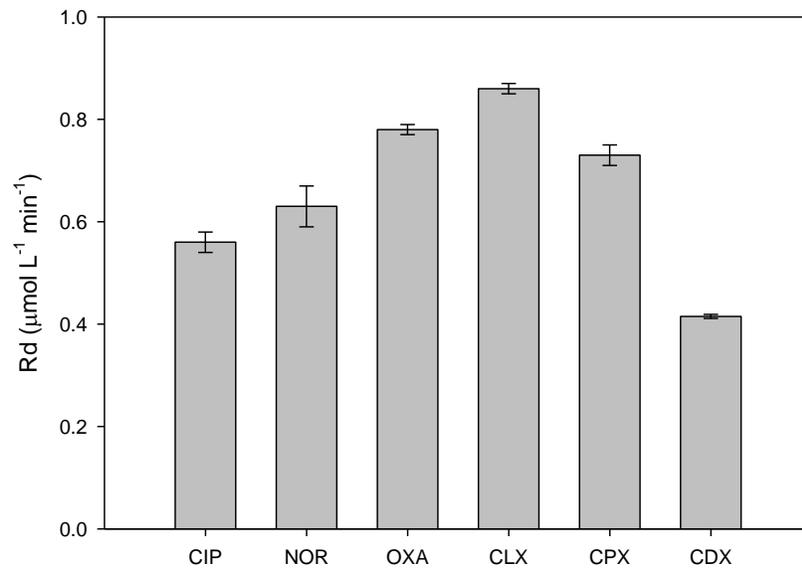
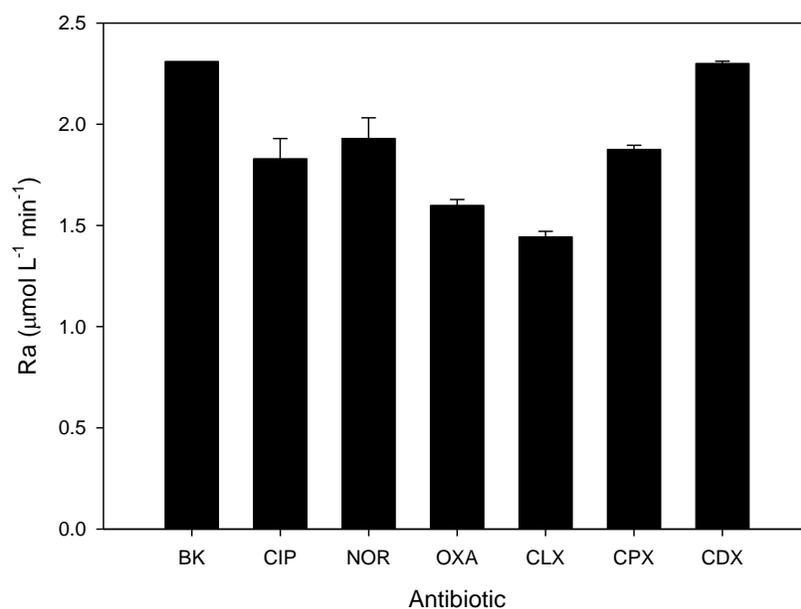


## Figure Captions

A

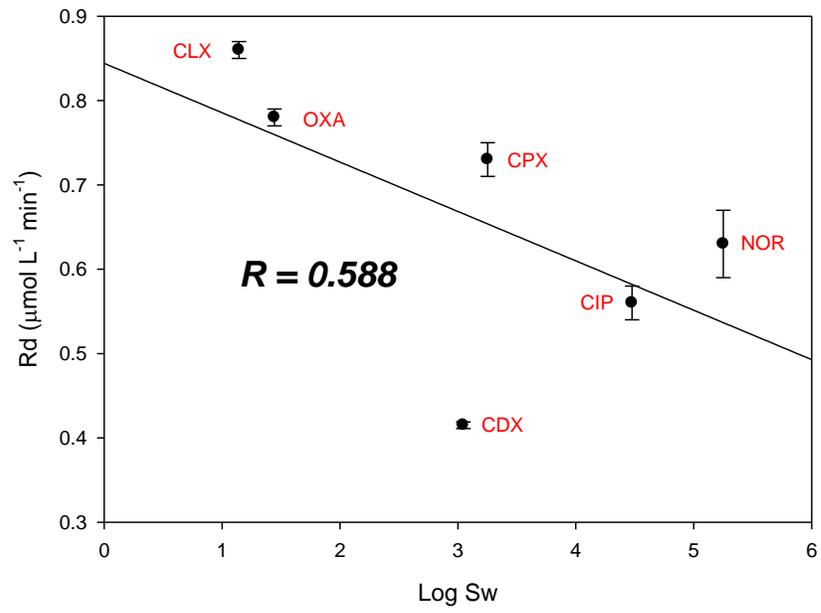


B

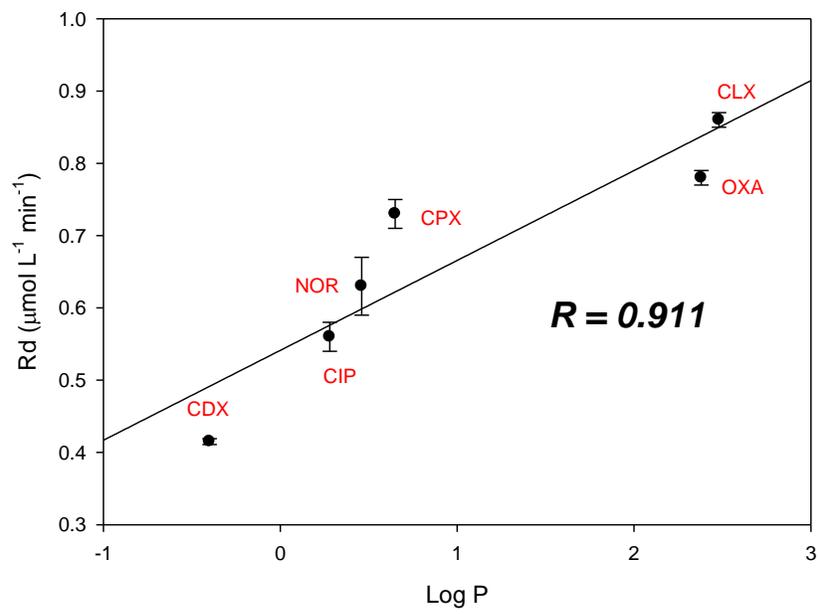


**Figure 1. (A).** Degradation rates ( $R_d$ ) for sonochemical treatment of the antibiotics **(B).** Rates of hydrogen peroxide accumulation ( $R_a$ ) during the degradation process. BK: blank (i.e., distilled water without antibiotics). Conditions: Power density:  $88 \text{ W L}^{-1}$ ,  $f$ : 375 kHz,  $V$ : 300 mL,  $[\text{Antibiotics}]$ :  $40 \mu\text{M}$ ,  $\text{pH}_{\text{initial}}$ : 6.5.

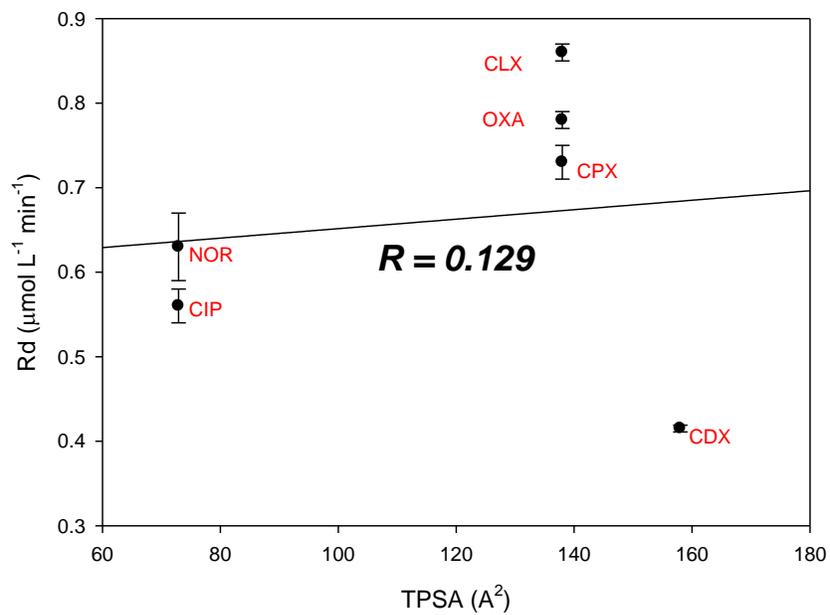
A



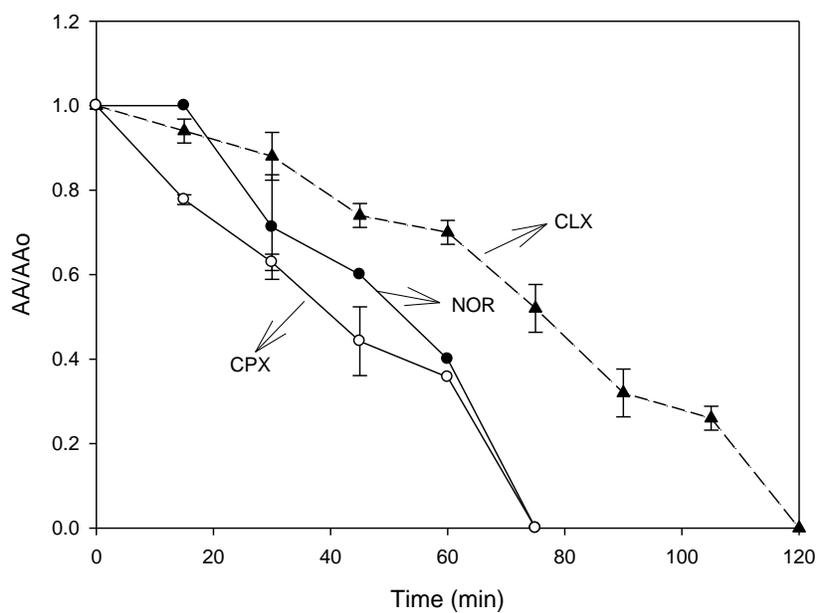
B



C

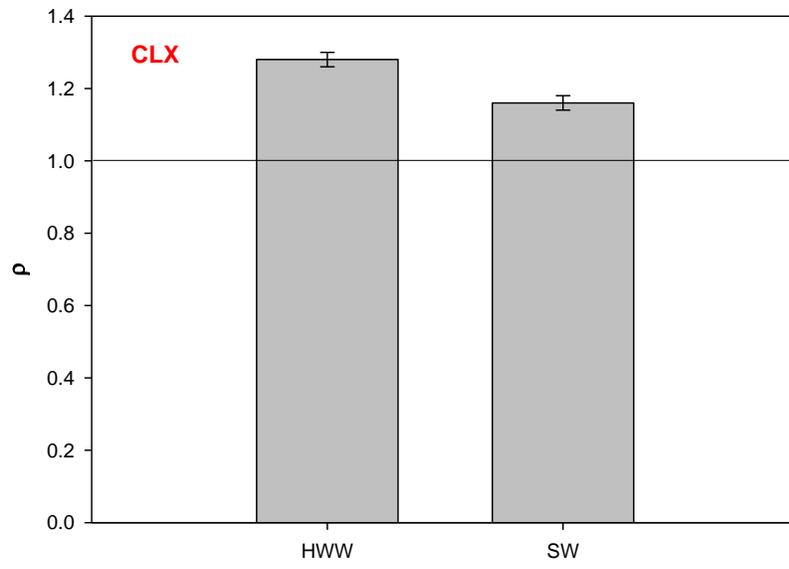


**Figure 2.** Correlation of physico-chemical properties and degradation rate of the considered antibiotics. **(A)** Rd vs. Solubility in water ( $S_w$ ). **(B)** Rd vs. Octanol-water partition coefficient ( $\text{Log } P$ ). **(C)** Rd vs. Topological polar surface area (TPSA). The values antibiotics properties were taken from [22].

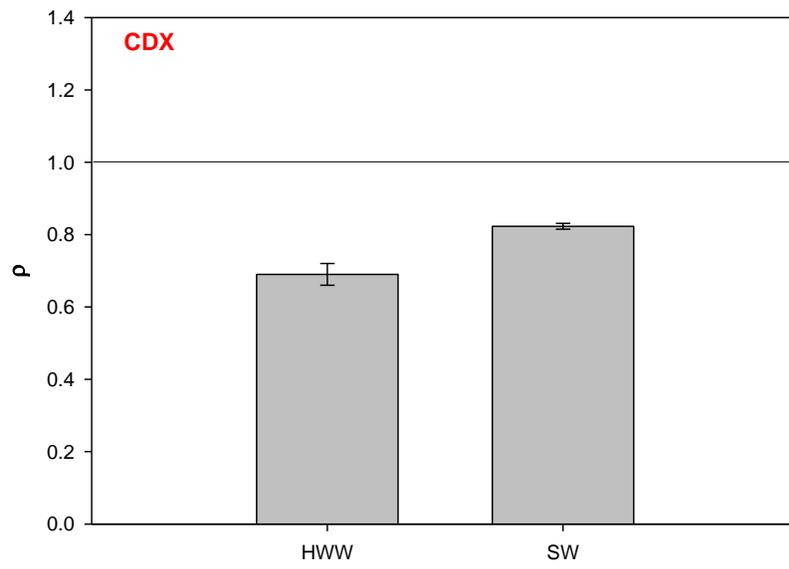


**Figure 3.** Antimicrobial activity (AA) evolution during the sonochemical treatment of CLX, NOR and CPX. Indicator microorganism: *S. aureus*. Ultrasonic conditions as indicated in Fig. 1.

A



B



**Figure 4.** Treatment of representative antibiotics by ultrasound in distilled water (DW), hospital wastewater (HWW) and seawater (SW). **(A)** CLX antibiotic. **(B)** CDX antibiotic. Ultrasonic conditions as indicated in **Fig. 1**.

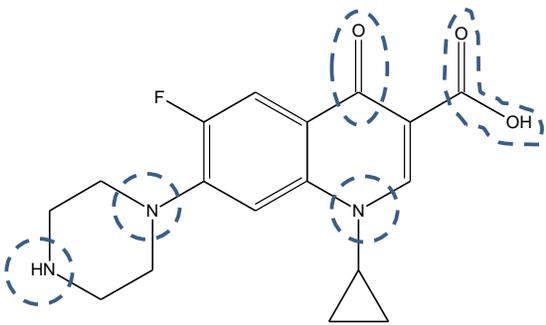
### Table captions

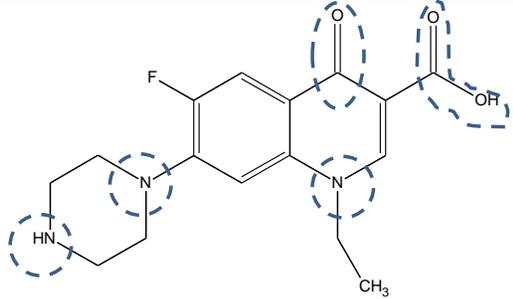
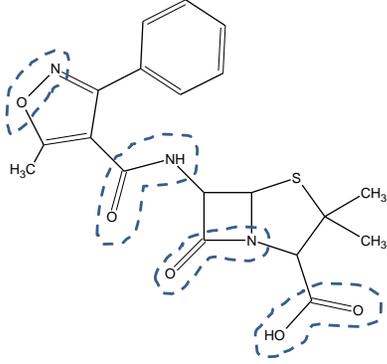
**Table 1.** Inhibition degree of sonochemical degradation (IDS) by addition of 2-propanol.

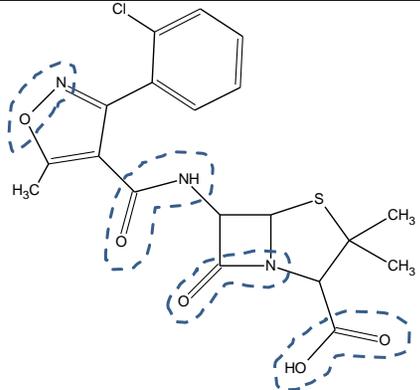
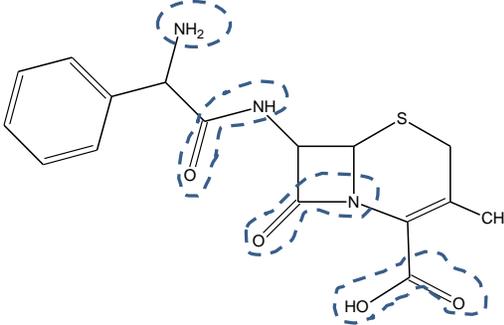
<b>Representative antibiotics</b>	<b>IDS (%)*</b>
Norfloxacin (NOR)	100
Cephalexin (CPX)	80
Cloxacillin (CLX)	55

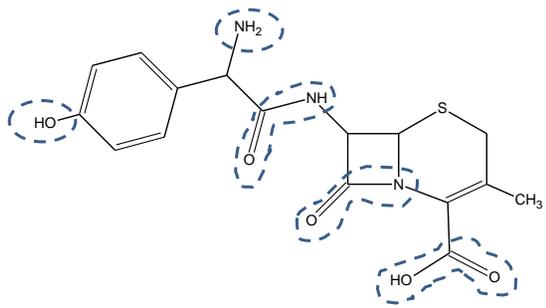
\*In presence of IPA (40 mM)

**Table 2.** Estimation of antibiotics hydrophobicity using the constitutional approach (Lemke method [26]).

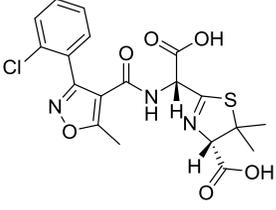
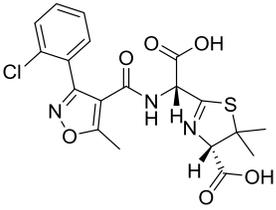
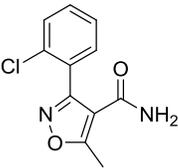
Structure (Hydrophilic moieties are signaled with dashed lines)	Hydrophilic Contribution			Hydrophobic contribution		
	Fragment	Amount	contribution	Fragment	amount	Contribution
<p style="text-align: center;">CIP</p> 	Amine	3	-3.00	Phenyl	1	+2.00
	Ketone	1	-0.55	Aliphatic carbon	9	+4.50
	Carboxylic Acid	1	-0.70	Fluor	1	+0.14
	<b>Addition of contributions</b>	-4.25		<b>Addition of contributions</b>	+6.64	
	<b>Total hydrophobicity = +6.64 -4.25 = 2.39</b>					
	<p style="text-align: center;">NOR</p>	Amine	3	-3.00	Phenyl	1
Ketone		1	-0.55	Aliphatic carbon	8	+4.00
Carboxylic Acid		1	-0.70	Fluor	1	+0.14

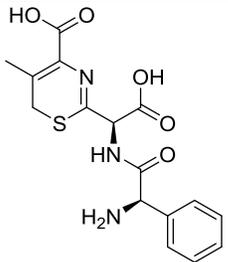
	<b>Addition of contributions</b>	-4.25	<b>Addition of contributions</b>	+6.14		
<p style="text-align: center;"><b>Total hydrophobicity = +6.14 -4.25 = 1.89</b></p>						
<p style="text-align: center;"><b>OXA</b></p> 	Amine	1	-1.00	Phenyl	1	+2.00
		1	-1.00	Aliphatic carbon	10	+5.00
		1	-0.70	Thioether	1	+0.00
		2	-1.40	---	---	---
<b>Addition of contributions</b>		-4.10		<b>Addition of contributions</b>		+7.00
<p style="text-align: center;"><b>Total hydrophobicity = +7.00 -4.10 = -2.90</b></p>						
<p style="text-align: center;"><b>CLX</b></p>	Amine	1	-1.00	Phenyl	1	+2.00
		1	-1.00	Aliphatic Carbon	10	+5.00

	Carboxylic Acid	1	-0.70	Thioether	1	+0.00
	Amide	2	-1.40	Chloro	1	+ 0.50
	<b>Addition of contributions</b>	-4.10		<b>Addition of contributions</b>	+7.50	
	<b>Total hydrophobicity = +7.50 -4.10 = 3.40</b>					
<p style="text-align: center;"><b>CPX</b></p> 	Amine	1	-1.00	Phenyl	1	+2.00
	Carboxylic Acid	1	-0.70	Aliphatic carbon	7	+3.50
	Amide	2	-1.40	Thioether	1	+0.00
	---	---	---	---	---	---
	<b>Addition of contributions</b>	-3.10		<b>Addition of contributions</b>	+5.50	
	<b>Total hydrophobicity = +5.50 -3.10 = 2.40</b>					
<p style="text-align: center;"><b>CDX</b></p>	Amine	1	-1.00	Phenyl	1	+2.00

	Carboxylic Acid	1	-0.70	Aliphatic carbon	7	+3.50
	Amide	2	-1.40	Thioether	1	+0.00
	Hydroxyl	1	-1.00	---	---	---
	<b>Addition of contributions</b>	-4.10		<b>Addition of contributions</b>	+5.50	
	<b>Total hydrophobicity = +5.50 - 4.10 = 1.40</b>					

**Table 3.** Identified primary degradation products from sonochemical treatment of representative antibiotics.

Parent antibiotic	Transformation product	Chemical name	Calculated $\Delta\text{Log } P^*$
CLX	<p>DP1</p> 	(S)-2-((R)-carboxy(3-(2-chlorophenyl)-5-methylisoxazole-4-carboxamido)methyl)-5,5-dimethyl-4,5-dihydrothiazole-4-carboxylic acid ( <i>E</i> )	-3.92
	<p>DP2</p> 	(S)-2-((R)-carboxy(3-(2-chlorophenyl)-5-methylisoxazole-4-carboxamido)methyl)-5,5-dimethyl-4,5-dihydrothiazole-4-carboxylic acid ( <i>Z</i> )	-3.92
	<p>DP3</p> 	3-(2-chlorophenyl)-5-methylisoxazole-4-carboxamide	-0.97
NOR	<p>DP4</p> 	1-ethyl-6-fluoro-7-(piperazin-1-yl)quinolin-4(1 <i>H</i> )-one	+0.18
CPX	<p>DP5</p>	2-	-1.47

	 <p>The image shows the chemical structure of a thiazine derivative. It features a six-membered 1,3-thiazine ring with a methyl group at the 5-position and a carboxylic acid group at the 4-position. Attached to the 6-position of the ring is a side chain consisting of a chiral center bonded to a hydroxyl group (OH), an amide group (NH-C(=O)-), and a phenyl ring. The amide nitrogen is further substituted with a primary amino group (H<sub>2</sub>N).</p>	<p>[[[amino(phenyl)acetyl]amino] (carboxy)methyl]-5-methyl-6<i>H</i>- 1,3-thiazine-4-carboxylic acid</p>	
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\* $\Delta\text{Log P} = \text{Log P of transformation product} - \text{Log P of parent antibiotic}$ . The Log P values

were calculated using the molinspiration online software [31].