTOF-MS(/MS) analysis of synthetic cannabinoids

In total, 27 new cannabinoids as the 3rd generation SCs were detected for the first time in 130 products sold on the UK market during the period just before and after the new ban came into 131 place (December 1st, 2012– January 31st, 2015). Analyses were first performed by GC– EI-MS. 132 133 It allowed several cannabinoids to be confirmed by the use of reference standards or tentatively identified by comparison with the GC-MS spectra included in Cayman Chemical Web page. 134 The same samples were also analyzed by LC-QTOF-MS(/MS) in order to gain more 135 136 confidence in the tentatively identified compounds and to study the fragmentation pathways of 137 these new cannabinoids. Data given in this paper refers only to LC-QTOF-MS(/MS) accurate-138 mass analysis, because this is the most relevant and new information considered of interest for the readers. 139

Making use of LC-QTOF-MS(/MS), the sample extracts were injected in full-140 acquisition mode working under MS^E mode, acquiring the low and high collision energy 141 spectra during the same injection^[7]. Narrow-window extracted ion chromatograms were then 142 143 obtained (± 100 ppm mass window) at the theoretical mass of the expected protonated 144 molecules. In all cases, mass errors obtained were lower than 5 ppm for the protonated 145 molecule. The sodium and potassium adducts were also commonly found. In a second step, MS/MS experiments were performed in an additional injection, obtaining the accurate-mass 146 147 product ion spectra after isolation of the precursor ion selected taking into account the structure of the cannabinoids. MS/MS experiments were much useful to justify the product ions obtained 148 149 and to propose the fragmentation pathway of the compounds. Variation in the amount of SC 150 present was not tested, as analysis was purely qualitative.

151 Below, our results are briefly commented, emphasizing the major product ions 152 observed. The exact masses, as shown in Tables 1-5, were used for the discussion of the product ions observed and to facilitate the reading. Furthermore, to help the discussion on the chemical
structures of cannabinoids identified, the compounds were classified in different groups
considering their fragmentation pattern (Fig. 1). For those interested in more details regarding
fragmentation, we recommend reading the information given in the supplementary material
(SM). Figures included in SM (Figures S.1-S.25) show the accurate/experimental masses
provided by LC–QTOF-MS(/MS).

159

160 Cannabinoids containing an adamantyl group linked by an amide and SDB-006

This group of cannabinoids includes four compounds: APICA, 5F-APICA, APINACA and 5F-APINACA, all of which have an adamantyl group linked to the core by an amide bond. The core can be an indazole (APINACA and 5F-APINACA) or indole (APICA and 5F-APICA) and the tail a pentyl (APINACA and APICA) or a 5-fluoropentyl (5F-APICA and 5F-APINACA) chain (Fig. 1).

In all four compounds identified, the most abundant product ion at 30 eV corresponded to the adamantyl group (ion C, m/z 135.1174, C₁₀H₁₅) (Fig. 2a). Table 1 shows the product ions as well as the corresponding elemental compositions for all cannabinoids included in this group. Regarding SDB-006 (m/z 321.1967), the product ion resulting from the breaking of the central amide (m/z 214.1232) and that corresponding to the pentyl indole group (m/z 188.1439) are the most abundant ones (Fig. 2b; Table 1). LC–QTOF-MS(/MS)spectra at different collision energies for all cannabinoids in this section are included in SM (Figs. S.1-S.5).

173

174 Cannabinoids with a quinolinyl ester, NM-2201 and 5F-MN-18

Four cannabinoids belong to this group of compounds containing a quinolinyl ester: PB-22,
5F-PB-22, BB-22 and 5F-NPB-22 (Table 2). In addition, two related compounds were also
identified and are discussed here. NM-2201 is structurally similar to these cannabinoids; the

only difference is the naphthalene group instead of a quinolinyl group. 5F-MN-18, which is closely related to NM-2201, has an amide linkage. This resulted in a similar fragmentation pattern. The compound FUB-PB-22 also contained a quinolinyl ester; however it will be discussed below as a cannabinoid with a *para*-fluorotoluene chain because the mass spectra were quite similar to other cannabinoids containing this moiety.

183 For PB-22, 5F-PB-22, BB-22, 5F-NPB-22 and NM-2201, the main product ion (B) was 184 formed by cleavage of the ester bond (Table 2; Fig. 3). Another important product ion (E) was 185 related to presence of an indole or indazole in the structure. For indole-based structures, PB-186 22, 5F-PB-22, BB-22 and NM-2201, ion E at *m/z* 144.0449 (C₉H₆NO) was observed, whereas for indazole-based structures and as 5F-NPB-22 and 5F-MN-18, the product ions E 187 188 corresponded to m/z 145.0402 (C₈H₅N₂O). Additionally, for 5F-NPB-22 and 5F-MN-18, the 189 m/z 213.1028 (ion C) was observed, corresponding to the loss of hydrogen fluoride (HF) from 190 *m/z* 233 (Figs. S.6-S.11).

191

192 Cannabinoids with a branched end group

Most cannabinoids have a ring structure as end group (naphthalene, quilolinyl, adamantyl, etc.),
but nine new cannabinoids from this study have a branched side chain instead: ADB-PINACA,
AB-PINACA, 5F-AB-PINACA, 5F-Cumyl-PINACA, AB-CHMINACA, MDMB-CHMICA,
and 5F-AMB as well as AB-FUBINACA and ADB-FUBINACA (Fig. 1). The latter two
contain a *para*-fluorotoluene chain and will be discussed in the next section.

The most prominent product ion (ion D) in all spectra was the result of the cleavage of the central amide bond (Table 3; Fig. 4). The m/z 145.0398 (G) was also abundant in all spectra (C₈H₅N₂O), and resulted from the carbonyl-indazole group (except m/z 144.0441 for MDMB-CHMINACA due to the indole group, C₉H₆NO) after double cleavage at the central amide bond and at the root of the pentyl or 5-fluoropentyl side chain. (Figs. S.12-S.18).

204 Cannabinoids with a *para*-fluorotoluene chain

AB-FUBINACA, ADB-FUBINACA and FUB-PB-22 all have a *para*-fluorotoluene side chain (Figs. 1, 5) and shared common fragmentation pathways. At higher collision energies, where fragmentation is promoted, these three compounds showed two abundant product ions. The first (m/z 253.0777) was the result of the cleavage of the central amide bond (ion D), for AB-FUBINACA and ADB-FUBINACA, or of the ester (m/z 252.0825) for FUB-PB-22. The second, at m/z 109.0454 (ion E, C₇H₆F), was due to the presence of the *para*-fluorotoluene side chain (Table 4; Figs. S.19-S.21).

212

213 Cannabinoids with two chromatographic peaks

214 Two chromatographic peaks were observed in the LC-QTOF-MS chromatograms at the 215 expected *m/z* for five compounds, concretely AB-FUBINACA, ADB-PINACA, AB-PINACA, 5F-AB-PINACA, and AB-CHMINACA (Table 3; Fig. 5a). For these compounds, the two 216 217 chromatographic peaks presented different fragmentation, being all compatible with the structure of the corresponding cannabinoid. All of them possess a terminal amino group and an 218 enantiomeric carbon at the linker part. Moreover, some common product ions were also 219 220 observed, but with different relative intensities. In all cases, the first chromatographic peak 221 presented an abundant protonated molecule, whereas the second presented as peak base at 10 222 eV with the product ion corresponding to the loss of NH₃. This did not happen in GC–MS, 223 where only one chromatographic peak was observed. This might be explained by the 224 occurrence of rotamers. However, isolation and further spectroscopic experiments is needed to 225 confirm and support this hypothesis.

227 Cannabinoids with a carbonyl link

THJ-018 and THJ-2201 have similar structures, differing only in the absence and presence of a fluorine atom at the end of the chain, respectively. The main product ions were ion F, at m/z145.0402 (C₈H₅N₂O), due to the carbonyl-indazole group, and ion B, which corresponded to the loss of the naphthalene group (C₁₀H₈) (Table 5). In the case of THJ-2201 (Fig. 6a), a subsequent loss of HF was also observed (ion C). (Figs. S.22-S.23).

Other two cannabinoids were included in the same group, EG-018 (Fig. 6b) and BZ-234 2201. They present similar fragmentation (Table 5), with the major product ions being m/z235 155.0497 (C₁₁H₇O, corresponding to the carbonyl-naphthalene group) and 127.0548 236 (corresponding to the naphthalene group) (Figs S.24-S.25).

237

238 Unidentified novel synthetic cannabinoids

In addition to the SCs identified and discussed above, two samples contained unknown cannabinoids. After initial GC–MS experiments, their identification was not possible at the time of analysis. Therefore, subsequent analysis by LC–HR-MS was compulsory to elucidate the chemical structures of these compounds (for details see SM).

The unidentified compound **1** was found in an herbal sample. After studying its fragmentation by both GC– EI-MS and LC–QTOF-MS(/MS) and taking into account the fragmentation observed for other cannabinoids, we suggest it to be F2201 (Fig. 7). This compound is not new actually (already administered as CAS 1391485-39-4), but it had never been seen on the drug market up to the moment of the analysis.

GC–MS analysis of another herbal sample showed the presence of two compounds: SDB-006 and an unknown compound **2**. After studying the MS/MS accurate-mass spectra obtained by LC–QTOF-MS(/MS) and comparing its fragmentation with that observed for the other cannabinoids, we suggest the compound to be dealkyl-SDB-006 (Fig. 8).

Third generation synthetic cannabinoids on the UK market detected during the overall study

As most synthetic cannabinoids on the UK market were banned by the legislation coming into place in February 2013^[13], it is not surprising that new SCs entered the UK market just before or mostly after the ban. As shown in the previous sections, a total of 25 cannabinoids (see Fig. 1) plus two unknown compounds (third generation SCs), new to the UK market, were detected in this work.

Three out of 25 compounds were found to be not previously reported cannabinoids and were identified in samples collected before the ban was in place (sampling period 1): APINACA (also known as AKB48), 5F-APINACA (5F-AKB48) and 5F-APICA (STS-135). These three compounds are closely related, because 5F-APINACA replaces a hydrogen atom by a fluoride atom in APINACA, and 5F-APICA is the indole analogue of 5F-APINACA.

Three other non-reported cannabinoids were detected in the four months immediately after the ban (sampling period 2): BB-22, PB-22 and 5F-PB-22. Again, these compounds are closely related; as only the side chain was different; BB-22 has a methylcyclohexyl side chain, PB-22 a pentyl side chain, and 5F-PB-22 a 5-fluoropentyl side chain.

269 Up to January 2015 (sampling period 3), a further 21 SCs, not previously reported in 270 the UK market, were identified: 5F-Cumyl-PINACA, FUB-PB-22, 5F-NPB-22, EG-018, THJ-271 018, THJ-2201, NM-2201, BZ-2201, F2201, SDB-006, dealkyl-SDB-006, 5F-MN-18, APICA, 272 MDMB-CHMICA (also incorrectly known as MMB-CHMINACA), AB-CHMINACA, AB-PINACA, 5F-AB-PINACA, ADB-PINACA, 5F-AMB, AB-FUBINACA and ADB-273 FUBINACA. As it can be seen, some of these compounds were structurally-related to earlier 274 found cannabinoids. For example, in 5F-Cumyl-PINACA, the adamantyl group of 5F-275 APINACA is replaced by a cumyl group. FUB-PB-22 replaces the fluoropentyl chain of 5F-276

PB-22 with a *para*-fluorotoluene group. Two additional cannabinoids identified, EG-018, and
SDB-006, were not related to the previously reported findings. Finally, the two compounds,
not identified after initial GC–MS analysis, could be tentatively reported as F2201 and dealkylSDB-006 in this study.

It remains a question how effective the new legislation has been. Several compounds 281 282 disappeared from the market, and as such, the ban already worked, but these products have been replaced rapidly by new compounds. However, the emergence of new compounds is not 283 284 solely due the legislative change, as many new cannabinoids emerged on the UK market 285 without new laws. Similar to what occurred in Japan, where new cannabinoids entered the market without a ban^[14]. Other driving factors could be a legislative change elsewhere, 286 287 commercial purposes and/or supply problems. In any case, it seems clear that rapid 288 replacements exist in the market of SCs, with continuous appearance of new compounds, 289 making their control troublesome for analytical laboratories.

290

291 Sampling period immediately before and after the ban

In the first sampling period, 49 samples were acquired between December 1st, 2012 and February 26th, 2013, when the date the new ban in the UK came into place. Another 54 samples were acquired in the second sampling period after the ban and before June 30th, 2013. Data obtained in the analysis of these 103 samples were used to evaluate the immediate effect of the ban on the market of SCs. Identification of the compounds found in all these samples was supported by the use of reference standards or known samples. Ten different SCs were identified in 87 samples, while the remaining 16 did not contain SCs.

Before the ban, 88% of SCs found corresponded to compounds that were subsequently prohibited by the 2013 legislation. After the ban, 98% of the occurrences were of new (legal) SCs, *i.e.*, compounds not controlled within the 2013 legislation. The only sample to contain a

302 banned substance after the prohibition was a powder labelled LY2183240, which contained 303 UR-144 (Fig. 9). It is possible that the person selling this sample was simply stuck with the 304 leftover UR-144 when the ban came into place, and decided to sell it as a non-controlled 305 substance, thus still making money for an otherwise worthless amount of the SC. This would 306 mean that the user is not aware that they are buying an illegal substance and is not informed as 307 to what drug they are taking, putting them at more risk. According to these data, the change in 308 legislation seemed to have the desired effect of clearing the market of the banned products. 309 This is not a surprising observation, as SCs are sold as legal highs, and it is within the vendors' interest not to sell illegal substances. This was also seen in a study by Kikura-Hanajiri et al.^[14], 310 who investigated the cannabinoid market in Japan following a change in legislation. 311

312 When comparing the number of active ingredients per sample, there is a clear 313 distinction to be made between powders (advertised as a pure compound) and herbal smoking 314 mixtures. All the 16 powders analysed during this period contained one active ingredient. 315 However, in herbal mixtures, the number of SCs that were present was variable. Before the 316 ban, 33% of the herbal blends analysed did not contain any drugs, 45% contained one active ingredient, while 22% contained two SCs. After the ban, the great majority of samples (83%) 317 318 contained only one active ingredient, while 15% contained no drugs; just 2% of the samples 319 contained two SCs. It seems that immediately after the ban, manufacturers were using only one ingredient per sample. It might be due to a fear of mixing compounds that were relatively 320 321 unknown for them.

In this sampling round, several products with the same brand name were sampled more than once, because they came from a different source or from different times. The results showed that an important number of them changed ingredients and this was not always due to the change in legislation. Before the ban, three brands were sampled more than once, and for all of them, different compositions were found for the analyzed samples. "Mary Joy Evolution" 327 contained UR-144 and MAM2201 or only MAM2201; "Blue cheese" contained XLR11 or XLR11 and MAM2201; and "Abyss" contained either a mixture of UR-144 and MAM2201 or 328 only MAM2201. Only one brand, "Doob" was available before and after the change in 329 330 legislation. Before the ban, it contained AM2201 or a combination of AM2201 and UR-144. It is unclear why manufacturer decided to change the product, but it may be due to a supply 331 332 problem with one of the ingredients or simply due to profit. However, the sample of "Doob" we obtained after the ban contained only 5F-APINACA, a different SC. As UR-144 and 333 334 AM2201 were both banned, it is likely that the manufacturer switched to another SC. Changing 335 of active ingredients can put users at risk as other ingredients may have different pharmacokinetic or dynamic properties. 336

After the ban, four brands were sampled more than once and two of them did not change their ingredients, while two did. "Clockwork Orange" and "Chillem Blue" always contained 5F-AKB48 as the only active ingredient, while "Dutchy" contained either 5F-AKB48 or AKB48 and "Magic Dragon" contained either 5F-AKB48 or 5F-PB-22. Hence, it seems that it was not only due to the ban that manufacturers decidde to switch to other ingredients.

342

343 New physical forms for synthetic cannabinoids

During the three sampling campaigns (December 2012-January 2015), most SCs found on the 344 345 abuse market were sold as herbal smoking mixtures (*i.e.*, dried herbs sprayed or mixed with 346 SCs) or as powders. In the latter case, it is believed that the user mixes it with herbs before consumption. However, during this period, two other forms were found on the UK market. E-347 cigarette refills (Fig. 10a) are meant to be loaded into an electronic cigarette; they are present 348 349 as solutions in a volatile solvent, such as propylene glycol. However, the refill purchased from a UK website contained the SC, 5F-Cumyl-PINACA. It is unknown for us whether this method 350 351 of drug consumption is less or more harmful than the traditional smoking of dried herbs.

- 352 Another form that was encountered was a resin-like material laced with SCs, such as "Squidgy"
- 353 (Fig. 10b). This sample contained 5F-AB-PINACA. It is unclear what the resin itself is made
- 354 of, but it seems to be marketed to resemble hashish (cannabis resin).

357 **Conclusions**

In this work, we have surveyed the UK legal high market between December 2012 and January 358 359 2015. Our results reveal that the legislative ban succeeded in pushing the corresponding 360 compounds from the UK market, but only one of the banned compounds (UR-144) detected after the date when the ban came into place (February 26th, 2013). However, a risk of banning 361 362 existing compounds is the emergence of new compounds (which as our result show, did 363 happen), with unknown and potentially more dangerous effects. Another problem associated with banning compounds is the lack of information for both drug users and healthcare workers. 364 365 Drug users do not know what they are taking after a ban, because branded products change 366 ingredients or vendors mislabel products to be able to sell leftover stock. For healthcare and 367 forensic professionals, there is little knowledge on new compounds, and they might be missed 368 in drug screenings.

369 In the face of the continuous changes in the products, it is necessary to reinforce 370 analytical measurements for the monitoring of SCs to be able of efficiently detect and identify the new products that are substituting the already known compounds present in the market. 371 372 Data presented for the third generation SCs in this work are useful not only for the monitoring 373 of the SCs that we have found in the samples, but also to facilitate the detection and tentative 374 identification of chemically-related compounds that share common product ions, which have 375 been exemplified in tentative identification of unknown compound 1 and 2 to be F2201 and 376 dealkyl-SDB-006, respectively, in this study. Product ion spectra for 27 SCs obtained from the 377 third generation SC products using five different collision energies have been presented for 378 such purpose. Such detailed data have not been reported to our knowledge. In addition, the 379 appearance of two chromatographic (total ion current chromatograms or extracted ion 380 chromatograms) peaks with a common octadecyl column appeared for AB-FUBINACA, ADB-381 PINACA, AB-PINACA, 5F-AB-PINACA and AB- CHMINACA all with the presence of a

- terminal amino group together with an enantiomeric carbon at the linker part merits mentioning
- again in this study.

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400	
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TABLES

447 Table 1 Product ions obtained by liquid chromatography – quadrupole time-of-flight- tandem mass spectrometry (LC–QTOF-MS/MS) for
 448 synthetic cannabinoids (SCs) with adamantyl amide groups, showing their exact mass and elemental compositions

Compound	$\mathbf{A}=[\mathbf{M}+\mathbf{H}]^{+}$	В	С	D	Ε	F	RT
$\Delta PIC \Delta (2NE1)$	365.2564	214.1232	135.1174	107.0861	93.0704	79.0548	14.9
	$C_{24}H_{33}N_2O$	C ₁₄ H ₁₆ NO	$C_{10}H_{15}$	C_8H_{11}	C7H9	C_6H_7	17.7
5E-APICA (STS-135)	383.2499	232.1138	135.1174	107.0861	93.0704	79.0548	1/1 1
JI-AI ICA (313-133)	$C_{24}H_{32}N_2OF$	C ₁₄ H ₁₅ NOF	$C_{10}H_{15}$	C_8H_{11}	C ₇ H ₉	C_6H_7	14.1
$\Delta PINACA (\Delta KB/8)$	366.2545		135.1174	107.0861	93.0704	79.0548	15.5
AI INACA (AKD40)	$C_{23}H_{32}N_{3}O$		$C_{10}H_{15}$	C_8H_{11}	C ₇ H ₉	C_6H_7	15.5
5F-APINACA (5F-	384.2451		135.1174	107.0861	93.0704	79.0548	14.5
AKB48)	C ₂₃ H ₃₁ N ₃ OF		$C_{10}H_{15}$	C_8H_{11}	C7H9	C_6H_7	17.5
SDB-006	321.1967	214.1232	188.1439	144.0449	132.0813	91.0548	12.9
	$C_{21}H_{25}N_2O$	C ₁₄ H ₁₆ NO	$C_{13}H_{18}N$	C ₉ H ₆ NO	C9H10N	C7H7	12.7

RT retention time in minutes

Compound	$A=[M+H]^+$	В	С	D	Ε	F	G	RT	
DB 22	359.1770	214.1232			144.0449	116.0500		13.7	
1 D-22	$C_{23}H_{23}N_2O_2$	$C_{14}H_{16}NO$			C ₉ H ₆ NO	C ₈ H ₆ N		13.7	
5E_PB_22	377.1665	232.1138			144.0449	116.0500		12.5	
J1-1 D-22	$C_{23}H_{22}N_2O_2F$	C ₁₄ H ₁₅ NOF			C ₉ H ₆ N	C ₈ H ₆ N		12.5	
BB_22	385.1916	240.1388			144.0449	116.0500		14.2	
DD-22	$C_{25}H_{25}N_2O_2$	C ₁₆ H ₁₈ NO			C ₉ H ₆ NO	C ₈ H ₆ N		14.2	
NM_2201	376.1711	232.1138			144.0449	116.0500		14.1	
11111-2201	C ₂₄ H ₂₃ NO ₂ F	C ₁₄ H ₁₅ NOF			C ₉ H ₆ NO	C ₈ H ₆ N		14.1	
SE NDR 22	378.1618	233.1090	213.1028	177.0453	145.0402	117.0453	90.0344	11.0	
JI-INI D-22	$C_{22}H_{21}N_3O_2F$	$C_{13}H_{14}N_2OF$	$C_{13}H_{13}N_2O$	$C_{12}H_5N_2$	C ₈ H ₅ N ₂ O	$C_7H_5N_2$	C_6H_4N	11.9	
5E MN 18	376.1825	233.1090	213.1028	177.0453	145.0402	117.0453	90.0344	13.4	
51'-10119-10	$C_{23}H_{22}N_3OF$	$C_{13}H_{14}N_2OF$	$C_{13}H_{13}N_2O$	$C_{12}H_5N_2$	$C_8H_5N_2O$	$C_7H_5N_2$	C ₆ H ₄ N	13.4	

Table 2. Product ions obtained by LC–QTOF-MS/MS for SCs with quinolyl esters, showing their exact mass and elemental compositions

Table 3 Product ions obtained by LC–QTOF-MS/MS for SCs with branched end groups, showing their exact mass and elemental
 compositions

Compound	$A=[M+H]^+$	В	С	D	Ε	F	G	RT
	345.2291	328.2025	300.2076	215.1184			145.0398	12.0
ADD-FINACA	$C_{19}H_{29}N_4O_2$	$C_{19}H_{26}N_3O_2$	C ₁₈ H ₂₆ N ₃ O	$C_{13}H_{15}N_2O$			C ₈ H ₅ N ₂ O	13.0
	331.2134	314.1869	286.1919	215.1184			145.0398	12.2
AD-FINACA	$C_{18}H_{27}N_4O_2$	$C_{18}H_{24}N_3O_2$	C ₁₇ H ₂₄ N ₃ O	$C_{13}H_{15}N_2O$			C ₈ H ₅ N ₂ O	12.2
5E AR DINACA ^a	349.2040	332.1774	304.1825	233.1090	213.1028	177.0463	145.0398	11.0
JI-AD-FINACA	$C_{18}H_{26}N_4O_2F$	$C_{18}H_{23}N_3O_2F$	C ₁₇ H ₂₃ N ₃ OF	$C_{13}H_{14}N_2OF$	$C_{13}H_{13}N_2O$	$C_{12}H_5N_2$	C ₈ H ₅ N ₂ O	11.0
5E-Cumyl-PINACA	368.2138			233.1090	213.1028	177.0463	145.0398	13.0
JI-Cully-I INACA	C ₂₂ H ₂₇ N ₃ OF			$C_{13}H_{14}N_2OF$	$C_{13}H_{13}N_2O$	$C_{12}H_5N_2$	C ₈ H ₅ N ₂ O	15.0
	357.2291	340.2025	312.2076	241.1341			145.0398	13.1
	$C_{20}H_{29}N_4O_2$	$C_{20}H_{26}N_3O_2$	C19H26N3O	$C_{15}H_{17}N_2O$			C ₈ H ₅ N ₂ O	13.1
MDMB_CHMICA	385.2491			240.1388			144.0441	1/ 1
WDWD-CITIVIICA	$C_{23}H_{33}N_2O_3$			C ₁₆ H ₁₈ NO			C ₉ H ₆ NO	14.1
5E AMB	364.2036	332.1774	304.1825	233.1090	213.1028	177.0463	145.0398	12.1
	C19H27N3O3F	$C_{18}H_{23}N_3O_2F$	C17H23N3OF	$C_{13}H_{14}N_2OF$	$C_{13}H_{13}N_2O$	$C_{12}H_5N_2$	C ₈ H ₅ N ₂ O	12.1

^a Two chromatographic peaks were observed in the LC–QTOF-MS chromatogram for these compounds. Only the product ions for the most
 intense one are shown. For additional details, see supplementary material (SM)

Table 4 Product ions obtained by LC–QTOF-MS/MS for SCs with a *para*-fluorotoluene group, showing their exact mass and elemental
 compositions

Compound	$A=[M+H]^+$	В	С	D	Ε	RT
AR EURINACA ^a	369.1727	352.1461	324.1512	253.0777	109.0454	11.5
AD-FUDINACA	$C_{20}H_{22}N_4O_2F$	$C_{20}H_{19}N_3O_2F$	$C_{19}H_{19}N_3OF$	$C_{15}H_{10}N_2OF$	C_7H_6F	11.3
	383.1883	366.1618	338.1669	253.0777	109.0454	11.0
ADD-FUDINACA	$C_{21}H_{24}N_4O_2F$	$C_{21}H_{21}N_3O_2F$	$C_{20}H_{21}N_3OF$	$C_{15}H_{10}N_2OF$	C_7H_6F	11.9
	397.1352			252.0825	109.0454	12.0
ΓUD-FD-22	$C_{25}H_{18}N_2O_2F$			C ₁₆ H ₁₁ NOF	C_7H_6F	12.9

460 ^a Two chromatographic peaks were observed in the UPLC-QTOF MS chromatogram for these compounds. Only the product ions for the

461 most intense one are shown. For additional details, see SM

Compound	A=[M+H] ⁺	В	С	D	Ε	F	G	Н	Ι	RT	
THI_018	343.18109	215.1184				145.0402	127.0548	117.0453	90.0344	147	
1115-010	$C_{23}H_{23}N_2O$	$C_{13}H_{15}N_2O$				$C_8H_5N_2O$	$C_{10}H_7$	$C_7H_5N_2$	C_6H_4N	14./	
тні 2201	361.1716	233.1090	213.1028		177.0453	145.0402	127.0548	117.0453	90.0344	13.6	
1115-2201	$C_{23}H_{22}N_2OF$	$C_{13}H_{14}N_2OF$	$C_{13}H_{13}N_2O$		$C_{12}H_5N_2$	$C_8H_5N_2O$	$C_{10}H_7$	$C_7H_5N_2$	C_6H_4N	15.0	
B7-2201	361.1716	233.1090		155.0497	177.0453	145.0402	127.0548	117.0453	90.0344	13.2	
DZ-2201	$C_{23}H_{22}N_2OF$	$C_{13}H_{14}N_2OF$		C11H7O	$C_{12}H_5N_2$	C ₈ H ₅ N ₂ O	$C_{10}H_7$	$C_7H_5N_2$	C_6H_4N	13.2	
FG-018	392.2014			155.0497		145.0402	127.0548			15.8	
LO-010	$C_{28}H_{26}NO$			$C_{11}H_7O$		$C_8H_5N_2O$	$C_{10}H_{7}$			13.0	

Table 5 Product ions obtained by LC–QTOF-MS/MS for SCs with a carbonyl link, showing their exact mass and elemental compositions

466 **FIGURE CAPTIONS**

- Fig. 1 Structures of synthetic cannabinoids (SCs) classified according to the structure and
 fragmentation
- 469 Fig. 2 Liquid chromatography quadrupole time-of-flight-tandem mass spectrometry (LC–
- 470 QTOF-MS/MS) spectra of a APICA and b SDB-006, at different collision energies with
 471 product ions identified together with the probable fragmentation modes
- 472 Fig. 3 LC–QTOF-MS/MS spectra of a PB-22 and b 5F-NPB-22, at different collision energies
 473 with product ions identified together with the probable fragmentation modes
- 474 Fig. 4 LC–QTOF-MS/MS spectra of a ADB-PINACA and b 5F-AB-PINACA, at different
- 475 collision energies with product ions identified together with the probable fragmentation476 modes
- 477 Fig. 5 a Total ion current chromatographic peaks obtained for AB-FUBINACA by LC–QTOF-
- 478 MS and **b**, **c** MS/MS spectra obtained for each chromatographic peak at different 479 collision energies with product ions identified
- 480 **Fig. 6** LC–QTOF-MS/MS spectra of **a** THJ-2201 and **b** EG-018, at different collision energies
- 481 with product ions identified together with the probable fragmentation modes
- Fig. 7 Tentative identification of unidentified compound 1. a Gas chromatography mass
 spectrometry spectra, and b LC–QTOF-MS/MS spectra at different collision energies
- 484 **Fig. 8** LC–QTOF-MS/MS spectra at different collision energies for unidentified compound **2**
- 485 Fig. 9 Compounds found in sampling campaigns between December 2012 and June 2013
- 486 (sampling periods 1 and 2) showing the SC profiles before and after the 2013 legislation
 487 in the UK
- 488 Fig. 10 a A product sold as an e-cigarette refill, and b a product sold as a resin

SC with adamantyl amide group				SC with quinolyl esters				SC with a para-fluorotoluene chain			
	X R ₂ N R ₁	$O = X^{N-R_1}$ R_2 R_3				O HN R ₂					
Compound	X	R ₁	R ₂	Compound	X	R ₁	R ₂	R3	Compound X		\mathbf{R}_2
APICA (2NE1)	С	Pentyl	Adamantyl	PB-22	С	Pentyl	0	N	AB-FUBINACA	N	H ₃ C CH ₃
5F-APICA (STS-135)	C	5-Fluoropentyl	Adamantyl	5F-PB-22	C	5-Fluoropentyl	0	N			1-(Aminocarbonyl)-2- methylpropyl
APINACA (AKB48)	N	Pentyl	Adamantyl	BB-22	С	Cyclohexylmethyl	0	N	FUB-PB-22	C	
5F-APINACA (5F-AKB48)	N	5-Fluoropentyl	Adamantyl	NM-2201	C	5-Fluoropentyl	0	С			8-Quinolinol, this group replaces NH-R ₂
SDB-006			Pentyl	5F-NPB-22	N	5-Fluoropentyl	0	N			NH ₂
	C	C Pentyl		5F-MN-18	N	5-Fluoropentyl	N	С	ADB-FUBINACA	N	H ₃ C — CH ₃ CH ₃ 1-(Aminocarbonyl)-2,2- dimethylpropyl

Fig. 1

SC with branched	SC with a carbonyl link								
	0	N-R1							
Compound	X	R 1	\mathbf{R}_2	Compound	X	\mathbf{R}_1	\mathbf{R}_2	Compound	\mathbf{R}_1
ADB-PINACA	N	Pentyl	1-(Aminocarbonyl)-2,2- dimethylpropyl	AB-CHMINACA	N	Cyclohexylmethy	1-(Aminocarbonyl)-	THJ-018	Pentyl
AB-PINACA	N	Pentyl	1-(Aminocarbonyl)-2- methylpropyl		1	1	2-methylpropyl	THJ-2201	5-Fluoropentyl
5F-AB- PINACA	N	5-Fluoropentyl	1-(Aminocarbonyl)-2- methylpropyl	MDMB-CHMICA	С	Cyclohexylmethy l	H ₃ C - CH ₃	EG-018	
5F-Cumyl- PINACA	N	5-Fluoropentyl	H ₃ C	5F-AMB	N	5-Fluoropentyl	June CH3	BZ-2201	O V V

Fig. 1 (continue)



Fig. 2 a





Fig. 3



Fig. 4





Fig. 5





Fig. 6



Fig. 7













