

## **SUPPORTING INFORMATION**

### **Efficient and Selective *N*-Methylation of Nitroarenes under Mild Reaction Conditions**

Elena Pedrajas, Iván Sorribes\*, Eva Guillamón, Kathrin Junge, Matthias Beller\*  
and Rosa Llusar\*

## 1. Catalyst characterization.

**Figure SI1.**  $^1\text{H}$  NMR spectrum of the complex  $[\text{Mo}_3\text{Pt}(\text{PPh}_3)_4\text{Cl}_3(\text{dmen})_3](\text{BF}_4)_3$  in  $\text{CD}_2\text{Cl}_2$

**Figure SI2.**  $^{13}\text{C}$  NMR spectrum of the complex  $[\text{Mo}_3\text{Pt}(\text{PPh}_3)_4\text{Cl}_3(\text{dmen})_3](\text{BF}_4)_3$  in  $\text{CD}_2\text{Cl}_2$ .

**Figure SI3.** ESI mass spectrum of the complex  $[\text{Mo}_3\text{Pt}(\text{PPh}_3)_4\text{Cl}_3(\text{dmen})_3](\text{BF}_4)_3$  in  $\text{CH}_3\text{CN}$  at 20 V.

**Figure SI4.** Cyclic Voltammogram recorded on a  $\text{CH}_2\text{Cl}_2$  solution containing  $3^+$  (a) and  $1^+$  (b) at scan rate of 100mV/s (*vs* Ag/AgCl).

## 2. Conditions optimization for the *N*-methylation of *p*-nitrotoluene (1a).

**Table SI1.** Screening of silanes.

**Table SI2.** Influence of the solvent on the catalytic *N*-methylation of 1a.

## 3. ESI mass spectra from the reaction mixture during the *N*-methylation of 1a.

**Figure SI5.** ESI mass spectrum from the *N*-methylation reaction after 8 hours.

**Figure SI6.** ESI mass spectrum from the mixture of  $[\text{Mo}_3\text{S}_4\text{Cl}_3(\text{dmen})_3](\text{BF}_4)_3$  (1) (0.003 mmol) and  $\text{Pt}(\text{PPh}_3)_4$  (2) (0.001 mmol) in THF after 10 minutes stirring at room temperature.

## 4. Reaction pathway investigation.

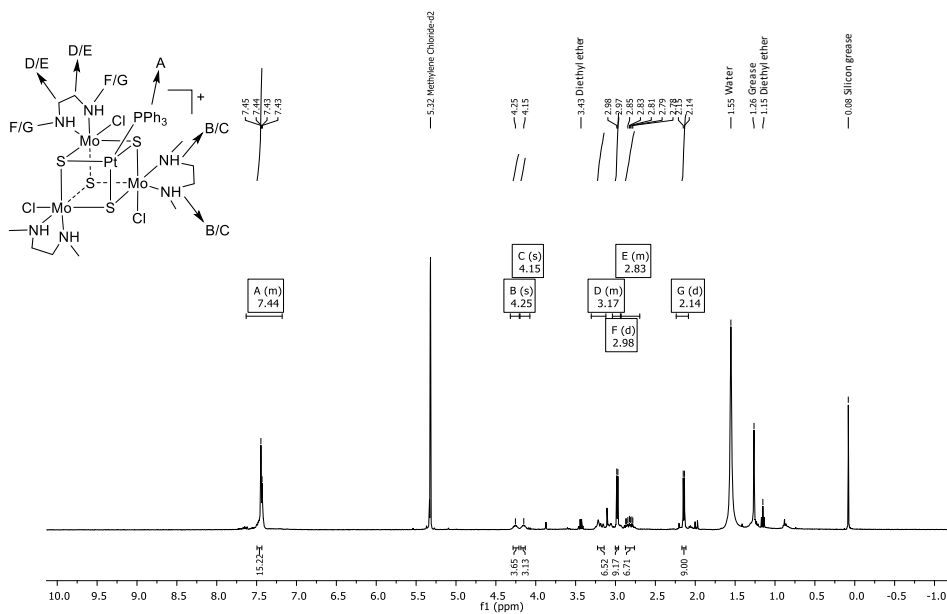
**Scheme SI1.** Proposed pathways for the direct *N*-methylation of nitroarenes with formic acid.

## 5. Characterization data of isolated products.

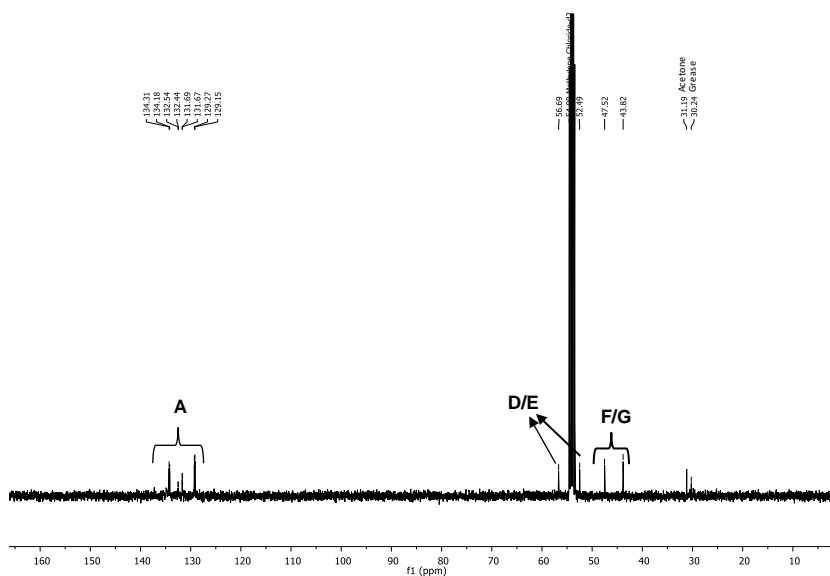
## 6. References

## 7. $^1\text{H}$ NMR and $^{13}\text{C}$ NMR spectra of isolated products.

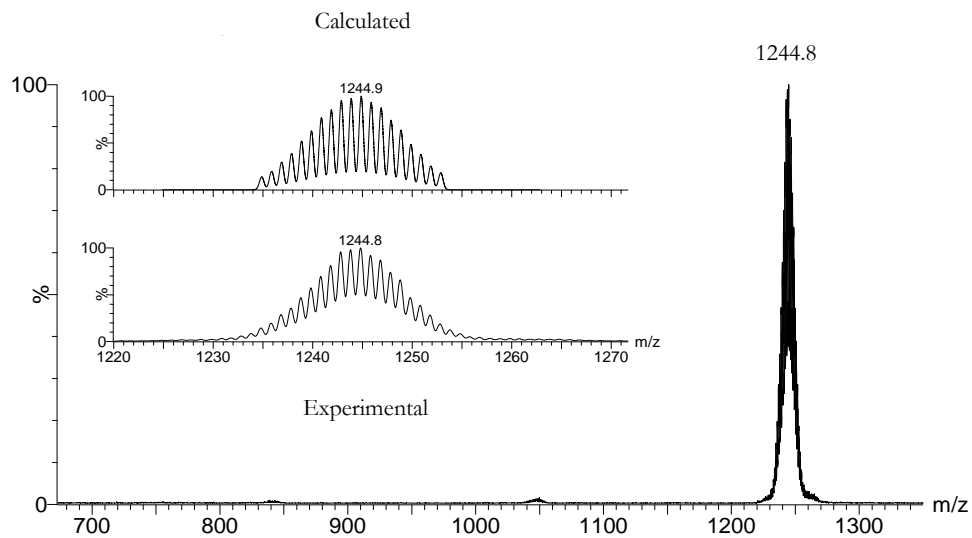
## 1. Catalyst characterization



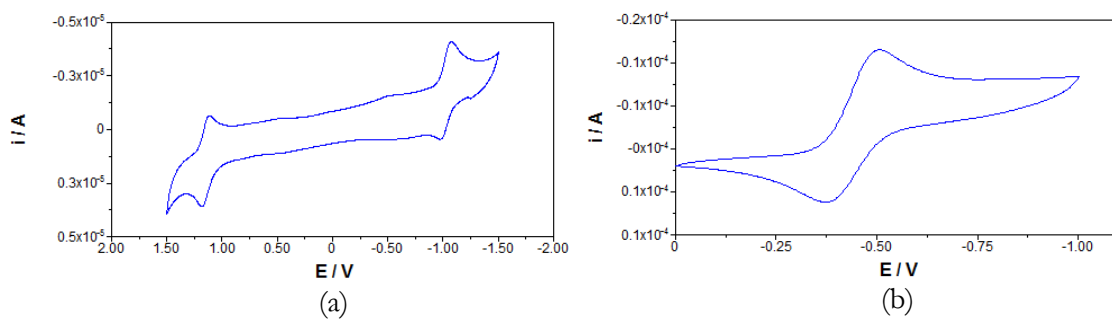
**Figure S11.**  $^1\text{H}$  NMR spectrum of the  $[\text{Mo}_3\text{Pt}(\text{PPh}_3)_4\text{S}_4\text{Cl}_3(\text{dmen})_3](\text{BF}_4)$  ( $\mathbf{3}(\text{BF}_4)$ ) complex in  $\text{CD}_2\text{Cl}_2$ .



**Figure S12.**  $^{13}\text{C}$  NMR spectrum of the  $[\text{Mo}_3\text{Pt}(\text{PPh}_3)_4\text{S}_4\text{Cl}_3(\text{dmen})_3](\text{BF}_4)$  ( $\mathbf{3}(\text{BF}_4)$ ) complex in  $\text{CD}_2\text{Cl}_2$ .



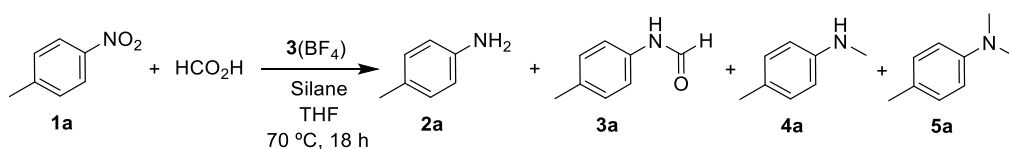
**Figure SI3.** ESI mass spectrum of the  $[\text{Mo}_3\text{Pt}(\text{PPh}_3)_4\text{Cl}_3(\text{dmen})_3](\text{BF}_4)$  complex ( $\mathbf{3}(\text{BF}_4)$ ) in  $\text{CH}_3\text{CN}$  at 20 V.



**Figure SI4.** Cyclic Voltammogram recorded on a  $\text{CH}_2\text{Cl}_2$  solution containing  $\mathbf{3}^+$  (a) and  $\mathbf{1}^+$  (b) at scan rate of 100 mV/s (*vs* Ag/AgCl).

## 2. Conditions optimization for the *N*-methylation of *p*-nitrotoluene (**1a**).

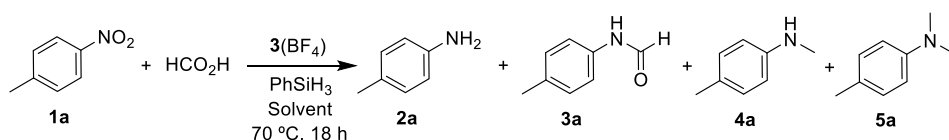
**Table SI1.** Screening of silanes.<sup>[a]</sup>



Entry	Silane	Conversion [%] <sup>[b]</sup>	Yield <b>2a</b> [%] <sup>[b]</sup>	Yield <b>3a</b> [%] <sup>[b]</sup>	Yield <b>4a</b> [%] <sup>[b]</sup>	Yield <b>5a</b> [%] <sup>[b]</sup>
1	PhSiH <sub>3</sub>	>99	1	0	2	97
2	Ph <sub>2</sub> SiH <sub>2</sub>	34	4	6	2	16
3	PhMe <sub>2</sub> SiH	0	0	0	0	0
4	Et <sub>3</sub> SiH	0	0	0	0	0
5	PHMS	21	3	0	1	6

[a] Reaction conditions: **1a** (0.1 mmol), HCO<sub>2</sub>H (8.5 equiv.), Silane (10 equiv.), Catalyst (3 mol%), THF (2 mL), 18 h, 70°C. [b] Determined by GC analysis using *n*-hexadecane as an internal standard.

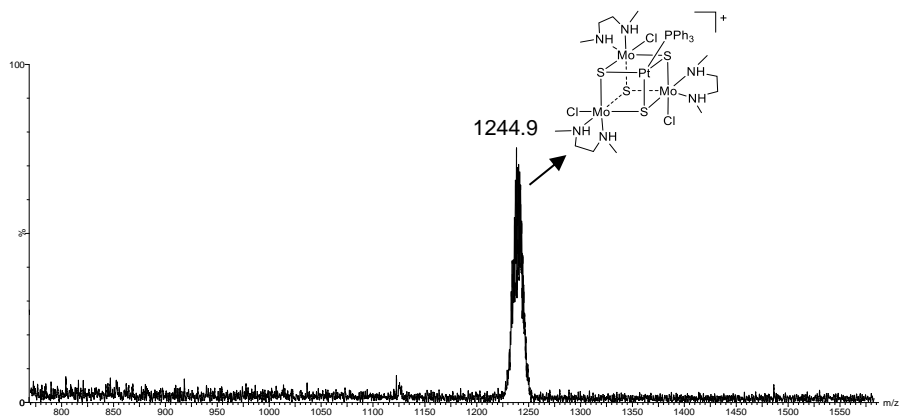
**Table SI2.** Influence of the solvent on the catalytic *N*-methylation of **1a**.<sup>[a]</sup>



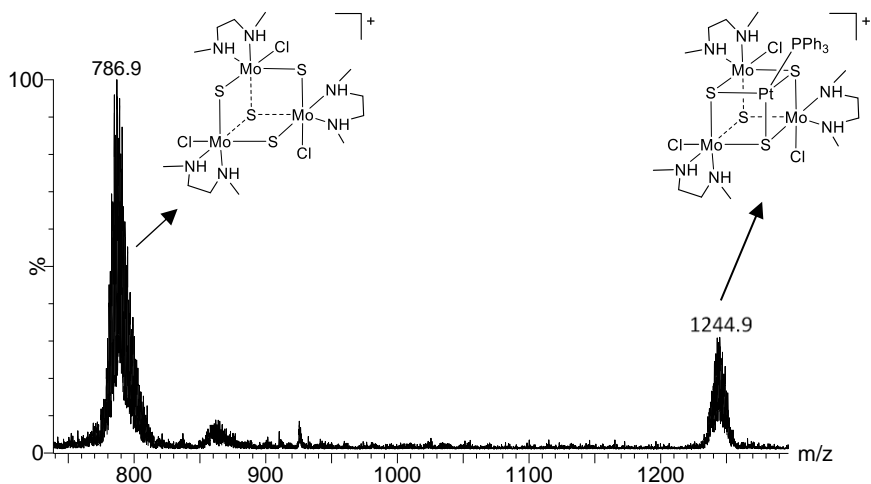
Entry	Solvent	Conversion	Yield <b>2a</b>	Yield <b>3a</b>	Yield <b>4a</b>	Yield <b>5a</b>
		[%] <sup>[b]</sup>	[%] <sup>[b]</sup>	[%] <sup>[b]</sup>	[%] <sup>[b]</sup>	[%] <sup>[b]</sup>
1 <sup>[c]</sup>	MeCN	50	10	1	5	4
2	MeOH	6	0	0	0	4
3 <sup>[c]</sup>	Toluene	99	41	5	13	13
4	THF	>99	1	0	2	97

[a] Reaction conditions: **1a** (0.1 mmol),  $\text{HCO}_2\text{H}$  (8.5 equiv.),  $\text{PhSiH}_3$  (10 equiv.), Catalyst (3 mol%), Solvent (2 mL), 18 h,  $70^\circ\text{C}$ . [b] Determined by GC analysis using hexadecane as an internal standard. [c] The urea intermediate 1,3-dimethyl-1,3-di-*p*-tolylurea is detected by GC-Mass.

### 3. ESI mass spectra from the reaction mixture during the *N*-methylation of **1a**.

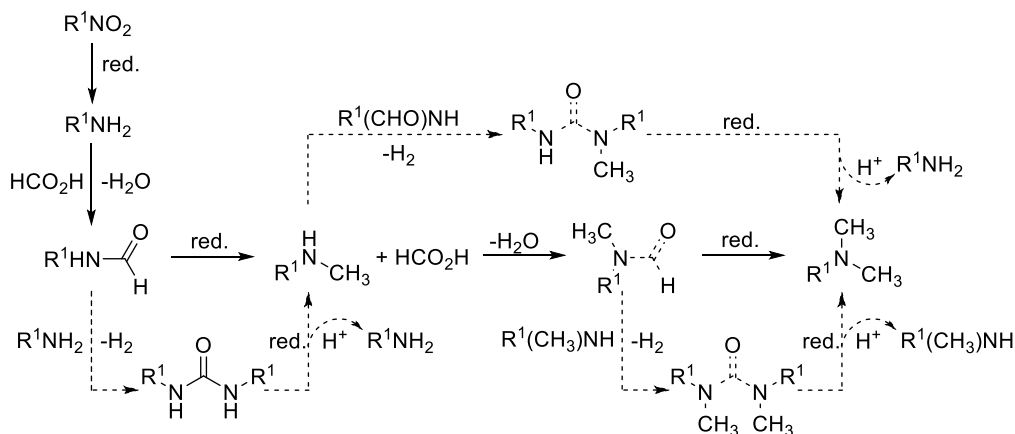


**Figure SI5.** ESI mass spectrum from the *N*-methylation reaction after 8 hours.



**Figure SI6.** ESI mass spectrum from the mixture of  $[\text{Mo}_3\text{S}_4\text{Cl}_3(\text{dmen})_3](\text{BF}_4)$  (**1**(BF<sub>4</sub>)) (0.003 mmol) and  $\text{Pt}(\text{PPh}_3)_4$  (**2**) (0.001 mmol) in THF after 10 minutes stirring at room temperature.

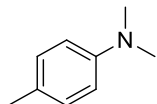
#### 4. Reaction pathway investigation.



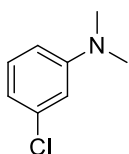
**Scheme SII.** Proposed pathways for the direct *N*-methylation of nitroarenes with formic acid in the presence of the heterobimetallic  $3^+$  catalyst.

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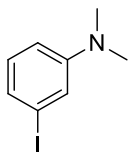
## 5. Characterization data of isolated products.



**N,N,4-trimethylaniline:**<sup>[1]</sup> <sup>1</sup>H NMR (300 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ 7.03 (d, *J* = 8.6 Hz, 2H), 6.66 (d, *J* = 8.6 Hz, 2H), 2.88 (s, 6H), 2.23 (s, 3H); <sup>13</sup>C NMR (75 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ 149.53, 130.00, 126.39, 113.56, 41.43, 20.48; MS (EI): *m/z* (rel. Int) 135.

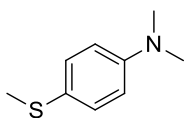


**3-Chloro-N,N-dimethylaniline:**<sup>[1]</sup> <sup>1</sup>H NMR (300 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ 7.13 (t, *J* = 8.1 Hz, 1H), 6.70 – 6.56 (m, 3H), 2.94 (s, 6H); <sup>13</sup>C NMR (75 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ 152.26, 135.31, 130.50, 116.33, 112.51, 111.04, 40.69; MS (EI): *m/z* (rel. Int) 155.

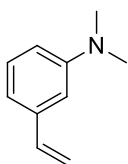


**3-iodo-N,N-dimethylaniline:**<sup>[1]</sup> <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 7.06 – 7.02 (m, 2H), 6.96 – 6.91 (m, 1H), 6.69 – 6.65 (m, 1H), 2.93 (s, 6H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ 151.68, 130.51, 125.33, 121.21, 111.70, 95.67, 40.45; MS (EI): *m/z* (rel. Int) 247.

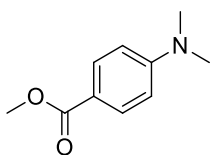




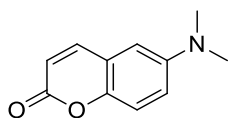
**N,N-dimethyl-4-(methylthio)aniline:**  $^1\text{H}$  NMR (300 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  7.14 (d,  $J$  = 9.0 Hz, 2H), 6.58 (d,  $J$  = 8.9 Hz, 2H), 2.81 (s, 6H), 2.29 (s, 3H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  134.63, 131.47, 128.45, 113.83, 41.02, 19.24; MS (EI):  $m/z$  (rel. Int) 167.



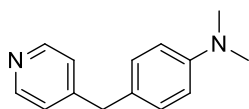
**N,N-dimethyl-3-vinylaniline:**  $^1\text{H}$  NMR (300 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  7.23 (t,  $J$  = 7.6 Hz, 1H), 6.88 – 6.77 (m, 2H), 6.77 – 6.65 (m, 2H), 5.78 (d,  $J$  = 17.6 Hz, 1H), 5.25 (d,  $J$  = 10.8 Hz, 1H), 3.00 (s, 6H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  151.61, 138.84, 138.39, 129.65, 115.06, 113.46, 112.88, 111.08, 40.94; MS (EI):  $m/z$  (rel. Int) 147.



**Methyl 4-(dimethylamino)benzoate:**  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  7.76 (d,  $J$  = 9.1 Hz, 2H), 6.56 (d,  $J$  = 9.1 Hz, 2H), 3.72 (s, 3H), 2.92 (s, 6H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  167.72, 153.99, 131.58, 117.44, 111.22, 51.79, 40.39; MS (EI):  $m/z$  (rel. Int) 179.



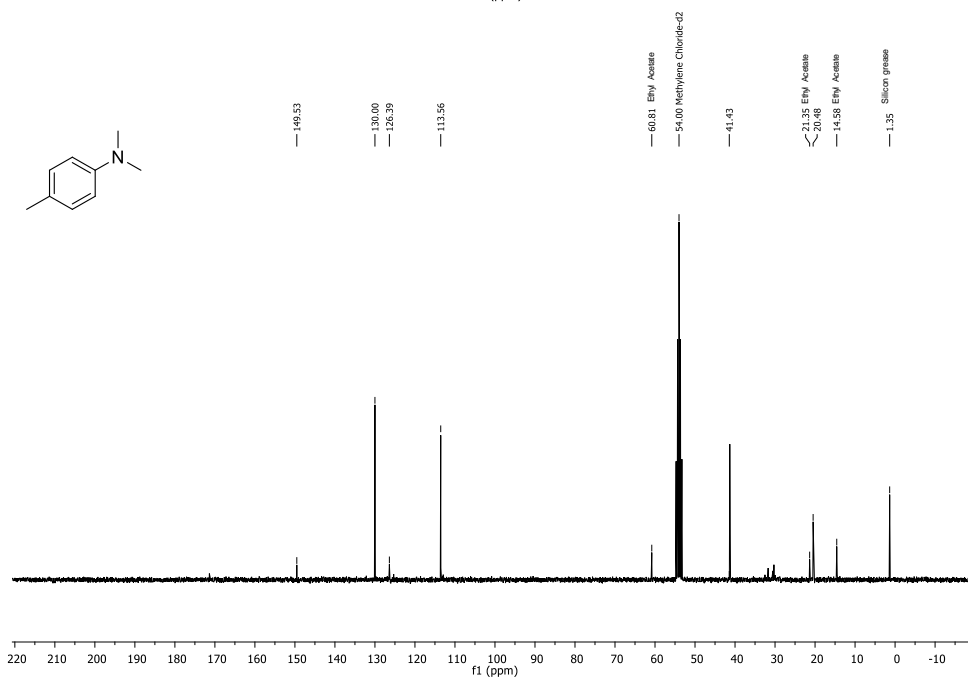
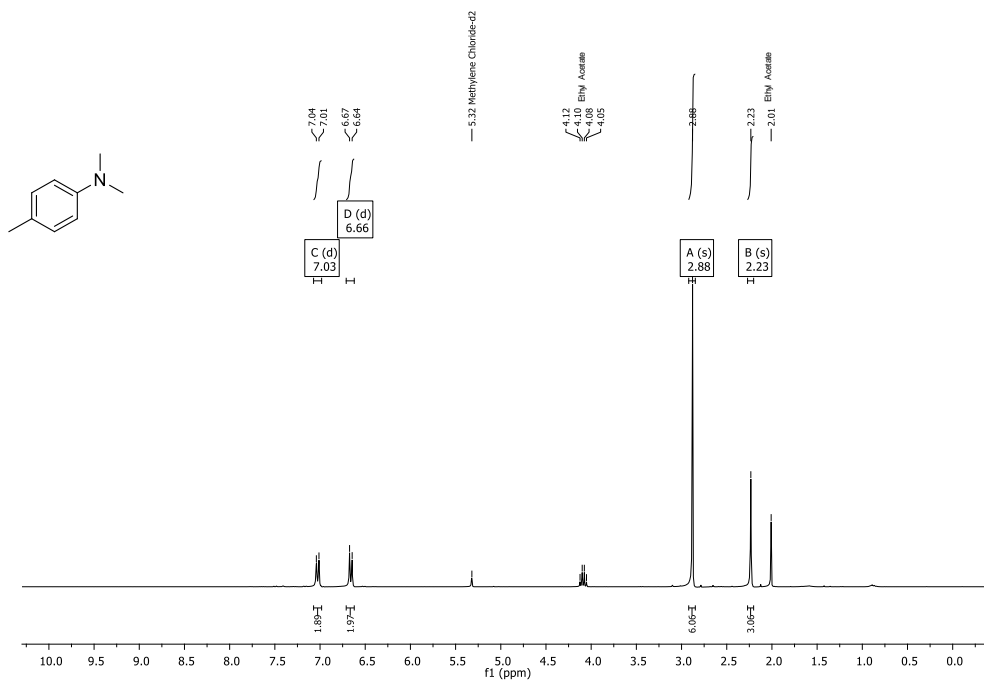
**6-(dimethylamino)-2H-chromen-2-one:**  $^1\text{H}$  NMR (300 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  7.65 (d,  $J = 9.5$  Hz, 1H), 7.20 (d,  $J = 9.1$  Hz, 1H), 6.97 (dd,  $J = 9.1, 3.0$  Hz, 1H), 6.70 (d,  $J = 3.0$  Hz, 1H), 6.32 (d,  $J = 9.5$  Hz, 1H), 2.96 (s, 6H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  161.52, 148.05, 146.58, 144.03, 119.58, 117.42, 117.22, 116.97, 109.50, 41.01; MS (EI):  $m/z$  (rel. Int) 189.

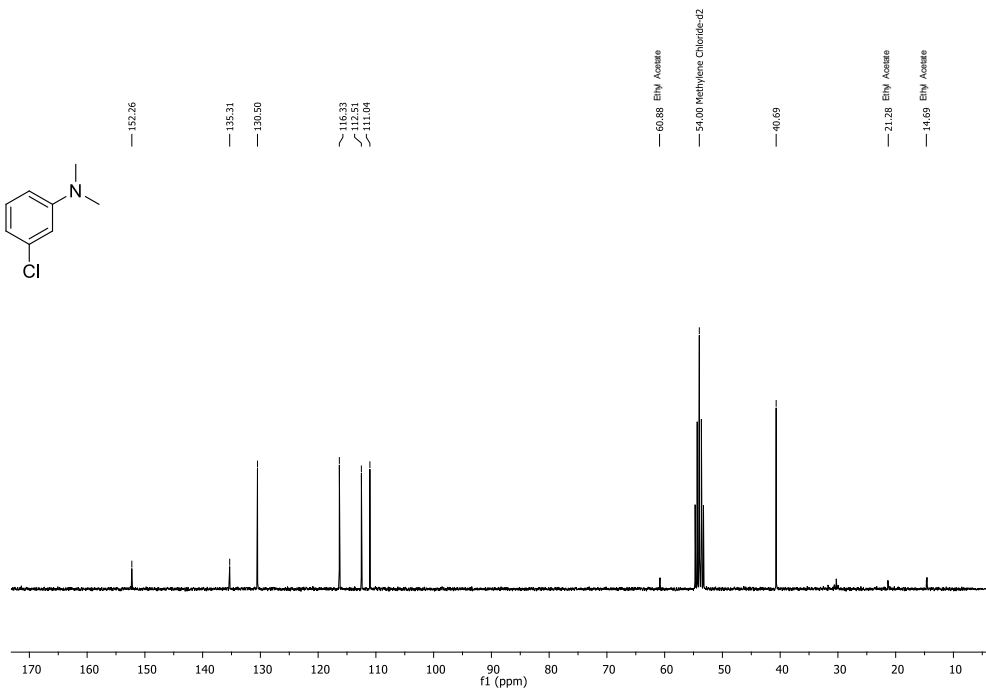
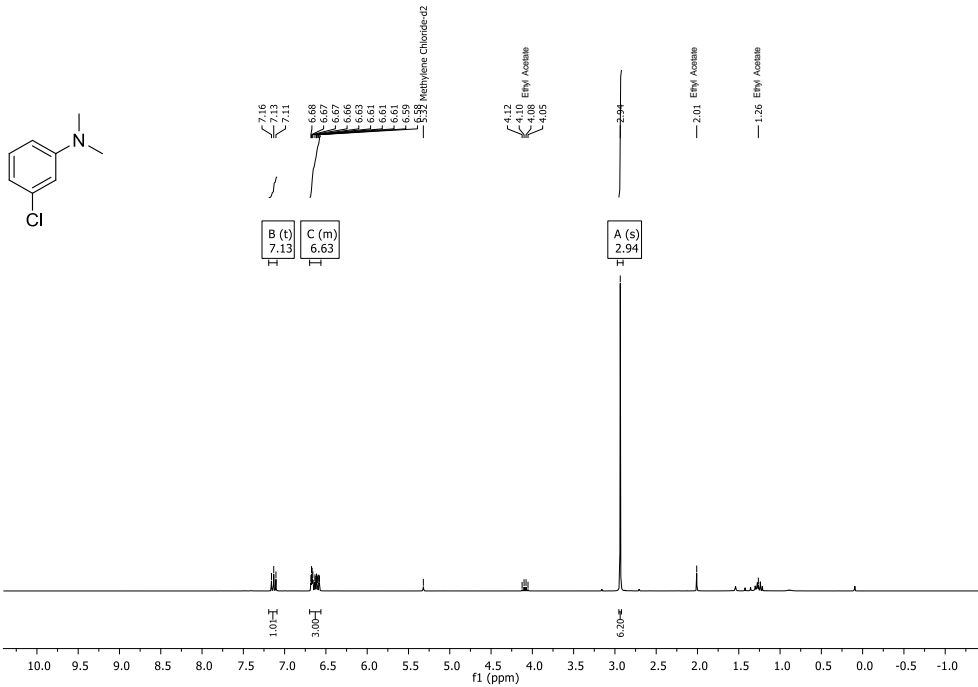


**N,N-dimethyl-4-(pyridin-4-ylmethyl)aniline:**  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  8.41 – 8.36 (m, 2H), 7.08 – 7.02 (m, 4H), 6.73 – 6.66 (m, 2H), 3.85 (s, 2H), 2.92 (s, 6H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$  151.97, 149.95, 130.13, 128.27, 127.42, 124.59, 113.39, 41.02, 40.77; MS (EI):  $m/z$  (rel. Int) 212.

## 6. References.

- [1] X. Jiang, C. Wang, Y. Wei, D. Xue, Z. Liu and J. Xiao, *Chem. Eur. J.* **2014**, *20*, 58–63.

7. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra of isolated products.



Efficient and selective *N*-methylation of nitroarenes under mild reaction conditions

