Analytical Methods

Cite this: Anal. Methods, 2012, 4, 196

www.rsc.org/methods PAPER

Target and non-target screening strategies for organic contaminants, residues and illicit substances in food, environmental and human biological samples by UHPLC-QTOF-MS†

Ramon Díaz, María Ibáñez, Juan V. Sancho and Félix Hernández*

Received 28th June 2011, Accepted 11th October 2011

DOI: 10.1039/c1ay05385j

In this paper, we illustrate the potential of ultra-high performance liquid chromatography (UHPLC) coupled with hybrid quadrupole time-of-flight mass spectrometry (QTOF MS) for large scale screening of organic contaminants in different types of samples. Thanks to the full-spectrum acquisition at satisfactory sensitivity, it is feasible to apply both (post)-target and non-target approaches for the rapid qualitative screening of organic pollutants in food, biological and environmental samples. Different strategies have been applied and compared in this work. The first approach consists of target screening based on automatically extracting the exact analyte masses with a narrow mass window (± 10 mDa). The selection of analytes can be made after MS acquisition as non-specific analyte information is required when injecting the samples. The second, non-targeted approach, consists of a first component detection step followed by the search of the detected components in home-made spectral libraries. In this work, two types of libraries have been evaluated: a theoretical database, including the molecular formula of a large number of pollutants (\sim 1000), and an empirical mass spectra library which includes a lower number of compounds for which reference standards were available. In all cases the confidence of the identification process was excellent, thanks to the value of information given in QTOF MS^E acquisition mode (i.e. simultaneous acquisition of low and high energy TOF MS spectra in a unique run). Both, target and non-target approaches, are complementary and both have advantages and drawbacks. Their application to different types of samples has allowed the detection of diverse organic compounds, for example the mycotoxin fumonisin B1 in food samples, cocaine and several metabolites in human urine, as well as several pesticides, antibiotics and drugs of abuse in urban wastewater.

Introduction

Nowadays, liquid chromatography (LC) hyphenated to mass spectrometry (MS) using a variety of mass analyzers is the technique of choice for the investigation of organic contaminants in most analyte/sample matrix combinations in environmental, food or toxicology fields. Mass analyzers used include triple quadrupole (QqQ),¹⁻⁶ time-of-flight (TOF), hybrid quadrupole time-of-flight (QTOF),^{2,7-11} quadrupole-linear ion trap (QLIT)^{3,12,13} or Orbitrap.¹⁴ Many examples can be found in the literature dealing with pesticide residue analysis in environmental,¹⁵ food^{16,17} or biological samples,¹⁸ using LC-MS based methods. Emerging contaminants, such as pharmaceuticals^{1,5} or drugs of abuse,¹⁹ amongst others, are increasingly being monitored in the environment by LC-MS because their medium-to-high polarity and low volatility make their determination fit

better with LC. Similarly this applies to metabolites and transformation products, which are generally more polar than their parent molecules.

LC-tandem MS (LC-MS/MS) operating in Selected Reaction Monitoring mode (SRM) with QqQ analysers are the workhorses nowadays in target analysis. 20 LC-MS/MS methods rarely include more than two hundred analytes, 2,21,22 and with a few exceptions,²³ most of them are focused on a single family of contaminants. Excellent sensitivity and notable selectivity are achieved by LC tandem MS, allowing reliable quantification and identification of a considerable number of compounds. However, the presence of other contaminants that might be present in the samples would be ignored in LC-MS/MS under SRM mode (the most common approach), due to the analyte-specific information acquired. There is a need in the field of public health to develop reliable methods for large-scale screening that are capable of detecting and identifying a large number of hazardous compounds that can potentially be present in environmental and food samples. For this purpose, full spectrum acquisition techniques capable of providing accurate mass measurements are a great help.

Research Institute for Pesticide and Water, University Jaume I, Av. Sos Baynat SIN, 12071 Castellón, Spain. E-mail: felix.hernandez@qfa.uji.es; Fax: +34 964387368; Tel: +34 964387366

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c1ay05385j

To solve the limitations of unit resolution mass spectrometers, two main alternatives, based on the use of high-resolution MS instruments, are of note at present: the time-of-flight²⁴ and Orbitrap^{25,26} analysers. Both provide full spectrum accuratemass data at satisfactory sensitivity. These capabilities are very helpful for detecting and identifying not only priority known pollutants but many other unknown contaminants that might be a risk for human health.27-29

Although quantitative applications have been reported using LC-TOF MS or LC-QTOF MS, 8,10,30 quantification does not seem to be the most attractive feature of these analysers. This may be due to the higher limits of detection and narrower linear dynamic range in comparison to QqQ analysers. One of the most interesting applications of TOF MS deals with the wide-scope screening of a large number of contaminants and residues in different types of samples, as that allows a significant amount of useful information on ionisable compounds present in the sample to be obtained.³¹ Generic (universal) sample treatments and chromatographic separations are required to broaden the scope of the method to as many compounds as possible. Besides, the elevated acquisition speed of TOF makes it compatible with ultra-high pressure (Ultra-Performance) liquid chromatography (UHPLC/UPLC). This technique provides fast, highresolution separations that will hopefully minimize matrix effects and render high mass spectra purity, improving the screening process.

Different strategies can be used to extract analytical information from full-acquisition accurate mass data. A genuine nontarget analysis involves the automated component detection from the total ion chromatogram (TIC) and the mass spectra deconvolution for a subsequent comparison with mass spectral libraries. Nevertheless, electrospray ionization (ESI) is not an ion source as stable and reproducible as electron ionization, 32 and commercial, standardized ESI mass spectra library are not available. Instead, theoretical mass spectra libraries, based on the molecular formula database, can be built which facilitate increasing the number of compounds that can be searched. These use accurate mass measurements and isotopic pattern information for identification. Home-made empirical libraries can also be used, but these normally include much fewer compounds due to the need to inject standards. These experimental libraries offer fragmentation and retention time information as well, providing more confidence in the compound identification process.33 However, the possibility of detecting and identifying the sample contaminants, using both mass spectra libraries in a non-target analysis, depends on the success of the deconvolution process, i.e. the capability of the software to find the component peaks and to obtain mass spectra as free as possible of sample interferents. Obviously, the more complex the matrix, the more difficult the deconvolution will be.

An efficient approach to overcome the component detection limitations is the use of "post-target" methodology, 34,35 i.e. the selection of the analytes to be searched is done after MS acquisition. A post-target screening facilitates the detection of the compounds as it is only focussed on those pollutants selected. It is unnecessary to totally deconvolute all components present in the samples, these mainly belong to matrix compounds. Furthermore, processing and reviewing steps become easier as fewer compounds are searched for and consequently detected.

In this work three sample types have been selected (wastewater, food and human urine) to explore the potential of UHPLC-(Q)TOF MS to detect and identify/elucidate organic contaminants and/or residues. Two strategies have been applied for this purpose: a post-target screening, based on mass filtering at the exact mass of the compound investigated (typically the (de) protonated molecule) using narrow mass extraction windows and a non-target methodology using both empirical and theoretical mass spectra libraries. QTOF MS has been used under MS^E mode, i.e. simultaneous acquisition at low (LE) and high collision energy (HE) functions, which provides useful information on the (de)protonated molecules (commonly at LE) and on the main fragments ions (commonly in HE). On the basis of this information, and on isotopic distribution observed in the spectra, the reliable identification of the compounds detected in the samples was feasible.

Experimental

Reagents and chemicals

HPLC-grade water was obtained from deionized water passed through a Milli-Q water purification system (Millipore, Bedford, MA, USA). HPLC-grade methanol (MeOH) and acetonitrile (ACN) were purchased from ScharLab (Barcelona, Spain). Formic acid (HCOOH) (>98%) was obtained from Fluka (Buchs, Switzerland). Sodium hydroxide (>99%) was obtained from ScharLab. Leucine enkephalin, used as lock mass, was purchased from Sigma Aldrich (St Louis, MO, USA).

Reference compounds were purchased from Acros Organics (Geel, Belgium), Bayer Hispania (Barcelona, Spain), Fort Dodge Veterinaria (Gerona, Spain), Vetoquinol Industrial (Madrid, Spain), Aventis Pharma (Madrid, Spain), Sigma Aldrich (St Louis, MO, USA), Cerilliant (Round Rock, TX, USA), Dr Ehrenstorfer (Augsburg, Germany), Riedel-de Haën (Seelze, Germany), the National Measurement Institute (Pymble, Australia) and Fluka. All reference materials presented purity higher than 93% (w/w).

Instrumentation

An UPLC Acquity system coupled with a hybrid quadrupole orthogonal acceleration-time-of-flight (Q-oaTOF) mass spectrometer (QTOF Premier, Waters, Milford, MA) provided with an orthogonal Z-spray lockspray electrospray interface (ESI) was used.

Mobile phases A and B were water and methanol respectively, both with 0.01% formic acid. The separation was performed on an Acquity C18 BEH analytical column (150 mm × 2.1 mm, i.d. 1.7 μm) at a flow rate of 0.3 mL min⁻¹ (at 60 °C). The initial percentage of methanol was 10%, which was linearly increased to 90% in 14 min, followed by a 2 min isocratic period and, then, returned to initial conditions during 2 min in total run duration of 18 min. The injection volume was 50 μ L.

Cone and nebulizer gas were nitrogen (Praxair, Valencia, Spain) at flow rates of 60 L h⁻¹ and 600 L h⁻¹, respectively. The nitrogen desolvation temperature was set to 350 °C and the source temperature to 120 °C. A cone voltage of 25 V and capillary voltages of 3.5 kV and 2.5 kV in positive and negative ionisation modes, respectively, were used.

TOF MS resolution was $\sim 10~000$ at full width half maximum (FWHM) in V-mode. MS spectra were acquired over an m/z range 50–1000. Collision gas was argon 99.995% (Praxair, Valencia, Spain), which was always turned on with a pressure of approximately 5×10^{-3} mbar.

Two acquisition functions were created with different collision energies. The first one, the low energy (LE) function, at low collision energy (4 eV) and the second one, the high energy (HE) function, with a collision energy ramp ranging from 15 to 40 eV. The scan time values of LE and HE functions were set to 0.2 and 0.15 s, respectively, both with an inter-scan delay of 0.05 s.

The lock mass (leucine enkephalin, 2 mg L^{-1} in ACN: water, 50: 50) was introduced *via* the lock spray needle at a flow rate of 30 μ L min⁻¹ using a reagent manager pump (Waters). A cone voltage of 60–70 V was selected and checked daily to obtain adequate signal intensity for this compound (around 500 counts).

Calibration of the m/z-axis was performed using the built-in single-syringe pump, directly connected to the interface. Calibration from 50 to 1000 m/z was conducted with a 1 : 1 mixture of 0.05 M NaOH : 5% HCOOH diluted (1 : 25) with water/ACN (20 : 80 v/v) plus imazalil (m/z 297.0561) at a final concentration of 500 μ g L⁻¹.

Data station operating software was MassLynx v 4.1. ChromaLynx XS application manager was used for non-target (deconvolution and library search) as well as for target analysis.

Sample treatment

8 wastewater samples—4 influent (IWW) and 4 effluent (EWW)—10 human urine and 6 food samples (2 oranges, 2 banana and 2 corn samples) were analysed for comparing the screening approaches.

50 mL of wastewater were pre-concentrated by off-line SPE using 200 mg Oasis HLB cartridges, eluted with 5 mL of MeOH, evaporated under a gentle nitrogen stream at 40 $^{\circ}$ C and reconstructed with 1 mL water : MeOH (90 : 10 v/v).

Food sample extraction was performed according to previous work developed by our group. 6,36 20 g of triturated and homogenized orange or banana samples were extracted with 60 mL water: MeOH (20:80 v/v) for 2 min using a high-speed blender, filtered and diluted with water: MeOH (20:80 v/v) to a final volume of 100 mL. Afterwards, an aliquot of the extract was diluted eightfold with water.

2.5 g of crushed corn sample were extracted with 10 mL ACN: water (80: 20 v/v) with 0.1% HCOOH and mechanically shaken for 90 min.⁶ Afterwards, the solution was centrifuged, and a 5 mL aliquot of supernatant was diluted twofold with water.

Human urine samples from healthy volunteers and from people involved in drug detoxification programmes were centrifuged, diluted fivefold with water and directly injected into the LC-QTOF instrument.

Software parameters

The deconvolution and spectra rejection parameters were selected as follows:

- minimum peak width at 5% height: 4 s,
- peak-to-peak baseline noise: 5,

- smoothing activated,
- mass tolerance (mass window width): 20 mDa,
- two mass chromatograms extracted for each component in LE function (5 mass chromatograms for HE), *i.e.* 2 or 5 coeluting ions to be extracted with the narrow window mass selected (+10 mDa).

The values for minimum peak width and mass window were selected as a function of the chromatographic resolution and mass accuracy data of our instrument.

Accurate mass scoring parameters were selected as follows:

- Number of ions used for accurate mass scoring: 2
- Minimum intensity (% of largest peak in the range): 10
- High precision mass tolerance (colouring in green): 2.5 mDa
- Low precision mass tolerance (colouring in yellow) = 5 mDa

These values were selected according to our own experience and characteristics of the LC-QTOF MS equipment used, but they might be modified according to the performance of the instrument used in each laboratory.

Results and discussion

Mass resolving power is an important issue for the correct detection and identification of the suspect compounds. Even if the 10 000 at 10% valley resolution (20 000 FWHM) required by the EC Decision 2002 (ref. 37) is not achieved by the (O)TOF mass spectrometer used, we consider that mass accuracy is really the key in the identification of the compounds in the wide-scope screening. Although strongly correlated, mass resolving power and mass accuracy are not strictly the same. In a previous work, 33 improving the resolution (about 18 000 FWHM) by doubling the path length using the so-called W-mode in different matrices (influent and effluent wastewater, surface water, pepper and cucumber) showed no significant effect for the compounds tested on mass accuracy achieved using UHPLC separation. On the other hand, a 20 mDa mass window has been used in this work for both, non-target and post-target strategies, as a compromise between ensuring correct chromatographic peak at both ends and attainable selectivity. Lower mass windows (e.g. 5 and 10 mDa) were also tested, but finally discarded as no satisfactory chromatographic peaks were always ensured. Newer instruments with stable mass accuracy across the peak could facilitate the screening process by reducing this mass window, even down to 1 mDa.

The non-target and post-target strategies studied in this work were applied to all selected samples to test the screening capabilities and for comparison purposes. A flowchart of the process is shown in Fig. 1.

1. Non-target screening

A true non-target screening using LC-(Q)TOF MS is a challenging task as it is very difficult to detect and identify trace level contaminants when no selection is made on the compounds to be searched.³⁸ In this work, non-target screening was applied to environmental, food and biological samples to evaluate the potential of the algorithm to detect components when dealing with complex matrices. For this purpose, the deconvolution software ChromaLynx XS in a non-target mode was used. The software applies a component detection algorithm (CODA) to

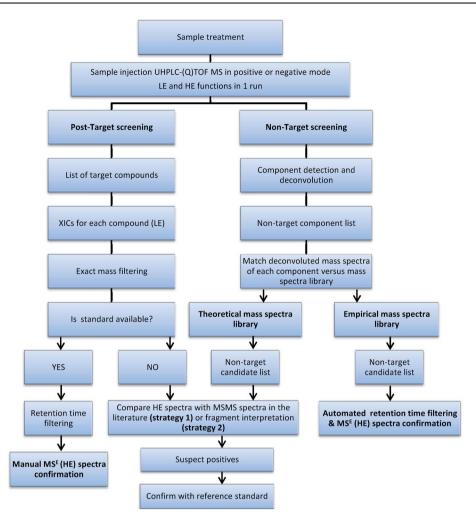


Fig. 1 Flowchart of the overall screening process.

deconvolute the TIC and detect the components present in the sample. Afterwards, it compares the spectra assigned to every component with those included in the home-made libraries. To facilitate the confirmation of the identity of the components detected, two functions were simultaneously acquired at different collision energies (MS^E). The LE function was used to obtain the (de)protonated molecules (occasionally adducts and fragment ions). The HE function was used to promote fragmentation, improving the identification of the positive findings as spectra obtained were quite similar to those of MS/MS experiments. 33,39 This acquisition provides reproducible spectra without the need of precursor ion pre-selection in the first quadrupole. The success for detecting and identifying non-target compounds using this approach obviously depends on the deconvolution process. In addition, MS^E provides not only fragmentation spectra but also isotopic pattern information of the fragments and it conserves adduct and/or dimer information. However, two main limitations were noticed when MS^E was applied to non-target screening:

(a) As there is no pre-selection of precursor ion in the quadrupole, the MS^E approach is less specific and might be conflictive when dealing with non-selective fragments in the presence of co-eluting related compounds. This occurs, for example, when

investigating amphetamine-like compounds amphetamine and methamphetamine. As can be seen in Fig. 2, both drugs elute at very close retention times and present poor and identical HE spectra, with the most abundant ion being the non-selective fragment at m/z 91 corresponding to tropylium ion. Moreover, as protonated molecules have relatively poor abundance in the LE function (especially amphetamine) it could be very difficult to distinguish both compounds at low concentration levels.

(b) The success of the MS^E approach can be limited by the quality of the spectrum.³⁹ Thus, low sensitivity or strongly interfered spectra end up making it unfeasible to match with library spectra as well as not being able to elucidate the component using fragment interpretation. In these cases, additional MS/MS experiments would be helpful in the identification/ elucidation process.

Finally, the software returns a match factor for the comparison of standard and candidate mass spectra and gives the mass errors for the 2 most abundant ions present in the LE function and for the 5 main fragment ions present in the HE function. A positive match can be filtered by a minimum match factor and retention time, if available. In this work, a relatively low match factor (in reverse fit) of 70% was selected as a compromise. This facilitated the reviewing process of positives without losing

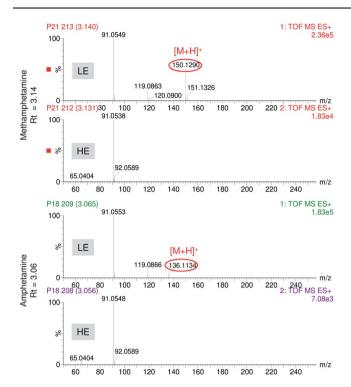


Fig. 2 LE and HE mass spectra for amphetamine-like compounds. LE spectra for methamphetamine and amphetamine show notable in-source fragmentation with different $[M + H]^+$ ion $(m/z \ 150.1290 \ and \ 136.1134)$. Identical HE spectra $(m/z \ 91.0548 \ corresponding to tropylium ion) are obtained for both compounds.$

potential hazardous compounds that could be present in the samples although with low match factors. Two types of mass spectra libraries were evaluated in this work as discussed in the following sections.

When no or unsatisfactory match is obtained, the components appear to be tentative. In these cases, the elucidation of the compound requires a lot of time and effort with a low possibility of success. Furthermore, the majority of non-matched components are likely to be matrix compounds.

1.1 Theoretical library. Initially, a database containing approximately one thousand pollutants of different families (pesticides, antibiotics, pharmaceuticals, illicit drugs, mycotoxins, anabolic steroids, personal care products and metabolites) was built (see ESI†). The compounds were included based on our own experience on LC-MS/MS analysis of environmental and food samples, and on bibliographic data on LC-MS amenable organic pollutants. The database was created separating positive and negative ionisable compounds. It contained information on the molecular formula (required by the software), exact mass of the neutral and the (de)protonated molecule, as well as supplementary information[†] of the compound type and on retention time, when available. From the molecular formulae of each compound, two theoretical mass spectra libraries (for positive and negative ionisation modes) were automatically built, containing theoretical nominal mass spectra of the (de)protonated molecule and sodium adducts as well as the theoretical isotopic pattern expected for each compound. Each library

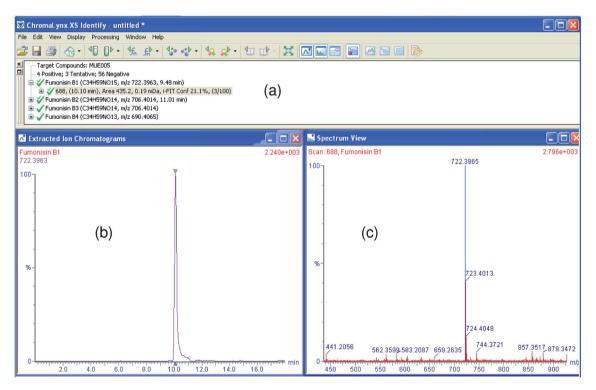


Fig. 3 ChromaLynx XS browser with accurate mass confirmation for Fumonisin B1 in corn using post-target screening. (a) Candidate list for compound with mass error <2.5 mDa (which offers retention time, area, mass error and i-FIT information), (b) nw-XIC for suspected candidate (at 20 mDa window). (c) mass spectrum (in blue, candidate peak is shown).

Table 1 Results of different screening approaches for representative samples. Common data for all approaches are shown: retention time (RT), retention time deviation (only when available) and mass error. For target screening, confirmation with MS^E was performed when reference standard was previously injected. Detection of sodium adduct and/or fragment ions in the LE function is reported. Reverse Fit is given for compounds detected by non-target approaches

				Target screening					Non-target screening		
Effluent wastewater				Tentative identification		_	LE function		Theoretical library	Empirical	
Compound	RT (min)	ΔRT (%)	Δmass/ mDa	Strategy 1 ^a	Strategy 2 ^b	Confirmed with standard?	Na adduct?	Fragment ions?	Match reverse fit	library Match reverse fi	
Antipyrine	5.43	_	1.4			_					
Bamethan	3.43	_	0.5	✓	×	_					
Bisoprolol	7.25	_	0.0	✓	✓	_			929		
Caffeine	4.19	_	0.2	✓	✓	_		***			
Carbendazim	4.45	0.04	0.5		,	Yes		√ (1)	923	898	
Celiprolol	6.41	— 0.10	0.7	✓	/				0.5.5		
Clarithromycin	10.11	0.10	1.7	,	.,	Yes —			855		
Clofibric acid Codeine	12.60 2.82	0.00	2.5 1.6	✓	×	Yes			943	794	
Diazinon	13.04	0.00	0.9			Yes			943	/94	
Diuron	9.98	0.01	0.9			Yes					
Erithromycin	9.98	0.02	1.2			Yes					
(-H ₂ O)	7.00	0.01	1.4			1 68					
Gabapentin	3.47	0.09	0.5	✓	✓	Yes		√ (2)	903		
Irbesartan	11.42	0.10	0.0	/	✓	Yes		. ,	953		
Ketoprofen	10.61	0.04	0.5			Yes	✓				
Metoprolol	5.55	_	0.0	✓	✓	_					
Nordiazepam	11.01	_	0.1		✓	_					
OD-PABA	5.49	_	0.9	✓	✓	_			953		
Oxazepam	10.19	_	0.4		✓	_					
Oxprenolol	2.77	_	0.2	✓	✓	_					
Propylphenazone	9.29		0.4	1	✓	_					
Terbutryn	11.69	0.14	1.0			Yes					
Thiabendazol	5.22	0.12	0.1			Yes			000	025	
Trimethoprim	3.83	0.10	0.9	,	,	Yes	,		980 849	925	
Valsartan Venlafaxine	11.59 7.17	0.05 0.46	1.9 0.3	✓	✓	Yes Yes	✓	√ (1)	049		
MDMA	3.78	0.40	0.3			Yes		√ (1) √ (2)			
Bezafibrate*	11.07	0.00	2.1			Yes	/	V (2)			
Gemfibrozil*	13.84	0.01	2.5			Yes	/				
*Compounds found				n		103	•				
				Target screening					Non-target screening		
_			Tentative identification		_	LE function		Theoretical	Empirica		
	RT	ΔRT	Δmass/			Confirmed with	Na	Fragment	library Match	library Match	
Compound	(min)	(%)	mDa	Strategy 1 ^a	Strategy 2 ^b	standard?	adduct?	ions?	reverse fit	reverse fit	
Imazalil	9.22	0.03	0.3			Yes			916	925	
Thiabendazol	5.21	0.03	0.3			Yes			893	916	
				Target screening					Non-target screening		
Banana peel		Tentative			LE Comption		TT1	Emminia-1			
Банана ресі				identification		_	LE function		Theoretical library	Empirical library	
	RT	ΔRT	Amase/	Strategy		Confirmed with	Na	Fragment	Match	Match	
Compound	(min)	(%)	mDa	1 ^{ab}	Strategy 2 ^b	standard?	adduct?	ions?	reverse fit	reverse fit	
Chlorpyrifos	14.61	0.01	0.5			Yes	✓				
Diazinon	13.03	0.01	0.3			Yes	•		854	863	
Imazalil	9.21	0.01	0.7			Yes			931	923	
	7.41	0.02	U.,								

Table 1 (Contd.)

			Target scr	eening				Non-target screening		
Corn				ion		LE function		Theoretical	Empirical	
RT (min)	ΔRT (%)	Δmass/ mDa	Strategy 1 ^{ab}	Strategy 2 ^b	Confirmed with standard?	Na adduct?	Fragment ions?	Match reverse fit	library Match reverse fit	
10.10 11.69 10.97 12.44	0.03 0.03 —	0.8 1.2 1.3 0.2	<i>y y</i>	<i>y</i>	Yes Yes —			785 681	800 686	
Target screening						Non-target screening				
Urine sample			Tentative identification			LE function		Theoretical	Empirical	
RT (min)	ΔRT (%)	Δmass/ mDa	Strategy 1 ^{ab}	Strategy 2 ^b	Confirmed with standard?	Na adduct?	Fragment ions?	Match reverse fit	library Match reverse fit	
3.27	0.05	0.8			Yes		(2)	893		
2.72	_	1.1	✓	✓	_		. ,			
2.68	0.02	0.3			Yes					
5.72	0.10	0.0			Yes					
3.27							(2)	874		
5.04								895	967	
								F1.6		
	0.19 0.00	0.8 0.6						716		
					Yes					
5.28 5.71	U.00 —	0.0	/	,	1 03					
1 1 1 1 1 R	min) 0.10 1.69 0.97 2.44 RT min) 27 7.72 6.68 7.72 0.04 2.25 1.16	min) (%) 0.10 0.03 1.69 0.03 0.97 — 2.44 — RT ART min) (%) .27 0.05 .72 — .68 0.02 .72 0.10 .27 0.20 .04 0.02 .25 0.11 .16 0.19	min) (%) mDa 0.10 0.03 0.8 1.69 0.03 1.2 0.97 — 1.3 2.44 — 0.2 RT ΔRT Δmass/min) (%) mDa 2.27 0.05 0.8 .72 — 1.1 .68 0.02 0.3 .72 0.10 0.0 .27 0.20 0.4 .04 0.02 1.0 .25 0.11 0.4 .16 0.19 0.8	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	

^a Strategy 1, used for tentative identification of the compounds when the standard was not available, consisted of comparing the main fragments observed in the HE function with common MS/MS product ions reported in the literature. ^b Strategy 2 was made by justifying the HE accurate mass fragments using a bond-disconnecting software.

(positive and negative modes) was used in the corresponding acquisition mode.

A drawback of the theoretical library (and also of the empirical mass spectra library) is that TOF MS spectra are stored in nominal mass for NIST format compatibility, and in this step the mass accuracy information given by TOF MS is lost. In order to minimize this limitation, the mass errors between the measured masses of the compound detected and the exact masses of the candidates formulae are calculated and used in a subsequent step, to rank them and to propose the most plausible identity (accurate mass scoring).³⁵

1.2 Empirical library. Details about reference standards injected and conditions for the creation of the empirical spectra library are reported in Díaz *et al.*³³ Briefly, around 230 reference standards of selected contaminants were injected in both, positive and negative, ionisation modes at low and high collision energy (MS^E mode) to obtain retention time and fragmentation information under the previously optimized conditions.³³ For each compound, two library entries (LE and HE spectra) were created including name, exact mass, retention time and spectra.

Detection/identification problems derived from LE adducts formation and/or important in-source fragmentation were prevented by analyzing the samples under exactly the same conditions as the reference standards. This favoured the task and minimized the risk of potential false negatives. Furthermore, HE mass spectra were automatically matched with those included in the empirical library which greatly facilitated the confirmation of

the compound identity. In our experience, HE provided highly reproducible spectra (independently of the type of sample analysed) when the component was found at relatively high abundance. As signal intensity is the main limitation during the component detection step, HE spectra facilitated identification of the compound when its spectrum was available in the library in those components detected by the non-target approach.

2. Post-target screening

Trying to avoid the dependence of the screening success on the component detection algorithm, a post-target screening strategy was applied including an extraordinarily large number of compounds in the search. The term "post-target" was first used by our group^{34,35,40} as a target screening without pre-selection of the analytes before analysis. It consists of searching for a list of target compounds after MS full-acquisition. Other authors name this approach, when reference standards are unavailable, as suspect screening.20 In the post-target screening, a database with the same compounds included in the theoretical library of the non-target approach was used (ESI†). ChromaLynx XS uses the molecular formula to calculate the exact mass for [M + H]+. Then, the software automatically performs the extraction of a nw-XIC (20 mDa) for each compound in the LE and HE functions and looks for peaks (S/N and peak width higher than pre-selected values) in the corresponding chromatogram. A list of potential candidates found in the sample is shown in different colours depending on accurate mass measurement; positive

Table 2 Illustrative example of the database used in the post-target approach. Database entries were created including molecular formula, retention time (when available), accurate mass and pollutant family, as well as bibliographic source when data were not empirically obtained by reference standard injection. Different entries were created for sodium adducts (marked as Na) and in-source fragments (marked as F1, F2, etc.). Entries marked with (*) indicate the main ion/s in the reference standard spectrum

Compound	Molecular formula	Rt (min)	Ion type	Accurate mass	Pollutant family	Source
Oxytetracycline*	$C_{22}H_{24}N_2O_9$	4.83	$[M + H]^{+}$	461.1560	Antibiotic	
Oxytetracycline F1	$C_{22}H_{22}N_2O_8$	4.83	Fragment ion	443.1454	Antibiotic	
Oxytetracycline F2	$C_{22}H_{19}NO_8$	4.83	Fragment ion	426.1189	Antibiotic	
Amphetamine	$C_9H_{13}N$	3.06	$[M + H]^{+}$	136.1126	Illicit drug	
Amphetamine F1	C_9H_{10}	3.06	Fragment ion	119.0861	Illicit drug	
Amphetamine F2*	C_7H_6	3.06	Fragment ion	91.0548	Illicit drug	
MDMA	$C_{11}H_{15}NO_2$	3.14	$[M + H]^{+}$	194.1181	Illicit drug	
MDMA F1*	$C_{10}H_{10}O_2$	3.14	Fragment ion	163.0759	Illicit drug	
MDMA F2	$C_8H_6O_2$	3.14	Fragment ion	135.0446	Illicit drug	
6-OH-4-Cl-dehydromethyltestosterone	$C_{20}H_{27}O_3Cl$	9.91	$[M + H]^{+}$	351.1727	Steroid	
6-OH-4-Cl-dehydromethyltestosterone (–H ₂ O)	$C_{20}H_{25}O_2Cl$	9.91	Fragment ion	333.1621	Steroid	
6 -OH-4-Cl-dehydromethyltestosterone $(-2 \times H_2O)$	$C_{20}H_{23}OCl$	9.91	Fragment ion	315.1515	Steroid	
6-OH-4-Cl-dehydromethyltestosterone (Na)*	$C_{20}H_{26}NaO_3Cl$	9.91	$[M + Na]^+$	373.1547	Steroid	
Ethisterone	$C_{21}H_{28}O_2$	_	$[M + H]^{+}$	313.2168	Steroid	JMS,42,2007,497-516
Ethisterone [M + Na + MeOH] ⁺	$C_{22}H_{31}NaO_3$	_	$[M + Na + MeOH]^+$		Steroid	JMS,42,2007,497-516
Fumonisin B1*	$C_{34}H_{59}NO_{15}$	10.13	[M + H] ⁺	722.3963	Mycotoxin	
Fumonisin B2*	$C_{34}H_{59}NO_{14}$	11.66	[M + H] ⁺	706.4014	Mycotoxin	
Aldicarb sulfoxide	$C_7H_{14}N_2O_3S$	3.19	M + Hj+	207.0803	Pesticide	
Aldicarb sulfoxide (Na)*	$C_7H_{13}NaN_2O_3S$	3.19	$[M + Na]^+$	229.0623	Pesticide	
Aldicarb sulfoxide F1*	C_4H_8S	3.19	Fragment ion	89.0351	Pesticide	
Aldicarb sulfoxide F2	C ₅ H ₉ NOS	3.19	Fragment ion	132.0483	Pesticide	
Tebufenozide	$C_{22}H_{28}N_2O_2$	12.54	$[M + H]^+$	353.2229	Pesticide	
Tebufenozide (Na)*	$C_{22}H_{27}N_2O_2Na$	12.54	[M + Na]+	375.2048	Pesticide	
Tebufenozide $(2M + Na)^*$	$C_{44}H_{55}N_4O_4Na$	12.54	[2M + Na] ⁺	727.4199	Pesticide	
Tebufenozide F1*	$C_{18}H_{20}N_2O_2$	12.54	Fragment ion	297.1603	Pesticide	
Tebufenozide F2*	C ₉ H ₈ O	12.54	Fragment ion	133.0653	Pesticide	
Azinphos-methyl	$C_{10}H_{12}N_3O_3PS_2$	10.49	$[M + H]^{+}$	318.0136	Pesticide	
Azinphos-methyl (Na)*	$C_{10}H_{11}N_3O_3PS_2Na$		$[M + Na]^+$	339.9956	Pesticide	
Azinphos-methyl F1	$C_8H_5N_3O$	10.49	Fragment ion	160.0511	Pesticide	
Azinphos-methyl F2*	C_8H_5NO	10.49	Fragment ion	132.0449	Pesticide	
Azoxystrobin	$C_{22}H_{17}N_3O_5$	10.97	[M + H] ⁺	404.1246	Pesticide	
Azoxystrobin (Na)*	$C_{22}H_{16}NaN_3O_5$	10.97	$[M + Na]^+$	426.1066	Pesticide	
Azoxystrobin F1*	$C_{21}H_{13}N_3O_4$	10.97	Fragment ion	372.0984	Pesticide	
Bifenazate	$C_{17}H_{20}N_2O_3$	11.92	[M + H] ⁺	1.0078	Pesticide	
Bifenazate (Na)*	$C_{17}H_{19}N_2O_3Na$	11.92	$[M + Na]^+$	22.9898	Pesticide	
Bifenazate F1	$C_{13}H_{11}NO$	11.92	Fragment ion	198.0919	Pesticide	
Bifenazate F2	$C_{12}H_{11}N$	11.92	Fragment ion	170.0970	Pesticide	
Dimethoate	$C_5H_{12}NO_3PS_2$	5.76	[M + H] ⁺	230.0075	Pesticide	
Dimethoate (Na)*	$C_5H_{11}NaNO_3PS_2$	5.76	$[M + Na]^+$	251.9895	Pesticide	
Methiocarb sulfone	$C_{11}H_{15}NO_4S$	6.27	$[M + H]^+$	258.0800	Pesticide	
Methiocarb sulfone (Na)*	$C_{11}H_{14}NO_4SNa$	6.27	$[M + Na]^+$	280.0620	Pesticide	
Methiocarb sulfone F1	$C_9H_{12}O_3S$	6.27	Fragment ion	201.0585	Pesticide	
Methiocarb sulfone F2*	C ₈ H ₉ O	6.27	Fragment ion	122.0732	Pesticide	
Methiocarb sulfoxide	$C_{11}H_{15}NO_3S$	5.73	[M + H] ⁺	242.0851	Pesticide	
Methiocarb sulfoxide (Na)	$C_{11}H_{14}NO_3SNa$	5.73	$[M + Na]^+$	264.0671	Pesticide	
Methiocarb sulfoxide (14a)	$C_9H_{12}O_2S$	5.73	Fragment ion	185.0636	Pesticide	
Thiamethoxam	$C_8H_{10}CIN_5O_3S$	4.26	$[M + H]^+$	292.0271	Pesticide	
Thiamethoxam (Na)*	C ₈ H ₉ ClN ₅ O ₃ SNa	4.26	$[M + Na]^+$	314.0091	Pesticide	
Thiamethoxam F1*	$C_8H_{10}N_4OS$	4.26	Fragment ion	211.0654	Pesticide	
Thiamethoxam F2	C_4H_2NSC1	4.26	Fragment ion	131.9675	Pesticide	
Thiobencarb	C ₁₂ H ₁₆ ClNOS	13.39	[M + H] ⁺	258.0719	Pesticide	
Thiobencarb (Na)*	$C_{12}H_{15}NaClNOS$	13.39	$[M + Na]^+$	280.0539	Pesticide	
Thiobencarb F1*	C ₁₂ H ₁₅ H ₄ CHVOS C ₇ H ₅ Cl	13.39	Fragment ion	125.0158	Pesticide	
Thiodicarb	$C_{10}H_{18}N_4O_4S_3$	9.36	[M + H] ⁺	355.0568	Pesticide	
Thiodicarb (Na)*	$C_{10}H_{18}N_4O_4S_3$ $C_{10}H_{17}NaN_4O_4S_3$	9.36	$[M + Na]^+$	377.0388	Pesticide	
Thiodicarb (Na). Thiodicarb F1	$C_{10}H_{17}NaN_4O_4S_3$ C_3H_5NS	9.36	Fragment ion	88.0221	Pesticide	
Bezafibrate	$C_{19}H_{20}CINO_4$	11.06	[M + H] ⁺	362.1159	Pharmaceutical	
Bezafibrate (Na)*	$C_{19}H_{20}CINO_4$ $C_{19}H_{19}NaClNO_4$	11.06	$[M + Na]^+$	384.0979	Pharmaceutical	
Chloramphenicol				323.0201		
	$C_{11}H_{12}Cl_2N_2O_5$	6.46	$[M + H]^{+}$ $[M + N_0]^{+}$		Pharmaceutical	
Chloramphenical Class	$C_{11}H_{11}NaCl_2N_2O_5$		[M + Na] ⁺	345.0021	Pharmaceutical	
Chloramphenicol F1*	$C_{11}H_{10}N_2O_4Cl_2$	6.46	Fragment ion	305.0101	Pharmaceutical	
Chloramphenical F2*	C ₁₀ H ₈ N ₂ O ₃ Cl ₂	6.46	Fragment ion	275.0002	Pharmaceutical	
Chloramphenicol F3	C ₁₁ H ₈ NOCl ₂	6.46	Fragment ion	241.0078	Pharmaceutical	
Gemfibrozil (Na)*	$C_{15}H_{22}O_3 C_{15}H_{21}NaO_3$	13.81	$[M + H]^{+}$ $[M + Na]^{+}$	251.1647	Pharmaceutical Pharmaceutical	
		13.81	LDA + NOIT	273.1467	Pharmacautical	

(green) for error <2.5 mDa, tentative (yellow) for error between 2.5 and 10 mDa, and negative (red) for error >10 mDa. Furthermore, as in the non-target approach, ChromLynx XS filters positive findings according to retention time deviation limit when this information is available in the database (reference standards previously injected). The retention time window was set in ± 0.5 min but accepted tolerance was 2.5%. Thus, the retention times for 231 analytes, injected when building the empirical library, were also introduced in the database. In this way, nw-XIC, top peak spectra and mass error as well as isotopic distribution fit (i-FIT) information, retention time (measured and expected when already known) and peak area were available for positive matches. Fig. 3 shows the ChromaLynx XS browser for a positive of mycotoxin Fumonisin B1 in a corn sample using this approach.

QTOF MS post-target screening has proved to be an efficient tool due to the high number of pollutants screened. The potential of this approach to detect different families of organic contaminants, for example drugs of abuse or antibiotics in environmental samples, has been reported recently.^{39,41} The easy reviewing step and the relevant information obtained, such as accurate mass spectrum of the peak, mass error for the protonated molecule and the most abundant fragments, and isotopic distribution, give high confidence to the confirmation of potential positives even without reference standards being available. The large number of contaminants included in the list (more than

1000) opens a new scenario in screening, favouring a more realistic overview when investigating organic contaminants in different applied fields. However, if only the predicted presence of the protonated molecule was taken into account in the LE function, potential in-source fragments would not be detected (e.g. as occurs in amphetamine-like compounds, see Fig. 2), not even sodium or other adducts that could be formed.

In this work, formic acid was added to the mobile phases. Under this situation, ammonia adducts and other adducts like $[M + MeOH + H]^+$ or $[M + K]^+$ would not normally be expected. However, sodium adducts are common for many LC-amenable compounds, and they might be present despite using formic acid. In our own experience, 88 out of 231 compounds (38%) included in the experimental library showed sodium adducts at relative abundance higher than 10%. Among them, 38 compounds (16% of the total compounds) presented the [M + H]⁺ ion at relative intensity lower than 10%, this becoming the [M + Na]⁺, the most abundant ion in the mass spectra. When analyzing real samples, sodium adducts might be found at higher abundance due to the normal presence of sodium in the sample matrices. Therefore, it is important to include sodium adducts in the screening to avoid potential false negatives in those cases where it is the most abundant ion (see Table 1). However, it seems reasonable not to include sodium adducts for all analytes investigated, as the processing and the reviewing step would be much longer and more tedious. The injection of reference standards and/or

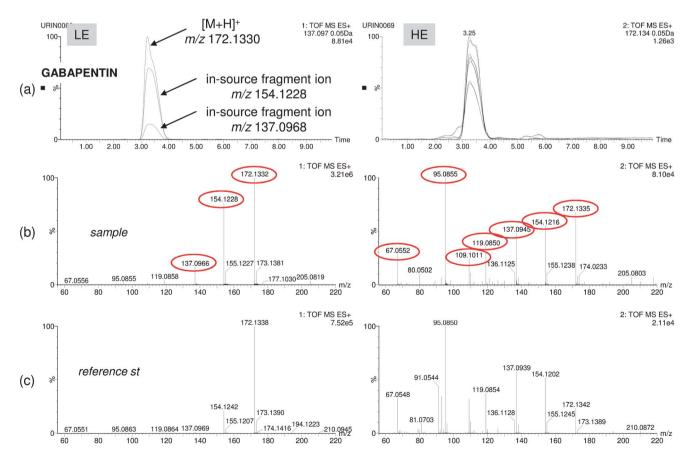


Fig. 4 Positive finding of the pharmaceutical gabapentin in human urine: (a) overlapped nw-XIC for three main ions (protonated ion at *m*/*z* 172 and insource fragments at *m*/*z* 154 and 137) in the LE function and seven coeluting ions in the HE function. LE and HE spectra for sample (b) and reference standard (c) showing good correlation for up to six abundant fragment ions.

literature search, along with analyst experience, are necessary parts of knowing when it is reasonable to include compound adducts to improve the confidence of the screening process.

A similar problem may occur when important in-source fragmentation takes place at the LE function. In this work, we used 25 V cone voltage as better sensitivity was observed for selected analytes in the 20-30 V range.³³ Obviously, this cone voltage is a compromise value as it is not the best choice for all compounds but it is impossible to optimize any variable for all LC-amenable compounds included in the database. As previously stated, amphetamine ([M + H]+ 136.1126) presents an insource fragment at m/z 91.0553 as the most abundant ion in the spectra, while the protonated molecule has an abundance lower than 10% (Fig. 2). Other examples are the insecticide carbaryl (fragment at m/z 145.0563) or pesticide metabolite aldicarb sulfoxide (fragment at m/z 89.0415). In these cases, analyte detection in samples based on testing [M + H]⁺ presence would be only feasible at relatively high analyte concentrations.

Other compounds, like anabolic steroids, are frequently ionised forming adducts with MeOH, acetonitrile, ammonium or sodium (as a function of the mobile phase and sample matrix composition) and/or they suffer in-source fragmentation with neutral losses of one, or even two, water molecules ($[M - H_2O +$ H_1^+ , $[M - 2H_2O + H_1^+)$. ⁴² The later drawback is more difficult to solve than adducts formation, but it could be circumvented by including empirical formula of the known fragment ions in the database. Again, information reported in the literature and/or from reference standards injection would be required to include expected fragments in the database. Although fragmentation behaviour is not completely known in most cases, in our

experience, this effect is less common than adduct formation. Indeed, only 6 out of 231 compounds (3% of the compounds studied) almost exclusively presented the fragment ion as base peak, with the protonated ion being practically absent. In these particular cases, monitoring this fragment is mandatory for compound detection. In-source fragmentation turns into a useful confirmatory tool when the reference standard is available and/ or its behaviour is well known. Thus, including most abundant fragments is always useful for automated confirmation.

As an illustrative example, Table 2 shows information on database entries for different types of analytes included in this work. The molecular formula of the ion, when adduct formation and/or in-source fragmentation occurred, was also introduced in the database, as well as the bibliographic source, when information on possible occurrence of these ions was not directly obtained from reference standard injection.

3. Application to samples

After application of the screening strategies to selected food, wastewater and human urine samples, the post-target approach was found to be the most efficient for wide-scope screening. In all samples analyzed, the number of positives was higher than using the non-target approach, in this way giving a more realistic overview of the presence of organic pollutants in the samples. A summary of the results obtained for selected samples is shown in Table 1. The list of pollutants found by target and non-target screening (those with an adequate match reverse fit) is reported together with the main information managed (mass error and retention time), as well as retention time deviation when

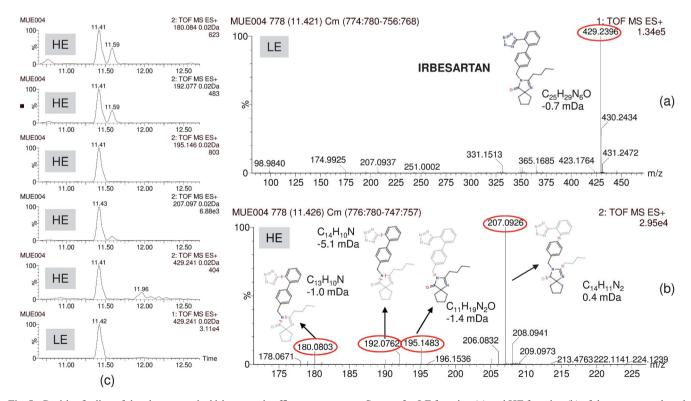


Fig. 5 Positive finding of the pharmaceutical irbesartan in effluent wastewater. Spectra for LE function (a) and HE function (b) of the suspect peak and justification of the HE fragments using MassFragment software. (c) nw-XICs (20 mDa mass window) for [M + H]⁺ in LE function and main fragments in HE function.

reference standard was available. Almost in all cases, Rt deviation was lower than 1%. However, the retention time window for positive match was ± 0.5 min due to the wide range of matrix analysed having, in some particular cases, deviations higher than 2% typically accepted as in the case of Venlafaxin. Confirmation using MS^E is also shown when it could be made. When reference standards were unavailable, information on fragmentation and retention time was absent. Two strategies were followed to improve the confidence in the compound identification.

The first strategy (Strategy 1 in Table 1) was to simply compare main fragments observed in HE acquisition with common MS/MS product ions reported in the literature for the suspect compound. This was the case for the antibiotic gabapentin, which was detected and identified in urine and wastewater by the presence of two abundant fragments in the HE spectrum with *m*/*z* 137.0966 and 154.1232 (Fig. 4). These fragment ions were also present in the LE function and had been reported by other authors for determination of gabapentin by QqQ.⁴³⁻⁴⁵ Elemental composition for these two fragments was calculated based on their accurate masses obtaining errors of 0.7 and 0.2 mDa, respectively.

The second strategy (Strategy 2 in Table 1) consisted of justifying the fragments accurate mass (typically observed in the HE spectra) using MassFragment software. This software applies a bond-disconnecting methodology to obtain possible structures for the fragment ions from a given molecule. An example of this approach is shown in Fig. 5, where identification of main fragments of the pharmaceutical irbesartan was carried out. For this

purpose. LE and HE combined spectrum of suspect irbesartan was extracted from the chromatographic peak (Fig. 5a and b). The main fragments were justified with the MassFragment tool obtaining reliable structures for all of them. In order to avoid spectrum interferences that could complicate the identification process, recognizing which ions are fragments and which are not, becomes mandatory. From this point of view, UHPLC resolution proved to be valuable for choosing perfectly coeluting ions (see Fig. 5c). Irbesartan is an angiostensin II receptor antagonist used in the treatment of hypertension that has been in the market for over 10 years. 46 Some fragments observed for irbesartan had been previously reported by ion trap;46 the most used SRM transition coincides with the most abundant fragment ion of the TOF spectra (m/z 207.0922). 47,48 However, as MassFragment is a bonddisconnecting software, correct justification is not always feasible. Thus, for m/z 192 ion unreliable structures were suggested. In these cases, previous analyst knowledge or better fragmentation prediction software is necessary.

In Table 1, the strategy used for the identification of each suspected positive is shown. Bibliographic search and fragment interpretation were helpful to confirm potential positives. When a disagreement occurred between experimental and literature data for fragment ions (if available), and when structures provided by MassFragment software did not fit with the structure of the candidate, the suspected positive could not be confirmed, and no further research was performed for its elucidation.

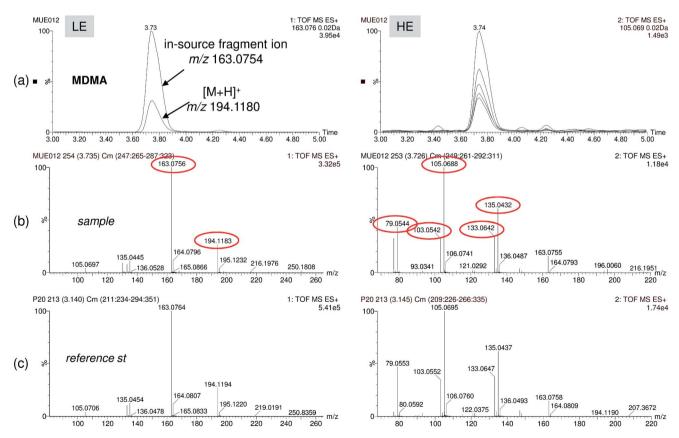


Fig. 6 Positive finding of the drug of abuse MDMA in effluent wastewater. (a) Overlapped nw-XIC for two main ions (protonated ion at m/z 194 and in-source fragment at m/z 163) in the LE function, and up to five coeluting ions in the HE function. LE and HE spectra for sample (b) and reference standard (c) showing good correlation for up to five abundant fragment ions.

Following the above mentioned strategies, high confidence in the identification process can be achieved. However, no definitive confirmation should be made without injecting the reference standard. Thus, for the most frequently detected pharmaceuticals, irbesartan, valsartan and gabapentin, the reference compounds were acquired. After injecting the standard solutions, all suspect positives in wastewater were confirmed. Our experience on identification of suspect organic contaminants by LC-QTOF under MS^E mode is that the great majority of suspect positives (around 95%) were subsequently confirmed when the reference standard was acquired. This means that acquisition of expensive standards could be made only when solid evidence exists on their presence in samples analyzed. The decision on which standards should be acquired would then be made on the basis of previous findings by QTOF MS.

To overcome some post-target limitations and to enhance detectability and identification reliability, improvements in the database approach were made to minimize "missing" compounds due to abundant adduct formation and/or in-source fragmentation. More entries were added in the pollutant database for compounds with a high degree of fragmentation and sodium adduct formation. This is easier when information for the compound is available. After reprocessing the samples using the new, enlarged database, two more compounds (gemfibrozil and bezafibrate) were found in

wastewater. These compounds were not detected before due to the abundant sodium adduct formation in positive electrospray ionisation (marked as * in Table 1). In addition, not only the detection step was improved but also the confidence in the identification, as for several analytes, both the protonated molecule and in-source fragments/sodium adducts were also detected (information shown in Table 1). To exemplify this feature, Fig. 6 shows a positive finding of MDMA in EWW. As can be seen, the protonated ion and main in-source fragment ion of the compound were both detected, the latter being much more abundant than the protonated molecule.

As a summary, two situations could be considered when using the post-target approach based on QTOF measurements:

- (a) Detection of target analytes for which standard is available and has been previously injected under the same conditions as the samples. In this case, retention time, in-source fragmentation and adduct formation became useful tools, making the confirmation of findings highly reliable, surely unequivocal.
- (b) Detection of suspect compounds for which reference standards are unavailable. Obviously, the situation requires extra-work and time. After a careful study of the full-scan accurate mass data obtained for the suspect compound, a reliable identification could be advanced. A definitive confirmation by injection of the reference standard would be required in the case that significant

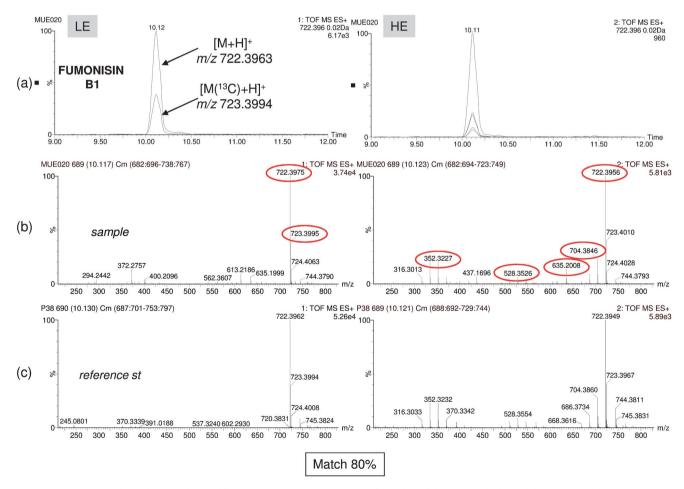


Fig. 7 Non-target screening using experimental library search. Accurate-mass confirmation of the mycotoxin fumonisin B1. (a) Overlapped nw-XICs of the main deconvoluted ions of Fumonisin B1 under LE and HE conditions. Mass spectrum at LE and HE functions for sample (b) and reference standard (c). Library match (80%) and accurate-mass confirmation of the ions (mass errors below 1.3 mDa).

environmental or legal implications were associated to the presence of the suspect compound. Here, the experience of the analyst and their background on mass spectrometry is of the utmost relevance.

Regarding the non-target screening results, it must be noted that the deconvolution process depends to a great extent on the intensity of the chromatographic peak. Using this approach, several contaminants were missed, as the number of compounds found in the samples was considerably lower than using the targeted one (Table 1). Furthermore, non-target screening with empirical library allowed us to detect very few compounds, not only because of the component detection limitations but also due to fewer entries in this library (231). However, confirmation of the identity becomes simultaneous and more reliable than with other approaches (i.e. theoretical library) as LE and HE spectra are compared with those included in the empirical library making unlikely the reporting of false positives. As an example, Fig. 7 shows a corn sample positive to fumonisin. In this figure, two and five coeluting ions were selected for component detection in the LE and HE functions, respectively (Fig. 7a). Both deconvoluted LE and HE mass spectra were automatically compared with those of fumonisin B1 included in the empirical mass spectra library with a match of 80% (Fig. 7b and c).

When employing the theoretical, library-based, non-target screening approach, the investigation of findings when reference standards were unavailable was carried out using the same two strategies discussed before for post-target screening.

This work shows that the post-target approach has better capability for wide-scope screening of different analyte/sample matrix combinations. However, the non-target approach still has some advantages, especially when using an experimental library, as a comparison of the suspect compound *versus* library spectra is automatically performed achieving a highly reliable identification. In addition, other non-expected compounds that might be present in samples at relatively high concentrations might be detected without any kind of selection (pre- or post-target). However, searching for unknowns is an analytical challenge, where the possibilities to elucidate the components detected are rare.³⁸ The main limitation for this approach is the difficulty of having large compound libraries similar to those used in GC-MS. At the moment, spectral libraries for LC-MS are home-made and are quite limited. Hopefully, in the near future large standardized libraries which will facilitate non-target screening will be available.

In this work, several contaminants have been found in the three types of samples investigated. Some of them have been tentatively identified without reference standard. The compounds detected belong to very different chemical classes and included emerging contaminants such as pharmaceuticals, UV filters and drugs of abuse, as well as several pesticides. Commonly used post-harvest fungicides imazalil and thiabendazol were identified in the orange and banana samples. In the case of the corn samples, the mycotoxins Fumonisin B1 and B2 were found, and also the less commonly detected Fumonisin B3 and B4 that were not previously included in the common pretarget approaches applied.

Conclusions

The comparison of different strategies based on the use of UHPLC coupled with QTOF MS for large-scale screening of organic pollutants in food, environmental and urine samples has been carried out. Thanks to the accurate-mass, full-spectrum acquisition in QTOF MS, it is feasible to apply both the target and non-target approaches, which can be seen as complementary within the public health field.

The application of the target approach to selected samples has been demonstrated as an efficient tool for screening a large number of pollutants. For this purpose, a database containing information on the exact mass of the (de)protonated molecule and on the fragment ions and adducts (typically sodium adducts) has been created containing more than 1000 entries. This database has been built on the basis of our own experience and from data reported in the literature on LC-MS analysis of the compounds. Once a compound is detected, the potential positives need to be confirmed taking into account the information obtained on accurate masses of the (de)protonated molecule and of fragment ions, as well as the isotopic distribution. This is feasible using the MS^E acquisition mode in the QTOF instrument, which allows the simultaneous MS data acquisition at low and high collision energy. The accomplishment of retention times and experimental MS^E fragmentation using reference standards obviously facilitates the confirmation step.

In this work, an empirical library containing 231 selected compounds has also been employed in both the target and non-target approaches. Building empirical spectral libraries has been found to be the best way to facilitate both screening types, although it requires the injection of a large number of reference standards to be efficiently applied.

The non-targeted screening presents important drawbacks at low compound concentrations, especially in more complex-matrix samples, due to the difficulties in the components detection step. Identification of non-target contaminants is greatly facilitated when the compound detected is included in the homemade libraries, otherwise the elucidation of the compound becomes an analytical challenge where the possibilities of success are rare.

An interesting advantage associated with TOF MS-based methodologies concerns the possibility of performing retrospective analysis. This allows investigation of the presence of organic contaminants that were included in the first screening. This can be done at any time, without the need of either new analysis or new sample injections.

Acknowledgements

This work has been developed with financial support from the Ministry of Education and Science, Spain (CTQ 2009-12347). R. Diaz is very grateful to Conselleria d'Educació (Generalitat Valenciana) for his pre-doctoral grant. The authors are grateful to Serveis Centrals d'Instrumentació Científica (SCIC) of University Jaume I for the use of UPLC-QTOF-MS (QTOF Premier) and to Generalitat Valenciana for the financial support (Research Group of Excellence, Prometeo/2009/054).

References

- F. Hernández, J. V. Sancho, M. Ibáñez and C. Guerrero, TrAC, Trends Anal. Chem., 2007, 26, 466.
- 2 A. R. Fernández-Alba and J. F. García-Reyes, TrAC, Trends Anal. Chem., 2008, 27, 973.

- 3 Y. Picó, C. Blasco and G. Font, Mass Spectrom. Rev., 2004, 23,
- 4 R. Rodil, J. B. Quintana, P. López-Manía, S. Muniategui-Lorenzo and D. Prada-Rodríguez, Anal. Chem., 2008, 80, 1307.
- 5 M. Petrovic, M. Gros and D. Barcelo, J. Chromatogr., A, 2006, 1124,
- 6 E. Beltrán, M. Ibáñez, J. V. Sancho and F. Hernández, Rapid Commun. Mass Spectrom., 2009, 23, 1801.
- 7 F. Hernández, J. V. Sancho, M. Ibáñez and S. Grimalt, TrAC, Trends Anal. Chem., 2008, 27, 862.
- 8 A. Kaufmann, P. Butcher, K. Maden and M. Widmer, J. Chromatogr., A, 2008, 1194, 66.
- 9 A. Polettini, R. Gottardo, J. P. Pascali and F. Tagliaro, Anal. Chem., 2008, 80, 3050.
- 10 I. Ferrer and E. M. Thurman, J. Chromatogr., A, 2007, 1175, 24.
- 11 A. Kaufmann, P. Butcher, K. Maden and M. Widmer, Anal. Chim. Acta. 2007. 586, 13.
- 12 M. J. Martínez Bueno, A. Agüera, M. J. Gómez, M. D. Hernando, J. F. García-Reyes and A. R. Fernández-Alba, Anal. Chem., 2007, **79**, 9372
- 13 C. A. Mueller, W. Weinmann, S. Dresen, A. Schreiber and M. Gergov, Rapid Commun. Mass Spectrom., 2005, 19, 1332.
- 14 A. C. Hogenboom, J. A. van Leerdam and P. de Voogt, J. Chromatogr., A, 2009, 1216, 510.
- 15 J. M. Marín, E. Gracia-Lor, J. V. Sancho, F. J. López and F. Hernández, J. Chromatogr., A, 2009, 1216, 1410.
- 16 F. Hernández, O. J. Pozo, J. V. Sancho, L. Bijlsma, M. Barreda and E. Pitarch, J. Chromatogr., A, 2006, 1109, 242.
- 17 B. Kmellár, P. Fodor, L. Pareja, C. Ferrer, M. A. Martínez-Uroz, A. Valverde and A. R. Fernandez-Alba, J. Chromatogr., A, 2008,
- 18 S. Inoue, T. Saito, H. Mase, Y. Suzuki, K. Takazawa, I. Yamamoto and S. Inokuchi, J. Pharm. Biomed. Anal., 2007, 44, 258.
- L. Bijlsma, J. V. Sancho, E. Pitarch, M. Ibáñez and F. Hernández, J. Chromatogr., A, 2009, 1216, 3078.
- 20 M. Krauss, H. Singer and J. Hollender, Anal. Bioanal. Chem., 2010, **397** 943
- T. Pihlström, G. Blomkvist, P. Friman, U. Pagard and B. Österdahl, Anal. Bioanal. Chem., 2007, 389, 1773.
- 22 C. Lesueur, P. Knittl, M. Gartner, A. Mentler and M. Fuerhacker, Food Control, 2008, 19, 906.
- 23 H. G. J. Mol, P. Plaza-Bolaños, P. Zomer, T. C. De Rijk, A. A. M. Stolker and P. P. J. Mulder, Anal. Chem., 2008, 80, 9450.
- 24 I. Ferrer, A. Fernandez-Alba, J. A. Zweigenbaum and E. M. Thurman, Rapid Commun. Mass Spectrom., 2006, 20, 3659.
- 25 Z. Herrera Rivera, E. Oosterink, L. Rietveld, F. Schoutsen and L. Stolker, Anal. Chim. Acta, 2011, 70, 114.

- 26 M. Kellmann, H. Muenster, P. Zomer and H. Mol, J. Am. Soc. Mass Spectrom., 2009, 20, 1464.
- I. Bobeldijk, J. P. C. Vissers, G. Kearney, H. Major and J. A. Van Leerdam, J. Chromatogr., A, 2001, 929, 63.
- 28 F. Hernández, S. Grimalt, O. J. Pozo and J. V. Sancho, J. Sep. Sci., 2009 32 2245
- 29 F. Hernández, M. Ibáñez, O. J. Pozo and J. V. Sancho, J. Mass Spectrom., 2008, 43, 173.
- S. Grimalt, O. J. Pozo, J. V. Sancho and F. Hernández, *Anal. Chem.*, 2007, 79, 2833.
- 31 T. Portóles, M. Ibáñez, J. V. Sancho, F. J. López and F. Hernández, J. Agric. Food Chem., 2009, 57, 4079.
- 32 P. Marquet, N. Venisse, E. Lacassie and G. Lachâtre, Analysis, 2000, 28, 925.
- 33 R. Díaz, M. Ibáñez, J. V. Sancho and F. Hernández, Rapid Commun. Mass Spectrom., 2011, 25, 355
- 34 F. Hernández, O. J. Pozo, J. V. Sancho, F. J. López, J. M. Marín and M. Ibáñez, TrAC, Trends Anal. Chem., 2005, 24, 596.
- 35 M. Ibáñez, J. V. Sancho, F. Hernández, D. McMillan and R. Rao, TrAC, Trends Anal. Chem., 2008, 27, 481.
- 36 O. J. Pozo, M. Barreda, J. V. Sancho, F. Hernández, J. Lliberia, M. A. Cortés and B. Bagó, Anal. Bioanal. Chem., 2007, 389, 1765.
- 37 European Commission Decision 2002/657/EC, Off. J. Eur. Communities: Legis., 2002, 221, 8.
- 38 M. Ibáñez, J. V. Sancho, O. J. Pozo, W. Niessen and F. Hernández, Rapid Commun. Mass Spectrom., 2005, 19, 169.
- F. Hernández, L. Bijlsma, J. V. Sancho, R. Díaz and M. Ibáñez, Anal. Chim. Acta, 2011, 684, 96.
- 40 T. Portolés, E. Pitarch, F. J. López, J. V. Sancho and F. Hernández, J. Mass Spectrom., 2007, 42, 1175.
- 41 M. Ibáñez, C. Guerrero, J. V. Sancho and F. Hernández, J. Chromatogr., A, 2009, 1216, 2529.
- 42 O. J. Pozo, P. V. Eenoo, K. Deventer and F. T. Delbeke, TrAC, Trends Anal. Chem., 2008, 27, 657.
- 43 R. Oertel, N. Arenz, J. Pietsch and W. Kiroh, J. Sep. Sci., 2009, 32,
- 44 B. Kasprzyk-Hordern, R. M. Dinsdale and A. J. Guwy, J. Chromatogr., A, 2007, 1161, 132
- 45 A. Ojha, R. Rathod, C. Patel and H. Padh, Chromatographia, 2007, **66**, 853.
- 46 R. P. Shah, A. Sahu and S. Singh, J. Pharm. Biomed. Anal., 2010, 51, 1037
- 47 G. N. W. Leung, D. K. K. Leung, T. S. M. Wan and C. H. F. Wong, J. Chromatogr., A, 2007, 1156, 271.
- 48 L. Kristoffersen, E. L. Øiestad, M. S. Opdal, M. Krogh, E. Lundanes and A. S. Christophersen, J. Chromatogr., B: Anal. Technol. Biomed. Life Sci., 2007, 850, 147.